INNOVATING SCIENCE®

"Cutting edge science for the classroom"

HANDS-ON SCIENCE FOR:

Elementary - Middle School - High School - College

Forensic Science

Safety Kits

AP Biology

Environmental Science

AP Chemistry

Elementary Science

Green Chemistry Life Science/Biology

Distance Learning

Biomedical Engineering

STEM Kits

Earth Science

Chemistry Demonstrations

Small Group Learning





Aldon 221 Rochester Street Avon, NY 14414 ph: (585) 226-6177 fax: (585) 226-6919 www.aldon-chem.com

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New Products for 2025

Distance Learning: Science in the Kitchen

This comprehensive kit incorporates a variety of scientific techniques all themed around common kitchen materials. Students will perform chromatography on commercial food colors, use chemical tests to identify an unknown cooking ingredient, examine the protein digesting ability of a common meat tenderizing enzyme, use titration to quantify vitamin C levels and then test an unknown juice or soda (not provided), and lastly examine some of the differences and similarities in the materials used to clean up (soap, hand dishwashing detergent, and machine dishwashing detergent). Kit contains enough materials for 5 groups. Teacher's Manual and Student Study Guide copymasters are included.

Kit Contains:

25g	Unknown Powder	
5mL	Blue Food Dye	
5mL	Green Food Dye	
5mL	Red Food Dye	
5mL	Yellow Food Dye	
5g	Gelatin Powder	
5g	Baking Powder	
5g	Papain Powder	4
5g	Baking Soda	5
5g	Corn Starch	6
5mL	Vinegar	5r
5mL	Dishwashing Detergent - Machine	5r
30mL	Iodine Potassium Iodide	30
10mL	5% Soap	5r

Ascorbic Acid Capsule Chromatography Paper Centrifuge Tube Spot Plate Toothpicks, Package Pipette Universal Indicator pH Strips Capillary Tubes Test Tubes, Plastic Medicine Cups 5mL 5% Calcium Chloride **Dishwashing Detergent - Hand** 5ml **Dilute Lugols Solution** 30mL Starch Indicator Solution 5mL



DOT Information: Not Regulated

IS1519

\$66.00

Distance Learning: Plant Tissue Micronutrients

Plants need a variety of nutrients and minerals to grow and maintain overall health. The three main nutrients that plants require to survive and thrive are nitrogen, phosphorus, and potassium (also known as NPK). In this activity, students will extract nutrients from plants of your choice and perform analyses to determine if nitrogen, phosphorus, and potassium concentration is abundant, adequate, or deficient. Kit contains enough materials to perform 5 tests.

Kit Contains:

50mL	Plant Nutrient Extractor
5mL	Nitrate Reagent #1
5g	Nitrate Reagent #2
5g	Phosphorous Reagent #1
5mL	Phosphorous Reagent #2
5g	Potassium Reagent #1
10mL	Potassium Reagent #2

- Filter Paper, Pack/100 4 Collection Bottles
- Spot Plate
- Stir Stick
- 5 Pipettes

DOT Info:

Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173.4 for domestic highway or rail transport only

IS1520

\$33.66



Distance Learning: Introduction to Mendelian Genetics

In this activity, students can simulate Mendel's work and determine patterns of inheritance. Using special chips and Innovating Science's exclusive "double dice," students will be able to simulate both monohybrid and dihybrid crosses. After the crosses, students will be able determine genotypic and phenotypic ratios for select traits and compare their values to the theoretical "ideal" values as put forth by Mendel. Kit contains enough materials for 2 groups. Teacher's Manual and Student Study Guide copymasters are included.

Kit Contains:

Monohybrid chip, female (red/yellow) Monohybrid chip, male (red/white) Dihybrid double die, female (clear) Dihybrid double die, male (colored) Plastic shaker cup

DOT Information: Not Regulated

IS1521

\$7.75



Aldon

New Products for 2025

Distance Learning: Ocean Acidification

Changes to the carbon cycle and climate due to human activities have had detrimental effects on the environment. One example of this is known as ocean acidification, which is the changing of the ocean's chemistry due to increased carbon dioxide in the atmosphere and ocean. In this activity, students will learn about the carbon cycle and how it relates to ocean acidification. Through three different activities, students will investigate the cause of ocean acidification, and the effects it can have on the ecosystem and marine life. Kit contains enough materials for 2 groups of students. Teacher's Manual and Student Study Guide Copymasters are included.

Kit Contains:10mLBromothymol Blue2x30mL0.1 N Hydrochloric Acid2x30mL1.0 N Hydrochloride Acid5gSodium Bicarbonate

2 Cups, Bomb shot 4 Antacid Effervescent Tablets 6 Seashells

DOT Info: Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173.4 for domestic highway or rail transport only

IS1522

\$12.99



Introduction to Complexometric Titration

Complex formation is frequently used in both qualitative and quantitative inorganic analysis. This activity introduces topics such as metal-ligand interactions using complexometric titration to determine the calcium content of both known and unknown samples. Kit contains enough materials for 15 groups. Teacher's Manual and Student Study Guide copymasters are included.

Kit Includes:

4x500 mL	EDTA 0.1 WI Htrant Solution
3x60 mL	Eriochrome Black T Indicator Solution
500 mL	Calcium Chloride 0.1 M Solution
500 mL	Unknown Calcium Solution #1
500 mL	Unknown Calcium Solution #2
3x500 mL	pH 10 Buffer

EDTA 0.4 M Titrant Colution

DOT Info: UN1170, Ethanol, 3, II, LTD QTY

WARNING! : This product can expose you to chemicals including Methanol and Methyl isobutyl ketone, which are known to the State of California to cause cancer and birth defects or other reproductive harm.. For more information go to http://www.P65Warnings.ca.gov.

IS2534



Arabidopsis Seeds

Innovating Science Fast Growing Plant Seeds Great for science fair projects and University Research. Quick growth time. 94% Viability. Pk/300. For Lab and Research Use Only.

DOT Info: Non-regulated

IS3731 \$30.78





Aldon

New Products for 2025

Synthesis of a Coordination Compound

Coordination complexes are brightly colored compounds consisting of a central metal atom and ligand molecules, atoms, or ions. Synthesize and purify a lime-green coordination complex called potassium ferrioxalate trihydrate in this multi-step synthesis and optionally analyze the complex via UV-Visible spectrophotometry. This kit contains enough materials for 15 groups, a Teacher's Manual, and Student Study Guide copymaster.

Kit Contains:

615mL0.5 M Oxalic Acid180mL1.0 M Potassium Oxalate180mL3% Hydrogen Peroxide135mL95% Ethanol180mL50% Ethanol325mL0.1 M Sulfuric Acid50gFerrous Ammonium Sulfate Hexahydrate15Pipettes15Filter Papers

DOT Info: UN1760, Corrosive liquids, n.o.s., (Oxalic acid), 8, III, Ltd Qty UN1170, Ethanol, 3, II, Ltd Qty

IS2535

\$49.95

50g

15

15



NYS Investigations for Life Science Lactose Intolerance: Inheritance and Variance of Traits

Milk Powder

Stirring Rods

Lactase Scoop

Glucose Color Charts

Lactose intolerance is common, but is actually a recessive genetic trait! Humans and other mammals are only meant to drink milk as infants and as they grow, they lose the ability to digest lactose. As populations began to develop, the domestication of cows and drinking milk into adulthood became more common, leading to a tolerance for lactose. Learn about variation, inheritance, and the role of DNA and proteins in expression of traits in this three-part laboratory activity! This kit contains enough materials for 15 groups, a Teacher's Manual, and a Student Study Guide copymaster.

Kit Contains:

3gLactase Powder15Glucose Test Strips30Mixing Cups30Multicolored ChipsMeasuring Cup

DOT Info: Not Regulated

IS3017 \$53.46



Unknown Compositions Samples: Potassium Hydrogen Phthalate

A series of 10 potassium hydrogen phthalate samples with varying purities ranging from 20% to 65%. Use these in practicing analytical techniques for determining substance purity.

DOT Info: Non-regulated

IS5900

\$76.63







New Products for 2025

Unknown Compositions Samples: Ferrous Ammonium Sulfate

A series of 20 ferrous ammonium sulfate samples with varying iron compositions ranging from 0.1% to 13%. Use these in practicing analytical techniques for determining substance purity.

DOT Info:Non-regulated

IS5901

\$127.10



Unknown Compositions Samples: Soda Ash

A series of 10 soda ash (sodium carbonate) samples with varying purities ranging from 20% to 50%. Use these in practicing analytical techniques for determining substance purity.

DOT Info: Non-regulated

IS5902 \$41.12



Photochemistry & Cyanotype Printing Demonstration

Explore the chemistry behind a classic photograph development method: Cyanotype Printing. This process involves photo-redox of iron compounds and complexation to form Prussian Blue pigment.

Kit Contents:

- 100mL Ferric Nitrate Solution 100mL Oxalic Acid Solution
- 200mL Potassium Ferricyanide Solution 200mL Intensifying Solution
- 4 Plastic Trays
- 4 Circular Paper Sheets
- 4 Medicine Cups
- Plastic Forceps

DOT Info: UN1760, Corrosive liquids, n.o.s., (Oxalic acid), 8, III, Ltd Qty

IS7050 \$30.84



Firefly in a Flask: Oscillating Chemiluminescence

Two glow-in-the-dark reactions occur in turn: one that is bright and quick, and another that is dull and slow. The oscillation between these reactions looks like a firefly on a summer night!

Kit Contains:3x900mLLuminol-Thiocyanate Solution3x100mLCopper Sulfate Solution3x30mLHydrogen Peroxide 30%

DOT Info: Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173.4 for domestic highway or rail transport only

IS7052

\$45.45



Absorption Spectroscopy Demonstration

Explore the properties of atomic absorption and discover how certain elements absorb several specific wavelengths of light by viewing the dark line (absorption) spectra of various colored solutions including solutions containing elements that astronomers observe: erbium and praseodymium.

Kit Contains:		
20mL	Red Food Dye	
20mL	Yellow Food Dye	
20mL	0.001 M Potassium Permanganate	
20mL	1 M Copper (II) Chloride	
20mL	0.1 M Erbium Chloride	
20mL	0.1 M Praseodymium Chloride	
6	Petri Dishes	
6	Pipettes	
3	Black Construction Paper Pieces	
2	Diffraction Grating Slides	
DOT Infe		
Small quantity exemption 173.4		

THIS PACKAGE CONFORMS TO 49 CFR 173.4 for domestic highway or rail transport only

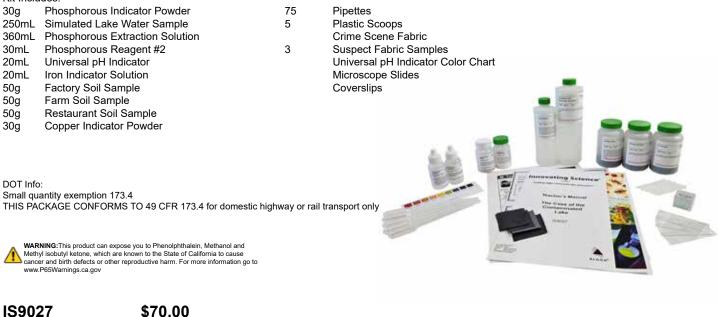


IS7051 \$89.00

The Case of the Contaminated Lake

You are a scientist from the Environmental Protection Agency (EPA) visiting Lakeville for the annual water quality inspection of Pine Lake, however something has gone awry. Someone intentionally polluted the lake! Perform chemical tests for contaminants on water and soil samples, analyze fibers, fingerprints, and footprints, and investigate suspect activity on social media in this environmental-forensic chemistry activity to uncover the culprit behind the crime. This kit contains enough materials for 15 groups of students. Teacher's Manual and Student Study Guide copymaster included.

Kit Includes:



New Products

Innovating Science®

Forensic Urine Drug Test Kit

Your class will utilize different lab techniques to determine the presence of alcohol, illicit drugs and pain medications present in the simulated urine samples of suspected criminals. The various tests include the Marguis Test for Morphine, Simon's test for MDMA, THC Immunoassay, Cocaine Immunoassay, and 6-Monoactylmorphine Test. This kit includes enough materials for 30 students. Teacher Manual and Student Study guide copy masters are included.

Kit Includes:

100mL	Driver urine sample
100mL	Passenger #2 urine sample
100mL	Patient #1 urine sample
100mL	Patient #3 urine sample
100mL	Patient #5 urine sample
400mL	Patient #7 urine sample
400mL	Patient #9 urine sample
400mL	Patient #11 urine sample
200mL	Patient #13 urine sample
200mL	Patient #15 urine sample 1
30mL	Synthetic ethanol test reagent
30mL	Synthetic Simon's reagent
500mL	Synthetic THC indicator Solution 1
30mL	Synthetic 6-monoacetylmorphine

100mL Passenger #1 urine sample 100mL Passenger #3 urine sample 100mL Patient #2 urine sample Patient #4 urine sample 100mL 400mL Patient #6 urine sample 400mL Patient #8 urine sample 400mL Patient #10 urine sample 200mL Patient #12 urine sample Patient #14 urine sample 200mL 20mL Synthetic cocaine indicator 60mL Synthetic marquis reagent Ferric chloride 1.0M 2mL 60mL Synthetic THC indicator Solution 2 Synthetic marguis reagent color charts indicator

WARNING: This product can expose you to chemicals including Phenolphthalein/Methanol/ Methyl isobutyl ketone, which are known to the State of California to cause cancer, reproductive harm. For more information go to www.P65Warnings.ca.gov.

DOT Info: UN1789. hvdrochloric acid. 8. III. LTD QTY



IS9050

Ocean Pollution

Examine five different types of ocean pollution in simulated ocean water. Remove debris, particulates, and oil, and test the water for toxins and dissolved nutrients. This kit contains enough materials for 15 groups working in pairs. A Teacher's Manual and Student Study Guide copymaster are included.

Kit Includes:

Simulated Polluted Ocean Water Solids 2gal 150mL Simulated Crude Oil 15mL Nitrogen Test Solution #1 15mL Nitrogen Test Solution #2 15mL Nitrogen Test Solution #3 7 Simulated Copper Test Strips 45g Sand Cork Dust 1g 15" Cotton String 15 Small Balloons Plastic Straws 3

15 Forceps

15

- 30 Pipettes
- Cups, 5 oz 15 31
- Medicine Cups 15 Plastic Containers Plastic Scoop Cotton Balls
- 15 Spoons
- 15 Coffee Filters Pipette, 3 mL



DOT Info:

Small quantity exemption 173.4

THIS PACKAGE CONFORMS TO 49 CFR 173.4 for domestic highway or rail transport only

\$148.00

IS9708 \$98.00 WARNING: This product can expose you to a chemical. crystalline slica, which is known to the State of California to cause cancer. For more information go to www.P65Warnings.ca.gov.

Ocean Acidification

Changes to the carbon cycle and climate due to human activities have had detrimental effects on the environment. One example of this is known as ocean acidification, which is the changing of the ocean's chemistry due to increased carbon dioxide in the atmosphere and ocean. In this activity, students will learn about the carbon cycle and how it relates to ocean acidification. Through three different activities, students will investigate the cause of ocean acidification, and the effects it can have on the ecosystem and marine life. Kit contains enough materials for 15 groups of students.Teacher's Manual and Student Study Guide Copymasters are included.

Kit Includes:

25mLBromothymol blue500mLHydrochloric acid, 0.1M500mLHydrochloric acid, 1.0M10gSodium bicarbonate

- Effervescent tablets
- Plastic cups
 Seashell pieces

16

DOT Info: UN1789, hydrochloric acid, 8, III, LTD QTY

IS9756 \$37.00



Determination of Dissolved Carbon Dioxide Concentration

The proper amount of dissolved carbon dioxide in water is critical for maintaining a healthy aquatic ecosystem. All aquatic life has an impact on and is in turn impacted by the dissolved carbon dioxide concentration in a system, so tracking the concentration and changes in concentration can help scientists to better understand the health of an aquatic system and the organisms that exist within it. Kit contains enough materials to perform 40 tests.

Kit Includes:

25mL Phenolphthalein Indicator

- 25mL Sodium Hydroxide Solution
- 3 Reaction Vessels1 Measuring Tube
- Measuring Tube
 Collection Bottles
- 3 Collection Bot1 Titrator

DOT Info:

Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173.4 for domestic highway or rail transport only

IS9757

\$36.99

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MARNING:This product can expose you to Phenolphthalein which is known to the State of California to cause cancer. For more information go to www.P65Warnings.ca.gov

Plant Tissue Macronutrients

Plants need a variety of nutrients and minerals to grow and maintain overall health. The three main nutrients that plants require to survive and thrive are nitrogen, phosphorus, and potassium(also known as NPK). In this activity, students will extract nutrients from plants of your choice and perform analyses to determine if nitrogen, phosphorus, and potassium concentration is abundant, adequate, or deficient. Kit contains enough materials to perform 50 tests.

Kit Includes:

500mL	Nutrient Extractor Solution
30mL	Nitrogen Reagent #1
30g	Nitrogen Reagent #2
30g	Phosphorus Reagent #1
30mL	Phosphorus Reagent #2
30g	Potassium Reagent #1
120mL	Potassium Reagent #2

DOT Info: UN1170, Ethanol, 3, II, LTD QTY

IS9758

\$142.89

Reagent Scoops
Stir Stick

- Pk/100 Filter Paper
- 4 Collection Bottles
- 1 Spot Plate
- 5 Pipettes



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The Carbon Cycle in Soil

Changes in the carbon cycle and climate due to human activities have had detrimental effects on the environment. In this activity, students will learn about the carbon cycle in soil and how soil impacts global warming. Through 3 different activities, students will investigate one cause of global warming, and the effects of a positive feedback loop. Kit contains enough materials for 15 groups of students. Teacher's manual and student study guide copymasters are included.

Kit Includes:		
Limewater		
Topsoil		
Bromothymol Blue		
Sodium Bicarbonate		
Antacid Effervescent Tablets		

Nylon Tubing

- One-hole Stoppers
- 30 Soil Respiration Chambers
- 15 Straws

30



DOT Info: Non-regulated

IS9759

\$116.00

Estuary Monitoring Water Testing Kit

Test a water sample of your choice for coliform bacteria, chloride, dissolved oxygen, biochemical oxygen demand, nitrate, pH, phosphate, and temperature. Kit contains enough materials to perform 30 tests.

Kit Includes:

60mL	Starch Indicator Solution	5g
60mL	Manganese Chloride Reagent	3
60mL	Alkaline lodide Reagent	15
60mL	Sulfuric Acid Reagent, 50%	5
120mL	Sodium Thiosulfate Titrant	3
60mL	Griess Reagent	1/pl
0.5g	Cadmium Metal Powder	1/pl
300mL	Silver Nitrate Titrant	30
45mL	Salinity Indicator Solution	3

- g Coliform Test Powder
- 3 Reaction Vessels
- 15 Pipettes
- 5 Titrators
- 3 Measuring Tubes
- 1/pk Toothpicks
- 1/pk pH Strips
- 30 Phosphate Powder Pillows
- 3 Water Collection Bottles
- Illection Bottles

DOT Info:

UN2924, Flammable liquids, corrosive, n.o.s (Ethanol, Phosphoric Acid), 3(8), III, LTD QTY

WARNING: This product can expose you to chemicals including strong inorganic mists containing sulfuric acid, methanol, cadmium and cadmium compounds, and chromium hexavalent compounds, which are known to the State of California to cause cancer and reproductive harm. For more information go to http://www.P65Warnings.ca.gov.

IS9760

\$180.00

Soil Macronutrients Test Kit

Plants need a variety of nutrients and minerals to grow and maintain overall health. The three main nutrients that plants require to survive and thrive are nitrogen, phosphorus, and potassium (also known as NPK). In this activity, students will extract nutrients from a soil sample and perform analyses to determine if nitrogen, phosphorus, and potassium concentration is abundant, adequate, or deficient. Find problems with plants before they happen. In this kit, students can learn about the 3 main nutrients that contribute to plant health and how to measure their concentrations in soil. Kit contains enough materials to perform 50 tests.

Kit Includes:

120mL	Phosphorus Extraction Solution
30mL	Phosphorus Reagent
15g	Phosphorus Indicator Powder
120mL	Potassium Reagent
15g	Potassium Indicator Powder
30g	Nitrogen Indicator Powder

DOT Info: UN1170, Ethanol, 3, II, LTD QTY

IS9761

\$99.99

120mL	Soil pH Indicator
120mL	Nitrogen Extraction Solution
5	Pipettes
5	Soil Collection Bottles
5	Plastic Scoops





Aldon

New Products for 2025

Urban Water Testing Kit

Test water sample for chloride, copper, iron, hardness, nitrate, pH, phosphates, and temperature. This kit will allow students to learn about how water quality affects urban areas. Kit contains enough materials to perform 10 tests.

Kit Inclu	des:
300mL	Silver Nitrate Titrant
45mL	Salinity Indicator Solution
180mL	EDTA Titrant
60mL	pH Buffer 10 Solution
15mL	Eriochrome Black T Solution
60mL	Griess Reagent
0.5g	Cadmium Metal Powder
10	Iron Test Tablets
10	Copper Test Tablets

Water Collection Bottles Reaction Vessels Pipettes Titrators Measuring Tubes 1/pk Toothpicks 1/pk pH Strips Phosphate Powder Pillows Coliform Test Powder



DOT Info:

UN2924, Flammable liquids, corrosive, n.o.s (Ethanol, Phosphoric Acid), 3(8), III, LTD QTY

3

6

5 8

15

10 5g

WARNING:This product can expose you to chemicals including methanol, cadmium and cadmium compounds, and chromium hexavalent compounds, which are known to the State of California to cause cancer and reproductive harm. For more information go to http://www.P65Warnings.ca.gov.

IS9762 \$140.00

Total Water Investigation Kit

Test water sample for alkalinity, ammonia, biochemical oxygen demand, chloride, coliform bacteria, copper, hardness, iron, nitrate, dissolved oxygen, pH, and phosphate. Kit contains enough materials to perform 10 tests.

Kit Inclu	des:		
300mL	Silver Nitrate Titrant	60mL	50% Sulfuric Acid
15mL	Phenolphthalein Indicator	15mL	Total Alkalinity Indicator
45mL	Salinity Indicator Solution	120mL	Sodium Thiosulfate Titrant
180mL	Hydrochloric Acid Titrant	10	Iron Test Tablets
180mL	EDTA Titrant 1	10	Copper Test Tablets
5mL	Ammonia Test Solution 1	5g	Coliform Test Powder
15mL	Ammonia Test Solution 2	6	Reaction Vessels
15mL	Ammonia Test Solution 3	15	Pipettes
15mL	Eriochrome Black T Solution	5	Titrators
60mL	pH Buffer 10 Solution	8	Measuring Tubes
60mL	Griess Reagent	1/pk	Toothpicks
60mL	Starch Indicator Solution	1/pk	pH Strips
0.5g	Cadmium Metal Powder	10	Phosphate Powder Pillows
60mL	Manganese Chloride Reagent	3	Water Collection Bottles
60ml	Alkaline Iodide Reagent		

60mL Alkaline lodide Reagent

DOT Info: UN2924, Flammable liquids, corrosive, n.o.s (Ethanol, Phosphoric Acid), 3(8), III, LTD QTY

WARNING: This product can expose you to chemicals including Phenolphthalein chromium hexavalent compounds, strong inorganic mists containing sulfuric acid methanol, cadmium and cadmium compounds, which are known to the State of California to cause cancer and reproductive harm. For more information go to http://www.P65Warnings.ca.gov

IS9763

\$225.00

10



Featured Product

Aldon

www.aldon-chem.com

11

co·lab.orate

Enable students to work together, whether they are 6 feet apart or 6 streets apart.

 Co-lab-orate is a game changer for teachers and students. Whether in class or remote, students can get their work done and teachers can get insights into their process.

Co-lab-orate is an innovative digital lab notebook that allows teachers to easily assign lab reports to their classes, and lets their students work individually or together to complete hands-on activities that are easy to access and grade. With Co-lab-orate, students don't have to pivot because they suddenly find themselves learning from home, they just work the way they always do!

Co-lab-orate is packed with features that make teaching and learning easy and collaborative!

- Create new labs or import PDFs of labs you already have
- Easily deliver assignments to all students and classes
- Import photos, and create and edit graphs, tables, and equations
- Access, complete, or grade hands-on lab activities using a mobile device, tablet, or laptop
- Communicate with other students and teachers through comments, or work independently
- Integrates with Google classroom, Canvas, and Schoology

What does it cost?

• Our goal is to enable as many teachers and students as possible. Teachers can purchase a license for up to 150 students for only \$149 per year. That's less than \$1 per student!

IS4900

\$179.00

Aldon Chemical Inventory Management System VERSION 3

Access Over 2000 Chemicals in the Database! Now Flexible Enough to Add Your Own Chemicals!

IS4800 - One year site license, up to 20 buildings/district - \$204.57 IS4801 - Five year site license, up to 20 buildings/district - \$818.29

IS4805 - One year site license - 1 building - \$51.34



Our Inventory Management System allows customers to login to their own home page and set up a database that helps track inventory at different locations, such as buildings or even classrooms. You can easily add, modify, or delete chemical inventory items, and view information about each chemical. This data is easily accessible from any computer using a web browser, helping you reach your safety goals!

- Available anywhere, from any web browser.
- Access your chemical inventory during an emergency.
- · Add your own inventory into the database
- Track all of your chemicals by Building, Room and Cabinet location
- Find SDS, Storage codes and more.
- Ipad or tablet compatible.

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Distance Learning kits from Innovating Science

Designed for two groups of students, these kits are perfect for a distance learning setting. Check out our hands-on activities for chemistry and life science classes. Our distance learning kits are aligned to the Next Generation Science Standards and include an instruction manual.

Distance Learning: Elephant Toothpaste

This activity is a fun, safe way to teach students the

decomposition of hydrogen peroxide can be visualized

by mixing in a small amount of dish soap and adding

amount of foam is formed that flows up and out of the

reaction bottle. This kit contains an Instruction Manual

and enough materials to perform the demonstration

yeast as a catalyst. As the reaction occurs, a large

effect a catalyst has on a chemical reaction. The

Distance Learning: Acids and Bases

Students will be introduced to acids and bases, and test a dilute acid and dilute base with neutral litmus paper to learn how pH strips indicate if something is acidic or basic. Then, they will test common items that could be found around their household to determine if they are acids or bases. Extra litmus paper is included if students want to test more items found in their home to gain a greater understanding of acids and bases. The kit contains an Instruction Manual and enough materials for 2 groups.

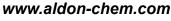


Distance Learning: Paper Chromatography

Chromatography is the oldest documented technique to separate chemical substances. In this activity, students will perform paper chromatography on three individual dyes, along with a mixture of dyes, to determine the composition of the mixture. Students will understand not only the components of a chromatography system, but also why different substances separate and move at different rates within the system. An Instruction Manual and enough materials for 2 groups are included.

Kit Includes:







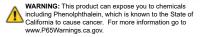
Distance Learning: Acids, Bases and the pH Scale

In this lab, students will gain an understanding of the basic differences between the properties of acids and bases, learn the role of hydrogen and hydroxide ions in acidic and basic substances, and comprehend the nature of the pH scale with regards to acid and base strength. Students will examine the effects of acids and bases on several pH indicators, determine the pH of several common household materials, and use the knowledge gained to determine the identity of four unknown solutions. Instruction Manual and enough materials for 2 groups are included.

- Kit Includes: 5mL Deionized water 5mL Dilute base (0.1M NaOH) 5mL Methyl red, 0.02% 5mL Phenolphthalein, 1.0% 5mL Household ammonia 5mL Filtered water 5mL Unknown solution #1 5mL Unknown solution #3 1 pkg Wide-range pH test strips
- 5mL Dilute acid (0.1M HCl) 5mL Litmus blue, 0.5% 5mL Bromothymol blue, 0.5% 5mL Vinegar 5mL Soap solution 5mL Vitamin C solution 5mL Unknown solution #2 5mL Unknown solution #4 2 Reaction trays



DOT Info: Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173.4 for domestic highway or rail transport only



IS1503

\$29.50

Distance Learning: Periodic Table - Nonmetals, Metals and Metalloids

During the nineteenth century, a Russian chemist name Dmitri Mendeleev began constructing a table of the elements. The result, called the Periodic Table of the Elements, is an organized classification of chemical elements based on certain properties of each element in relation to other elements. In this lab students will learn about three different categories of elements—metals, nonmetals, and metalloids. They will examine the physical properties of several elements and test their chemical reactivity, then use their results to classify each of the tested elements as either metal, nonmetal, or metalloid. Kit contains an Instruction Manual and enough materials for 2 groups of students.

Kit Includes: 4g Aluminum pellets 4g Carbon pellets 4a Silicon pellets 4g Zinc pieces 4g Sulfur pieces 6 Copper pieces 5g Sodium bicarbonate 6 Magnesium pieces 20mL Hydrochloric acid 1.0N 20mL Copper (II) chloride 14 Test tubes 4 Pipettes 2 Nails 6 Plastic cups, 5oz 1 Plastic scoop DOT Info: Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173.4 for domestic highway or rail transport only

IS1504

\$22.62



Distance Learning: DNA Extraction

In this lab students will learn the history of the discovery of DNA and DNA structure. They will understand the nature of genetic inheritance and the role of DNA and proteins in genetic expression while using biological detergents, enzymes, and ethanol to isolate DNA from plant material. You will need to supply the plant material. Kit contains an Instruction Manual and enough material for 2 groups of students.

Kit Includes: 50mL 7.5% SDS/1.5% NaCl 10mL Ethanol, 95% 2 Filters 4 Graduated pipettes 2 Plastic cups

1 Pepsin (to make 25mL of 0.5% solution) 2 Zipper bags 2 Plastic tubes 2 Stirrers

DOT Info: Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173.4 for domestic highway or rail transport only

IS1505

\$8.64



Distance Learning: Forensic

2 groups of students.

Chemistry of Unknown Substances

may encounter substances that they are not able to

analyze and identify them. In this experiment, students

will take on the role of a forensic scientist and use their

observation skills, senses, and a series of chemical tests

to determine the identity of two mystery substances. Kit

contains an Instruction Manual and enough materials for

4g Baking Soda

4 Reaction Plates

4g Mystery Substance #2

4mL Dilute Lugol's lodine

4g Plaster

4g Gelatin

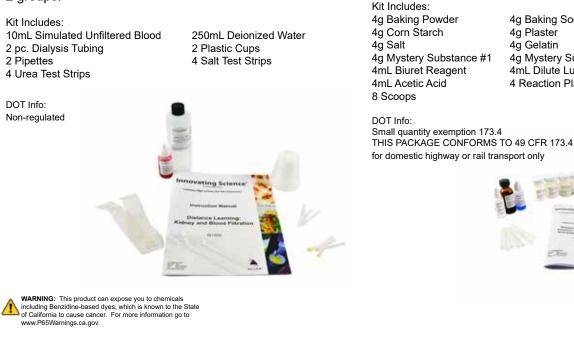
identify in the field. These substances will be sent to a crime lab, where a forensic scientist will work to

When collecting evidence at a crime scene, investigators

WARNING: This product can expose you to chemicals including Methanol/Methyl isobutyl ketone, which is known to the State of California to cause cancer and reproductive harm. For more information go to www.P65Warnings.ca.gov.

Distance Learning: Kidneys and Blood Filtration

Learn the role of the kidney in blood filtration and waste removal along with the many tasks performed by the functional units of the kidney, called nephrons, as well as nephron structure. Students will create an artificial kidney model to filter a simulated blood solution, and will be able to visually determine if filtration of the simulated blood occurred. They will also perform chemical tests on the resulting filtrate to detect any possible waste material that may have been removed by the artificial kidney. Kit contains an Instruction Manual and enough materials for 2 groups.



IS1506

\$15.95

IS1507

\$34.95



800-724-9877

Aldon

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Innovating Science®

Distance Learning: Forensic Chemistry of Blood Typing

Blood typing is a method of classifying blood based on the presence or absence of specific proteins, called antigens, on the surface of red blood cells. Blood type, an inherited characteristic, is valuable to know as it affects medical procedures, such as surgery and transfusions, and paternity testing, as well as serving as evidence in criminal investigations. Determining the blood type of a suspect in a crime can help provide supporting evidence that they are guilty, or eliminate them as a possibility. In this activity, students will determine the blood type of simulated blood samples collected from a crime scene, as well as from two suspects, to determine the likelihood of the suspects' involvement in the crime. Kit contains an Instruction Manual and enough materials for 2 groups.

Kit Includes: 4 Simulated Blood Samples 3mL Victim 3mL Suspect #1 3mL Suspect #2 3mL Crime Scene

1 set ABO/Rh Blood Typing Anti-sera (3mL of each) 8 Blood Typing Trays 1 Pkg Mixing Sticks

Note: This activity uses Innovating Science Simulated Blood and is safe for educational use.

DOT Info: Non-regulated

IS1508

\$24.41

Distance Learning: Forensic Chemistry - Drug Detection

Everyone who ate the school cafeteria's chili became ill. Could someone have tainted the chili? You are a forensic toxicologist, and it is up to you to determine if any of the ingredients in the cafeteria could have been substituted for aspirin, which appears to have been stolen from the nurse's office. You will perform a series of chemical tests on the cafeteria ingredients and a control sample of aspirin, to determine if all the ingredients are what they are supposed to

be. Kit contains an Instruction Manual and enough material for 2 groups.

1g Dehydrated yeast

1 Glucose QuickSolution (to make 30mL)

1 Sucrose QuickSolution (to make 30mL)

1 Lactose QuickSolution (to make 30mL)

Kit Includes: 4g Control Acetylsalicylic Acid 3 X 4g Chili Ingredients 4mL Dilute Lugol's lodine 4mL Ferric Nitrate Solution 0.2M 4mL Sodium Hydroxide 1.0N 4mL Deionized Water 2 Micro-reaction Plates 1 box Toothpicks 4 Scoops DOT Info: Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173.4 for domestic highway or rail transport only WARNING: This product can expose you to chemicals including Acetyl Salicylic Acid/Aspirin, which is known to the State of California to cause reproductive harm. For more IS1509 \$24.99 information go to www.P65Warnings.ca.gov.

Distance Learning: Cellular Respiration - What Sugar Does Yeast Like Best?

In this experiment, students will expose living yeast cells to three different sugars-glucose, sucrose, and lactose. The yeast cells should begin to utilize the sugars as a food source if they are capable of metabolizing them, and will begin to engage in aerobic respiration and/or fermentation. This activity allows students to understand that yeast may use different options for energy production, with varying degrees of effectiveness. A pH indicator will be used to indirectly determine the effectiveness of the three different sugars as a food source for the yeast. Kit contains an Instruction Manual and enough material for 2 groups of students.

Kit Contents: 10mL 0.5% Litmus 5mL 0.1M Sodium hydroxide 6 Plastic cups 2 Plastic scoops 2 Pipettes

DOT Info: Non-regulated

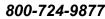
IS1510



Aldon

www.aldon-chem.com

\$19.03





Distance Learning: Chromatography of Plant Pigments

Chlorophyll is the most prevalent and well-known plant pigment related to photosynthesis. It is not, however, the only pigment necessary for photosynthesis to occur. Other pigments are also involved in the process, though they are often overlooked as they tend to be masked by the abundance of the green pigment chlorophyll. In this activity, students will use paper chromatography to separate the various pigments from a sample of plant pigment extract. Students will not only confirm the presence and learn the role of these "hidden" pigments, but will also learn about chromatography as a technique for separating molecules. Kit contains an Instruction Manual and enough materials for 2 groups of students.

Kit Includes: 30mL Chromatography solvent 2pcs Chromatography paper 2 Plastic cups, 10oz

20mL Plant pigment extract 2 Microscope slides 2 Petri dishes

DOT Info: Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173.4 for domestic highway or rail transport only

WARNING: This product can expose you to chemicals including Methanol/Methyl isobutyl ketone, which is known to the State of California to cause cancer and reproductive harm. For more information go to http://www.P65Warnings.ca.gov

IS1511 \$20.50



Distance Learning: Mohs Hardness Test

The Mohs hardness scale was invented by Friedrich Mohs in 1812, and has been a useful tool for geologists to determine the identity of different minerals. With this kit, students will become familiar with the Mohs hardness scale and how to perform the hardness test, allowing them to develop vital skills that will help them throughout their studies in earth science. This kit also provides students an opportunity to go out into their local environment to investigate the relative hardness of the minerals around them. Kit contains an Instruction Manual and enough materials for 2 groups of students.

Kit Includes: 2 pcs Mineral A 2 pcs Mineral C 1 Aluminum foil piece, 12"x6" 2 Streak plates

DOT Info: Non-regulated

IS1512



WARNING: This product can expose you to chemicals including Respirable crystalline silica, which is known to the State of California to cause cancer. For more information go to http://www.P65Warnings.ca.gov.



Distance Learning: Urinalysis Using Simulated Urine

2 pcs Mineral B

2 pcs Mineral D

2 Scratch plates

Urinalysis is one of the oldest medical diagnostic tests, and is still one of the most commonly used tests to this day. In this activity, students will perform a urinalysis on four different samples of simulated urine. Using a standard urinalysis test strip, students will determine the pH of the samples, and test for the presence or absence of glucose, protein, and ketones. Students will then analyze the results of the tests to determine a potential diagnosis for each patient. Kit contains an Instruction Manual and enough material for 2 groups of students.

Kit Includes: 4 Simulated urine samples, 40 mL each Patient A Patient B Patient C Control 8 Urine test strips 8 Plastic cups 2 Color charts

DOT Info: Non-regulated

IS1513

\$29.99



Aldon

Distance L Learning

Innovating Science®

Distance Learning: Osmosis and Diffusion

This lab allows students to learn about two forms of passive transport, diffusion and osmosis, and compare and contrast the similarities and differences between the two processes. Students will first investigate the process of diffusion, and determine the effect of solution temperature on the rate of diffusion of a solute. Then, they will set up an environment that is likely to facilitate osmosis, and gather data to determine whether or not osmosis occurred over a set period of time. Kit contains an Instruction Manual and enough material for 2 groups of students.

Kit Includes: 1 Sucrose QuickSolution (to make 20 mL of 1M solution) 3mL Blue food coloring 20mL Deionized water 2 pcs Dialysis tubing 4 Plastic cups 2 Graduated pipettes 2 Plastic vials

DOT Info: Non-regulated

IS1514

\$15.68

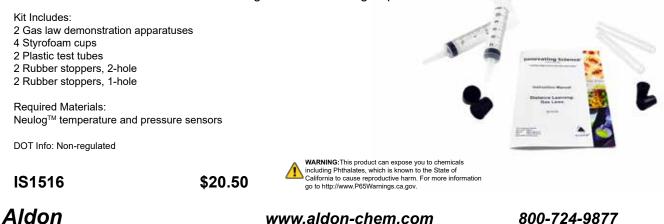
Distance Learning: A Safer Flame Test

The flame test is an analytical technique often used for the identification of certain elements, primarily metal ions. The color of the flame is observed, and the spectra of light emitted from the flame is viewed through a spectroscope. The traditional flame test poses safety concerns for students in a laboratory setting, however, this kit minimizes these concerns by utilizing small candles that are easier to manage. There is no loss in the clarity of the emission spectra, making it easy for students to observe and deduce what element is causing the colored flame. Kit contains an Instruction Manual and enough materials for 2 groups of students.

Kit Includes: 2mL Mineral salt alcohol solution 2 Diffraction grating slides 85g Sand 1 Spectroscope 4 Color flame candles 1 Micro-burner 4 Candle holding trays DOT Info: Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173.4 for domestic highway or rail transport only WARNING: This product can expose you to chemicals including Methanol and Respirable crystalline silica, which are known to the State of California to cause cancer and reproductive harm. For more information go to http://www.P65Warnings.ca.gov. IS1515 \$41.00

Distance Learning: Gas Laws

Gases are, for the most part, invisible, and therefore cannot be observed as readily as a solid or a liquid. Because of this, they are often examined and analyzed through measurable physical characteristics such as volume, pressure, temperature, and moles (number of particles). These properties have simple, consistent mathematical relationships that led to the development of the gas laws (Boyle's, Charles's, and Gay-Lussac's), and eventually the ideal gas law. By using these relationships you can determine the value of an unknown property based on the other three, or predict the effect that varying one component will have on the others. In this activity, students will use the Innovating Science apparatuses and your temperature and pressure sensors to measure and plot the physical characteristics of a gas. Kit contains an Instruction Manual and enough materials for 2 groups of students.



Distance Learning: Introduction to Chemical Properties

How do scientists identify unknown chemicals? What makes one solution different from another? This activity will help students answer these questions, and understand the impact different chemicals can have when added to various solutions. Students will test four unknown solutions and determine their identity by comparing their chemical properties to a variety of known samples. Kit contains an Instruction Guide and enough materials for 2 groups of students.

Kit Includes:

6mL Acetic acid, 1.0 M 6mL Ammonium hydroxide, 1.0 M 6mL Cupric sulfate, 0.1 M 6mL Hydrochloric acid, 0.1 M 6mL Sodium bicarbonate, 0.1 M 6mL Sodium carbonate, 0.1 M 6mL Sodium hydroxide, 0.1 M

DOT Info: Non-regulated

IS1517

3mL Unknown A 3mL Unknown B 3mL Unknown C 3mL Unknown D 2 Preprinted acetate sheets

6mL Bromothymol blue indicator

Distance Learning: Introduction to Ionic Reactions

\$20.82

This activity is designed to introduce students to ionic reactions. Students will test a series of microscale chemical combinations to learn how to identify if a chemical reaction has occurred. The contents of this kit have been optimized to minimize pre-lab preparation, waste generation, and risks of cross-contamination. Kit contains an Instruction Guide and enough materials for 2 groups of students.

Kit Includes:

6mL Sodium acetate, 0.1 M 6mL Sodium carbonate, 0.1 M 6mL Sodium chloride, 0.1 M 6mL Sodium ferrocyanide, 0.1 M 6mL Sodium hydroxide, 0.1 M 6mL Sodium iodide, 0.1 M 6mL Sodium oxalate, 0.1 M 6mL Sodium phosphate, 0.1 M 6mL Sodium sulfate, 0.1 M 1 box Toothpicks 2 Preprinted acetate sheets

DOT Info:

Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173.4 for domestic highway or rail transport only



Innovating Science

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Distance Learning: ABO/Rh Blood Typing

This popular kit that has now been scaled down for distance learning provides the most procedurally accurate simulation of the blood typing technique available. Test and determine the ABO/Rh blood types of four different simulated blood samples. Using Innovating Science's new simulated blood, combine blood samples and antisera, gently agitate the blood typing tray, and observe the results. No toothpicks, no stirring, and no waiting for results required. Instruction Manual included. Use gloves while performing tests.

Kit Includes: 4 Simulated Blood Samples Donor #1 Donor #2 Donor #3 Donor #4 1 btl Simulated Anti-A Serum 1 btl Simulated Anti-B Serum 1 btl Simulated Anti-Rh Serum 12 Blood Typing Trays



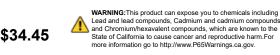
DOT Info: Non-regulated

IS3110

\$23.63

Aldon

IS1518 \$



www.aldon-chem.com

100110

AP® Biology Investigation #1: Artificial Selection

Students will study the process of artificial selection using Innovating Science fast growing plants. Students will identify phenotypic differences to observe. They will then use selective cross pollination to examine the prevalence of that phenotype in successive generations. Meets AP Science Practices 2, 5, and 7, and Big Idea 1. Materials provided for 32 students in eight lab groups.

Kit Includes:

- 1 Bag Potting Soil
- 1 Bag Vermiculite
- 1 pkg 7 oz cups
- 1 pkg 5 oz cups
- 8 Magnifers
- 1 Nutrient Quick Solution to make 1 Liter
- 1 pkg Innovating Science Fast Growing Plant Seeds
- 1 pkg Cotton Swabs
- 1 pkg Hydroponic Wicks

DOT Info: Non-regulated

IS3701



Brassica Rapa Seeds Pk/200

IS3733 \$20.54

\$84.00

AP[®] Biology Investigation #4: Diffusion and Osmosis

Students will study the movement of water and nutrients across a cell membrane and observe osmosis in living tissue. They will then investigate the relationship between surface area and volume as it relates to cells and diffusion. They will also examine the concept of molarity and how it relates to osmotic potential and the movement of water. Students will be able to explain how cell size and shape affect the overall rate of nutrient intake and water elimination. Students will use plant tissue to determine the molarities of unknown solutions based on the direction and degree of water movement. This kit contains enough materials for 8 groups. Teacher's manual and Student Study Guide copymasters are included. Meets AP[®] Science Practices 2, 4, and 5, and Big Idea 2.

Kit Includes: 45g Adar 40 pc. Dialysis tubing 15mL Bromothymol blue concentrate Plastic cups, 7oz 48 2x25mL 2.0M Hydrochloric acid Microscope slides,pkg/72 8 Agar block casting trays Coverslips 8 Plastic cups, 5oz Cork borer 8 Plastic knives 8 Metric rulers 8 Plastic stirrers Sucrose QuickSolution (to make 1L of 1.0M solution) Sodium chloride QuickSolution (to make 1L of 1.0M solution) Glucose QuickSolution (to make 1L of 1.0M solution) Ovalbumin QuickSolution (to make 1L of 5.0% solution) Food coloring set (red, blue, yellow, green) Sucrose QuickSolution set to make: 1L of 0.2M solution (Solution #3) 1L of 0.4M solution (Solution #2) 1L of 0.6M solution (Solution #5) 1L of 0.8M solution (Solution #1) 1L of 1.0M solution (Solution #4) 1L distilled water (Solution #6) DOT Info: Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173.4 for domestic highway or rail transport only WARNING: This product can expose you to chemicals including Lead and lead compounds. which are known to the State of California to cause cancer, reproductive harm. For more information go to www.P65Warnings.ca.gov. IS3704 \$176.00

AP[®] Biology Investigation #5: Photosynthesis

Students will learn the necessary components and conditions for photosynthesis to occur while using leaf disks to measure the accumulation of oxygen and relate it to the rate of photosynthesis. Students will then use guided inquiry to design and conduct an experiment to examine the effects of a chosen variable on the rate of photosynthesis. Teachers manual and Student Study Guide copymasters are included. There are enough materials provided for 8 lab groups. This lab meets AP Science Practices 1, 2, 3, 6, and 7, and Big Idea 2.

Kit	Includes:	

1	Hole Punch
16	Dispensers 10ml

- 16 Dispensers 10mL 16 Plastic Cups
 - Plastic Cups
- 30mLDilute Soap Solution50gSodium Bicarbonate
- og Soulum Bicarbohat



DOT: Non-regulated

IS3705

\$64.50

Aldon

AP[®] Biology Investigation #6: Cellular Respiration

Students will use a respiration chamber to measure and record the rate of oxygen consumption (cellular respiration) using germinating seeds versus a non-germinating control sample. Teachers manual and Student Study Guide copymasters are included. There are enough materials provided for 8 lab groups. Meets AP Science Practices 1, 2, 3, 6, and 7, and Big Idea 2.

Kit Includes: 1 pkg. pea seeds 24 respiration chambers 48 washers

1 btl. 15% KOH, 30 mL

1 tube petroleum jelly sealant

1 pkg. plastic beads 24 rubber stoppers 24 graduated pipets, 1 mL 1 pkg. non-absorbent cotton 1 pkg. absorbent cotton 8 trays

Refill Kit available **IS3706-REF** \$53.00





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IS3706

\$410.17

AP[®] Biology Investigation #8: Biotechnology: Bacterial Transformation

Students will genetically engineer bacteria through transformation with a plasmid that confers antibiotic resistance. They will then examine the number of resistant bacteria to determine the efficiency of the transformation procedure. Teachers manual and Student Study Guide copymasters are included. Materials provided for 8 lab groups. Meets AP Science Practices 1, 3, 5, 6, and 7, and Big Idea 3.

Kit Includes:

30mL Sterile Calcium Chloride 0.05M 8 Pipet Bulb 16 Microcentrifuge Tube 1.5mL 40 Petri Dish 1 Luria Broth 9mL Tube

16 Sterile Volumetric Pipet 8 Micropipet w/Plunger 10ul 8 Sterile Inoculating Loop 4 Luria Agar 200mL

Kit Contains coupon for perishable materials. Redeem by fax, phone or e-mail. Materials sent upon redemption of coupon:

2 btls 0.02g Ampicillin

- 1 Tryptic Soy Broth Tube 9mL
- 1 Pipet, Sterile, Disposable
- 1000 ng pUC 19 Plasmid 1 Tryptic Soy Agar Slant 6mL 1 E. Coli Culture (freeze dried)

DOT: Non-regulated

IS3708

\$231.30

AP[®] Biology Investigation #7: Cell Division: Mitosis and Meiosis

Students will study the cellular steps involved in DNA replication and cell division in both mitosis and meiosis. This investigation will allow students to examine the stages of mitosis in the preparation of plant root tips. Crossing over in meiosis will also be investigated through the use of Sordaria cultures. Meets AP Science Practices 1, 5, 6, and 7, and Big Idea 3. Materials provided for 32 students in eight lab groups.

Kit Contains coupon for perishable materials. Redeem by fax, phone or e-mail.

- 10 Disposable scalpels, sterile
- 20 Petri dishes, sterile
- Sordaria agar, 200mL
 - Sordaria mating agar, 200mL 1
- Microscope slides 1
- 8 Disposable inoculating loops, sterile
- Hydrochloric acid, 6.0M, 30mL 1
- Toluidine blue, 1%, 30mL 1

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- 8 Forceps
- 16 Plastic pipettes Coverslips 1
- Sand, 500g
- Plastic cups 8



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\$215.88



IS3707

AP[®] Biology Investigation #9: Biotechnology: Restriction Enzyme Analysis of DNA

In this lab students will understand the action and specificity of restriction enzymes, while learning the components involved in the process of DNA electrophoresis and the role of each component as it applies to the entire process of electrophoresis. Students will perform the electrophoresis process on DNA samples treated with different restriction enzymes and construct a standard curve using a known DNA sample while determining the approximate size of the DNA fragments in unknown samples. Meets AP Science Practices 3 and 6, and Big Idea 3. Materials provided for eight lab groups.

Kit Includes: Prepared agarose, 0.8%, 200mL TBE buffer, 5X, 500mL DNA Stain, 20X, 60mL DNA Samples, 100ul each: DNA Marker (Lambda DNA HindIII digest) Sample #1 (Lambda DNA) Sample #2 (Lambda DNA BstEII digest) Sample #3 (Lambda DNA BstEII digest) Sample #4 (Lambda DNA BstEII digest)



DOT: Non-regulated

IS3709

\$139.00

AP® Biology Investigation #11: Transpiration

Students will study the process of transpiration/transpiration pull and the role it plays in water/nutrient movement in plants. They will study the role of stomata in relation to the transpiration process using a potometer to examine the rate of transpiration in a bean seedling under a select set of environmental conditions. They will also examine the location/density of stomata on a leaf surface by preparing a stomatal peel. Lastly students will design and conduct an experiment to show how altering an environmental condition may affect the rate of transpiration. Meets AP Science Practices 1, 2, 4, 6, and 7, and Big Idea 4. Materials provided for 32 students in eight lab groups.

Kit Includes:

8 pc	Clear tubing, 12"	8	Tubing clamps
8	Pipettes, 1mL	8	Dispensers, 10mL
8	Planting trays, 6 cell	1 bag	Potting soil
1 pkg	Bean seeds	1 tube	Petroleum jelly
1 btl	Nail polish	1 pkg	Microscope slides



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IS3711

\$119.99

AP® Biology Investigation #12: Fruit Fly Behavior

Students will study whether fruit flies will move toward or away from important chemicals and food that aid in their survival. Using the Innovating Science choice behavior chamber, behaviors of the fruit-flies are observed and any pattern can be identified. Students will formulate their own theories based on the fruit flies' response to the chemicals and foods, and then determine what materials and experiments should be tested further. Meets AP Science Practices 1, 3, 4, 5, 6, and 7, and Big Idea 4. Materials provided for eight lab groups.

Kit Includes: 30mL Ethyl Alcohol 30mL Ammonia 5% 30mL Deionized Water 30mL Vinegar 8 Innovating Science[®] Behavior Choice Chambers Fruit Fly Vials

Kit Contains coupon for perishable materials. Redeem by fax, phone or e-mail. Materials sent upon redemption of coupon: 8 Fruit Fly Cultures

DOT: Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173.4 for domestic highway or rail transport only

WARNING: This product can expose you to chemicals including Methanol/Methyl isobutyl ketone, which is known to the State of California to cause cancer, reproductive harm. For more information go to www.P65Warnings.ca.gov.

IS3712



\$228.00

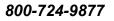
IS3712-CHOICE Innovating Science Behavior Choice Chambers pkg/8 \$132.61

AP® Biology Investigation #13: Enzyme Activity

Students will learn the nature and specificity of enzyme-catalyzed reactions. Students will then use an extract of turnips to examine the reaction between hydrogen peroxide and the enzyme peroxidase while testing one or more factors that influence the rate of enzyme reactions. In this lab students will develop data collection strategies and analyze their results. This kit contains enough materials for 8 groups. Teachers manual and Student Study Guide copymasters are included. Meets AP Science Practices 4, 5, 6, and 7, and Big Idea 4.



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AP[®] Chemistry Guided Inquiry Reflect the Updated AP® Chemistry Curriculum!

Lab #1: Effect of Concentration on **Transmitted Light**

Students will be guided through an investigation to study food dyes and determine how the absorbance of light can be used to study color and determine concentrations of chemicals in solutions. The activity contains enough materials for 15 groups of students as well as a Teacher's Guide and Student Study Guide Copymasters. This lab meets Big Idea 1, Investigation 1, and Primary Learning Objective 1.15.

Kit Includes:

- Food Dye Blue #1, 0.5% 10mL 8 **Test Tubes**
- 4 Graduated Cylinders

DOT: non-regulated

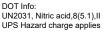


Lab #2: Beer's Law - Mass Percent of **Copper in Brass**

Students will design a laboratory procedure to analyze the amount of copper in brass using a spectrophotometer. Students identify the correlation among wavelength, absorbance, and concentration for each of three possible ions that may be obtained from brass: copper, zinc, and iron. The activity contains enough materials for 15 groups of students as well as a Teacher's Guide and Student Study Guide Copymasters. This lab meets Big Idea 1, Investigation 2, and Primary Learning Objective 1.16.

Kit Includes: 50mL Cupric Nitrate 0.1M 50mL Ferric(III) Nitrate 0.1M 50mL Zinc Nitrate 0.1M 100mL Cupric Nitrate 0.4M **Brass Pellets** 20g

50mL Cupric Sulfate 0.1M 50mL Ferric Sulfate 0.1M Zinc Sulfate 0.1M 50mL 100mL Nitric Acid, 70%



WARNING: This product can expose you to chemicals including Strong inorganic acid mists containing sulfuric acid, which are known to the State of California to cause cancer. For more information go to www.P65Warnings.ca.gov



IS8102

\$51.38

Lab #3: What Makes Water Hard?

Students will investigate the suitability of gravimetric analysis for determining the amount of water hardness in the form of calcium carbonate found in various water samples. The activity contains enough materials for 15 groups of students as well as a Teacher's Guide and Student Study Guide Copymasters. Meets Big Idea 1, Investigation 3, Primary Learning Objective 1.19.

Kit Includ 40g 50g 200mL 200mL 200mL 200mL 200mL 200mL 200mL	les: Calcium Chloride Sodium Carbonate Sodium Carbonate, 0.5M Filter paper, pk/100 Water Sample #1 (0.75M calcium chloride) Water Sample #2 (0.2M calcium chloride) Water Sample #3 (0.5M calcium chloride) Water Sample #4 (0.1M calcium chloride) Water Sample #5 (0.05M calcium chloride) Water Sample #6 (0.9M calcium chloride)	
DOT: Noi IS810	n-regulated 3	\$42.00

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Lab #4: Acid in Fruit Juices and Soft Drinks

Study how the concentration of acids in various consumer beverages may be determined by titration with sodium hydroxide. Students will determine the proper indicator to use in the titration of a weak acid. Students will create an experiment to calculate the molar concentration of acid in a beverage. The activity contains enough materials for 15 groups of students as well as a Teacher's Guide and Student Study Guide Copymasters. This lab meets Big Idea 1, Investigation 4, Primary Learning Objective 1.20.

Kit Includes:

10g 100mL 500mL 25mL 1	Potassium hydrogen phthalate Hydrochloric Acid, 0.1M Sodium Hydroxide, 0.1M Methyl Red 0.02% pH Strips 1-14 Pkg/100	4 X 25g 100mL 25mL 25mL	3 Sodium Hydroxide Acetic Acid 0.1M Phenolphthalein, 1% Bromothymol Blue 0.04%
THIS PACK	l quantity exemption 173.4 AGE CONFORMS TO 49 CFR omestic highway or rail transport only		

WARNING: This product can expose you to chemicals including Phenolphthalein, which is known to the State of California to cause ncer. For more information go to www. P65Warnings.ca.gov



Calcium Carbonate

Cupric Sulfate, anhydrous

Sodium Bicarbonate

Potassium Nitrate Salicylic Acid

Calcium Metal

Ethanol, 95%

Zinc Metal

Sucrose

Urea

IS8104

\$45.00

Lab #6: What's In That Bottle?

Students will identify unknown chemicals based on laboratory testing of their physical and chemical properties. Students will identify the 4 different kinds of bonds that exist in chemicals: ionic, polar covalent, nonpolar covalent and metallic. Students review the properties of each solid using various tests. The activity contains enough materials for 15 groups of students as well as a Teacher's Guide and Student Study Guide Copymasters. Meets Big Idea 2, Investigation 6, Primary Learning Objective 2.22.

Kit Includes:

10g	Ammonium Chloride	10g	
10g	Magnesium Oxide	10g	
10g	Benzoic Acid	10g	
10g	Aluminum Metal	10g	
10g	Paraffin Wax	10g	
10g	Sodium Acetate	10g	
10g	Sodium Carbonate	10g	
10g	Copper Metal	10g	
10g	Cupric Sulfate, pentahydrate	10g	
10g	Sodium Chloride	10g	
30mL	Universal Indicator	30mL	
30mL	Sodium Hydroxide 0.1M	30mL	
30mL	Hexane	30mL	
30mL	Deionized Water	32	
LED Conductivity Tester			

DOT Info: Small quantity exemption 173.4 This package conforms to 49 CFR 173.4 for domestic highway or rail transport only

IS8106 \$82.99

WARNING: This product can expose you to chemicals including Respirable crystalline silica. Phenolphthalein/Methanol/Methyl isobutyl ketone and n-Hexane, which are known to the State of California to cause cancer, reproductive harm. For more information go to www.P65Warnings.ca.gov

Lab #5: Separation of Molecules

Students collect data using different solvents to identify the optimal solvent for separation. They will then illustrate the intermolecular forces that are acting on the molecules in the separation. Students evaluate the chromatograph with different solvents and establish a connection between molecular structure and intermolecular attraction to the solvent. The activity contains enough materials for 15 groups of students as well as a Teacher's Guide and Student Study Guide Copymasters. Meets Big Idea 2, Investigation 5, Primary Learning Objective 2.10.

Kit Includes:

30mL	Food Dye Blue #1, 0.5%
30mL	Food Dye Yellow #5, 0.5%
30mL 200ml	Food Dye Red #40, 0.5% 9.1 Petroleum ether/Acetone
200mL 200mL	Ethyl Alcohol, 95%
200mL	Isopropyl Alcohol, 99%
200mL	Acetone
200mL	Deionized Water
15	Chromatography Paper
30	Glass Vials
4	Capillary Tubes
THIS PACK 173.4 for do	I quantity exemption 173.4 AGE CONFORMS TO 49 CFR omestic highway or rail transport only NG: This product can expose you to als including Methanol, which is known tate of California to cause reproductive For more information go to www. mings.ca.gov.
IS810 \$132.	- And a subset of a



Lab #7: Green Chemistry and Purification

First, students will design their own experiment to separate two substances using green chemistry principles. Students will also design and perform an experiment to quantitatively measure the weight percent of the mixture. The activity contains enough materials for 15 groups of students as well as a Teacher's Guide and Student Study Guide Copymasters. Meets Big Idea 3, Investigation 7, Primary Learning Objectives 3.5 and 3.3.

Kit Includes:

300g 85% Sodium Bicarbonate/ 15% Sodium Carbonate Mixture

DOT Info: Non-regulated



Lab #8: Determination of the Actual Percentage of Hydrogen Peroxide

Students will determine the actual concentration of the hydrogen peroxide in the bottle by titration and determine if it is lower than the value on the label. Hydrogen peroxide will degrade over time, and students will determine how much it degrades. The activity contains enough materials for 15 groups of students as well as a Teacher's Guide and Student Study Guide Copymasters. Meets Big Idea 3, Investigation 8, Primary Learning Objective 3.9.

Kit Includes:

1000mL	Iron Ammonium Sulfate 0.1M
1000mL	Potassium Permanganate 0.02M
1000mL	Sulfuric Acid, 6M
250mL	Hydrogen Peroxide 3%
250mL	Hydrogen Peroxide 6%
500mL	Deionized Water
DOT Info:	

WARNING: This product can expose you to chemicals including Strong inorganic acid mists containing sulfuric acid, which are known to the State of California to cause cancer. For more information go to www.P65Warnings.ca.gov.

DOT Info: UN2796, Sulfuric acid,8,II,Ltd Qty



Lab #9: Examining the Composition of a Pain Reliever

Students will test the solubility of each possible component of a commercially available pain reliever in an organic solvent, ethyl acetate, and in a basic aqueous solution of sodium bicarbonate. These results will help the student create a procedure that will be used to separate components in a mixture and determine percent composition. The activity contains enough materials for 15 groups of students as well as a Teacher's Guide and Student Study Guide Copymasters. Meets Big Idea 3, Investigation 9, Primary Learning Objective 3.10.



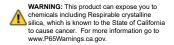
Lab #10: How Long Will That Marble Statue Last?

Students will observe and measure the evolution of carbon dioxide gas from the decomposition of calcium carbonate when mixed with an acid. Students will also create experiments to determine the rate of reaction with different concentrations of hydrochloric acid. The activity contains enough materials for 15 groups of students as well as a Teacher's Guide and Student Study Guide Copymasters. Meets Big Idea 4, Investigation 10, Primary Learning Objective 4.1.

Kit Includes:			
200g	Marble Chips		
500mL	Hydrochloric Acid, 1.0M		
2x500mL	Hydrochloric Acid, 3.0 M		

DOT Info: UN1789. Hydrochloric Acid. 8.III Ltd Qtv

500mL Hydrochloric Acid, 6.0M Silicone Tubing Dispensers,10mL





IS8110 \$105.00

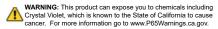
Lab #11: Rate Law of the Fading of a Dye Using Beer's Law

15 15

Students will determine the rate law for the reaction of crystal violet and sodium hydroxide. Students will also prepare dilutions of stock crystal violet solutions to generate a Beer's law calibration curve. This lab will require students to integrate prior chemistry knowledge involving spectroscopy, Beer's law, solution dilution, calibration curves, and chemical kinetics. The activity contains enough materials for 15 groups of students as well as a Teacher's Guide and Student Study Guide Copymasters. Meets Big Idea 4, Investigation 11, Primary Learning Objective 4.2.

Kit Includes: 1500ml Crystal Violet, 1% 1500mL Sodium Hyroxide 0.2M 1000mL **Deionized Water**

DOT Info: UN1824, Sodium hydroxide solution,8,II,Ltd Qty





IS8111 \$36.50

Lab #12: Designing an Effective Hand Warmer

Students study the various energy changes that occur with the formations of solutions for laboratory salts. From this data they will create the best and safest hand warmer. Students will determine the heat of solutions for each solid and analyze the cost and safety information with provided safety data sheet. The activity contains enough materials for 15 groups of students as well as a Teacher's Guide and Student Study Guide Copymasters. Meets Big Idea 5, Investigation 12, Primary Learning Objective 5.7.

> Sodium Carbonate Sodium Acetate

Sodium Chloride

200g

100g	Calcium Chloride
100g	Lithium Chloride
100g	Ammonium Nitrate

100g 100g 100g Magnesium Sulfate

DOT Info: UN1942, Ammonium nitrate, 5.1, III, Ltd Qty



IS8112

\$52.99

Lab #13: Le Chatelier and the Colors of the Rainbow

Students will investigate Le Chatelier's principle and why it works. They will also investigate this principle by testing several systems at equilibrium and then selecting specific ones to produce the colors of the rainbow based on specific applications of Le Chatelier's principle. Students will then be challenged by selecting which reaction system to use for which color in producing the rainbow while trying to only use a given "stress" once. The activity contains enough materials for 15 groups of students as well as a Teacher's Guide and Student Study Guide Copymasters. Meets Big Idea 6, Investigation 13, Primary Learning Objective 6.9.



Lab #14: Structure & Concentration In Acid and Base Titrations

Students will conduct a series of acid–base titrations and determine the concentrations of two unknowns. They will create a procedure to collect quantitative titration data using a buret and pH meter. Using titration data that they collected, they will determine the concentration of each unknown. The activity contains enough materials for 15 groups of students as well as a Teacher's Guide and Student Study Guide Copymasters. Meets Big Idea 6, Investigation 14, Primary Learning Objective 6.13.

Kit Includes: 4 X 500mL 4 X 500mL 2 X 1 L 4 X 500mL 4 X 500mL 4 X 500mL 4 X 500mL	Hydrochloric Acid 0.2M,Part A Sulfuric Acid 0.1M, Part B Nitric Acid 0.05M, Part C Calcium Hydroxide 0.1M Ammonium Hydroxide 0.2M, Part E Acetic Acid 0.1N, Part F Sodium Hydroxide 0.1N	
DOT Info: UN2796, Sulfuric acid,8	8,II,Ltd Qty	
to chemicals includi acid mists containin are known to the Sta	g sulfuric acid, which ate of California to nore information go to	
IS8114	\$57.00	

Lab #15: Buffering Activity of Common Household Products

Many household products contain buffering chemicals. Students will design a procedure to determine the buffering agents that are in different household products such as foods, beverages and over the counter drugs. The activity contains enough materials for 15 groups of students as well as a Teacher's Guide and Student Study Guide Copymasters. Meets Big Idea 6, Investigation 15, Primary Learning Objective 6.20.

Kit Includes:4 X 500mLHydrochloric Acid 0.1N4 X 500mLSodium Hydroxide 0.1N500mLCitric Acid Solution 0.02M2 X 500mLIsopropyl Alcohol 90%

DOT Info: UN1219, Isopropanol solution,3,II,Ltd Qty



IS8115

\$51.00

Lab #16: Preparation of Effective Buffers

Students will design a buffer that can maintain a pH within a narrow range when certain amounts amounts of acid and base are added. The activity contains enough materials for 15 groups of students as well as a Teacher's Guide and Student Study Guide Copymasters. This lab meets Big Idea 6, Investigation 16, Primary Learning Objective 6.18.

- Kit Includes:
- 1000mL 0.1M acetic acid
 1000mL 0.1M ammonia
 1000mL 0.1M sodium dihydrogen phosphate
 1000mL 0.1M citric acid
 1000mL 0.1M sodium monohydrogen citrate
 1000mL 0.2M sodium hydroxide
 1000mL 0.2M hydrochloric acid
 1000mL pH 7 buffer

DOT Info: UN1824,Sodium hydroxide solution,8,II,Ltd Qty

- Sodium acetate
- Sodium hydrogen phosphate

25g

25g

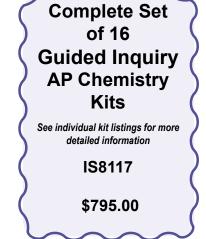
25g

50g

50g

25g

- Ammonium chloride Sodium dihydrogen citrate
- Sodium dihydroge Sodium citrate
- Sodium citrate Sodium chloride





IS8116

\$92.00

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AP®Chemistry - Classics

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With *Innovating Science's*[®] 46 comprehensive lab activities, you will find everything to satisfy the AP Chemistry[®] laboratory requirements. Each lab activity includes all the chemicals you will need for each experiment plus a full Teacher's Guide and Student Study Guide copymasters. All activities contain enough materials for 15 groups of students.

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Thermochemistry and Hess' Law

Employ Hess's Law of enthalpy based on the observed data for two reactions. Students can measure the temperatures of the reaction in a calorimeter and analyze the data calculating the enthalpy of each reaction.

Kit Includes:8 x 250mLHydrochloric Acid, 2M/2N Solution8 x 250mLSodium Hydroxide, 2M/2N Solution2 x 500mLAmmonium Chloride, 2M Solution2 x 500mLAmmonium Hydroxide, 2M Solution

DOT Info: UN1789, Hydrochloric acid, 8, III, Ltd Qty UN1824, Sodium hydroxide solution, 8,III, Ltd Qty



IS8001

Mole Ratio of Reactants

\$49.99

Using the method of continuous variation, two solutions are combined in various ratios. To select the ratio that produces the most product or consumes the most reactants, students must find an empirical method which is proportional to the amount of reaction that occurs. The reaction selected for this experiment is exothermic and the optimum ratio produces the greatest temperature change. In this experiment the total numbers of moles of reactants are kept constant while varying each reactant. The measurements are made on each different ratio until the optimum ratio, the stoichiometric ratio in the equation, is made which consumes the greatest amount of reactants, produces the greatest amount of product and produces the greatest amount of heat.

Kit Includes: 3 X 250mL 1 X 237g

Sodium hypochlorite 13% concentrate Sodium thiosulfate, anhydrous

DOT Info: UN1791, Hypochlorite solutions, 8, III, Ltd Qty

IS8003

\$42.15

Aldon

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Activity Series

An activity series of metals is a table of metals arranged in the order of their decreasing chemical activity or the ease at which the metal will give up one or more electrons to form positive ions. This table is similar to the electrochemical series of elements. For example if you take the group of metals magnesium, mercury and nickel, magnesium is the most reactive and mercury the least. To empirically determine which of these metals is more reactive, place a piece of the metal in a salt solution of the other. The more reactive metal will replace the less reactive metal and the less reactive will appear in the solid form. The reactive metal has been oxidized; the less reactive metal has been reduced.

Kit Includes: 45 pieces . 45 pieces 45 pieces 45 pieces 50ml 50mL 50mL 50mL 2.15g 0.6g 25mL 2 x 75mL 1

Copper Metal Zinc Metal Magnesium Metal Lead Metal Copper Nitrate, 0.1M Zinc Nitrate, 0.1M Magnesium Nitrate, 0.1M Lead Nitrate, 0.1M Potassium Bromide Potassium Bromate Sulfuric Acid, 1.0M Chlorine Water Iodine Water Sodium Bromide, 0.1M Potassium Iodide, 0.1M Sodium Chloride, 0.1M Mineral Oil Empty Bottle (for bromine water)



DOT Info:

Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173.4 for domestic highway or rail transport only





800-724-9877

IS8002

\$109.00

29

Le Chatelier's Principle

Le Chatelier's Principle is a qualitative rule, which allows the prediction of the effect of temperature, pressure and concentration changes on chemical reactions. The principle states: A chemical system at equilibrium when stressed by external forces will adjust in such a way as to minimize that stress. For example when a system is subjected to increased pressure it adjusts so that it will occupy less volume. This offsets the pressure increase. If ice is placed under an increased pressure, it melts because the water obtained from a given mass of ice occupies less volume. In the formation of ammonia (the Haber process) from hydrogen and nitrogen, the product of the reaction (NH₃) occupies less volume than the two uncombined gases. The increase in pressure favors the production of ammonia.

This experiment is divided into three separate reactions demonstrating how different types of stress effect equilibrium. Students are asked to predict the outcome of each situation and then prove or disprove their predictions.

Kit Includes: 1 X 25mL 1 X 10g 1 X 25mL 1 X 200mL 1 X 10g 1 X 25mL

2 X 25mL

1 X 50g

Bromothymol blue pH indicator solution 0.04% Potassium thiocyanate crystal Ferric nitrate solution 0.2M Potassium thiocyanate solution 0.002M Sodium phosphate, dibasic Concentrated hydrochloric acid, 36% Sodium hydroxide solution 0.1N Sodium chloride

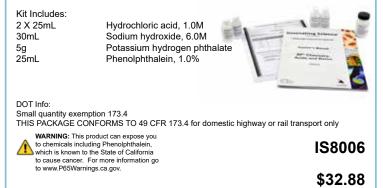
DOT Info: Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173.4 for domestic highway or rail transport only



IS8004 \$45.23

Acids and Bases

A solution of sodium hydroxide which is approximately 0.01M will be standardized with a solid acid, potassium hydrogen phthalate, to determine the exact molarity of the base. The standardized base will then be titrated against a monoprotic unknown acid to determine the molarity of the acid. A pH indicator and/or a pH meter will be used to determine the equivalence point.



Ester Formation

The yield of ester can be increased either by removing one of the products of the reaction as it is formed or by increasing the concentration of one of the reactants. The normal procedure is to remove water using the dehydrating agent sulfuric acid, which also acts as the catalyst. In this experiment, we will combine various alcohols with acetic and butyric acids to form esters. The esters can be identified by their distinctive odors.



Reaction Kinetics

In a chemical reaction, reactants are converted to product at a given rate. This rate can be changed by altering either the temperature of the reaction, the form of the reactants, the concentration of reactants or products or by adding a catalyst. The time it takes for a reaction to occur is monitored by observing a color change. By varying concentration of reactants the rate law is determined. By varying the temperature the activation energy is determined. The data should be graphed and analyzed during the lab so additional measurements can be made if necessary.

Kit Includes: 2 x 100mL 2 x 100mL 2 x 100mL 5 capsules 2 x 100mL Pkg. of 150

Potassium lodide, 0.1% Solution (0.006M) Sodium Thiosulfate, 0.001M Solution Hydrogen Peroxide, 3% Solution (0.88M) Starch Soluble (5 caps per bottle) Hydrochloric Acid, 0.1M Solution UN1789 Pipettes

DOT Info: Non-regulated

IS8007 \$52.00



Dissociation Constants of Weak Acids

Students determine the equilibrium constant of three unknown acids and compare the data to known values to identify them.



Beer Lambert Law

The probability that a photon of light will be absorbed by a solution is directly proportional to the concentration of the absorbing molecules, the thickness of the sample and the degree to which the molecules absorbs light. This relationship is called the Beer-Lambert Law and is used in analytical chemistry to determine concentrations of various compounds spectrophotometrically. A series of 5 solutions with different concentrations of iron are prepared and analyzed using the spectrophotometric assay for iron.

Kit Includes: 25mL 5g 50mL 100mL

Ferric chloride, 0.1M UN2582 Ascorbic acid 1,10 Phenanthroline Acetate buffer

DOT Info: Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173.4 for domestic highway or rail transport only



IS8010

\$39.59

Qualitative Analysis

Colorimetric or turbidimetric methods of analysis have been developed for most of the elements and ionic species. These methods can be employed to quickly determine the chemical composition of an unknown substance. Before attempting to analyze an unknown it is useful to develop a logical process to follow. Care must be taken in sampling, collecting and washing precipitates, performing confirmatory tests and most importantly, recording data. It is also good laboratory practice to always analyze a known with each unknown as a double check on the reagents being used. The unknowns will be analyzed for one of six possible anions: Cl⁻, Br, l⁻, PO₄²⁻, NO³⁻, and SO₄²⁻. A standard solution for each anion will also be run.

DOT Info: UN2031, Nitric acid, 8, II UN2672, Ammonia solutions, 8, III, Ltd Qty UN1830, Sulfuric Acid solution, 8, II, Ltd Qty Single shipper packaging surcharge applies UPS HAZARD CHARGE APPLIES



\$90.46

WARNING: This product car

expose you to chemicals including

Strong inorganic acid mists containing sulfuric acid, which are known to the State of California to

cause cancer. For more information

go to www.P65Warnings.ca.gov.

Stoichiometry in the Synthesis of an Ionic Compound

Many combinations of mono and trivalent cations yield crystals of the same stoichiometry and structure. The crystals are normally in the form of an octahedron. Sodium, potassium and ammonium ions are often the monovalent species whereas aluminum and chromium are examples of the trivalent ions. Potassium alum, $KAl(SO_4)_2 \cdot 12H_2O$, is the most common and is used in water purification, paper manufacturing and as a mordant in dyeing. In this experiment we will perform a series of chemical reactions which lead to the synthesis of the ionic compound potassium alum.

Kit Includes: 2 x 25g 2 x 250mL 4 x 250mL 4 x 250mL

4 x 250mL Ethanol/Water V/V 50/50% Solution DOT Info: UN1814, Potassium hydroxide, solution, 8, II, Ltd Qty UN2796, Sulfuric acid, 8, II, Ltd Qty

Aluminum Metal Powder

Potassium Hydroxide, 3.0M Solution Sulfuric Acid, 3M Solution

\$49.34



WARNING: This product can expose you to chemicals including Strong inorganic acid mists containing suffiric acid and Methanol/Methyl isobutyl ketone, which are known to the State of California to cause cancer, reproductive harm. For more information go to www.P65Warnings.ca.gov.



IS8011

UN1170, Ethanol, 3, III, Ltd Qty

Equilibrium Constant of an Ionic Compound

The idea of the reversibility of chemical reactions was first stated by C. Berthollet in 1799. He noted deposits of sodium carbonate in certain salt lakes in Egypt and concluded that they were produced by the high concentration of sodium chloride and dissolved calcium carbonate. This is the reverse of the standard laboratory procedure in which calcium carbonate is produced from the reaction on sodium carbonate and calcium chloride. A standard curve will be established for the absorbance of a colored species and then used to determine the concentrations of unknown solutions. The results will allow us to determine the equilibrium constant for the formation of the colored species.

Kit Includes: Iron (III) nitrate, 0.2M 500mL 2 X 100mL Potassium thiocyanate, 0.002M 100mL Nitric acid, 0.5M UN2031



DOT Info: UN3264, Corrosive liquid, acidic, Inorganic, n.o.s., (Nitric acid), 8,III,Ltd Qty

IS8012

\$31.00

Freezing Point Depression

Molecular mass is a parameter, which is useful in determining the identity of an unknown compound. One technique to determine the molecular mass of an unknown is to measure the effect the compound has on the freezing point of a solvent in which the unknown is dissolved. The freezing point of a solution is a colligative property. That is, it is a property which varies based on the number of particles (solute) dissolved in the solvent and not on the chemical makeup of the particles themselves. Other colligative properties, which also can be used to determine molecular mass, are osmotic pressure, vapor pressure and boiling point. The nonpolar solvent 2,6-di-tert-butyl-4-methylphenol has a freezing point of approximately 70°C. A quantity of para-dichlorobenzene will be dissolved in the solvent and the effect on the freezing point determined. The freezing point depression constant will be calculated for the solvent. The experiment will be repeated with each of two unknowns and the molecular weight of the unknowns will be determined from the freezing point depression.

Kit Includes: 2 x 25g 25g 25g 25g

p-Dichlorobenzene Crystals Stearic Acid Flakes Naphthalene Flakes UN1334 DOT Info:

Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173.4 for domestic highway or rail transport only

WARNING: This product can expose you to chemicals including p-Dichlorobenzene and Naphtalene, which are known to the State of California to cause cancer. For more information go to www.P65Warnings.ca.gov.



Oxidation-Reduction Reactions

The term oxidation can mean the chemical combination of a substance with oxygen and reduction can be the removal of oxygen from a compound. When oxygen reacts with any other element (except fluorine) it acquires electrons from that element. The element that donated the electrons is said to be reduced. Three experiments will be run where a compound, which is colorless in solution when reduced, is converted to a deeply colored solution when oxidized. The complete balanced reactions for each step should be written showing the transfer of electrons during oxidation and reduction.

Kit Includes: 10g 100mL 2 X 100mL 25mL 25mL 3 X 100mL 25ml 3 X 250mL

Ferrous ammonium sulfate Sulfuric acid, 6.0M Potassium thiocyanate 1.0M Potassium permanganate Hydrogen peroxide, 3% Stannous chloride 0.1M Methylene blue 1% Potassium hydroxide 1.0M Dextrose

DOT Info:

IS8013

100g

UN1814, Potassium hydroxide, solution, 8, II, Ltd Qty UN2796, Sulfuric acid, 8, II, Ltd Qty

WARNING: This product can expose you to chemicals including Strong inorganic acid mists containing sulfuric acid, which are known to the State of California to cause cancer. For more information go to www.P65Warnings.ca.gov



Vapor Pressure

\$46.99

Raoult's Law can be proven empirically by measuring the change in vapor pressure as solute is added to a solvent. The measurements will be taken at different temperatures to demonstrate the effect of temperature on vapor pressure.

Kit Includes 2 x 2.5kg Sodium Chloride Crystals

DOT Info: Non-Regulated



IS8015

\$31.00

Aldon

www.aldon-chem.com

Di-Tert-Butyl-4-Methylphenol Crystals

Electrochemical Cells

The tendency of oxidation-reduction reactions is to proceed to an equilibrium state. These reactions occurring in electrochemical cells provide another way for us to express the driving force in chemical reactions. When reagents that accept or donate electrons are arranged so that the electrons can enter or leave the reaction through a metallic conductor, an electrochemical cell is established. A half-cell contains a metal in contact with a solution of its salt. Each metal will develop a different electrical potential based on its electron configuration. The standard reduction potential listed in various references is the voltage that a half-cell develops when combined with a hydrogen half-cell. First, construct a simple chemical battery and determine from the standard reduction potentials what the output of the battery will be (if a voltmeter is available the actual and theoretical voltages can be compared). Second, construct an electrolysis cell and demonstrate how hydrogen and oxygen can be produced from the electrolysis of water.

Kit Includes:	
2 x 500mL	Cupric Sulfate, 0.5M Solution
25mL	Bromothymol Blue, 0.04% Solution
8 x 500mL	Sodium Sulfate, 1M Solution
Pkg. of 15	Copper Metal Strips
Pkg. of 15	Magnesium Metal Strips 5"UN1869
Pkg. of 15	Dialysis Tubing Strips 6"

WARNING: This product can expose you to chemicals including Strong inorganic acid mists containing sulfuric acid, which are known to the acid mists State of California to cause cancer. For more information go to www.P65Warnings.ca.gov



WARNING: This product can expose you to chemicals including Methanol, which is

known to the State of California to cause reproductive harm. For more information go

to www.P65Warnings.ca.gov

WARNING: This product can expose you

compounds, which are known to the State of

information go to www.P65Warnings.ca.gov.

California to cause reproductive harm. For more

to chemicals including Mercury/mercury

\$159.75

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IS8016

DOT Info

\$57.00

Thin Layer Chromatography

Chromatography is an analytical tool used to separate similar compounds. Thin-layer chromatography utilizes a sheet coated with silica gel. The samples are applied to the silica gel sheet and separated by migrating an appropriate solvent up the sheet. This type of separation is called adsorption chromatography because the separation of similar compounds is based on the selective adsorption of the compounds on the silica gel solid phase. In this lab samples of various plant leaf materials may be extracted or the red and green leaf extracts included may be used. The samples of extracted leaf pigments are applied to a silica gel sheet and separated using a mixture of organic solvents.

Kit Includes:
2 x 10g
2 x 25mL
Pkg. of 15
6 x 250mL
2 x 50mL

Alfalfa Powder Red Extract Chromatography TLC Sheets Chromatographic Solvent Isopropyl Alcohol, 70%

DOT Info: UN1230, Methanol, 3, PG II, Ltd Qtv UN1219, Isopropanol, 3, PG II, Ltd Qty



IS8017 Aldon \$89.00

WARNING: This product can

expose you to chemicals

www.aldon-chem.com

\$44.50

Thermochromism

2 x 25mL Methyl Alcohol UN1230

for domestic highway or rail transport only

Small quantity exemption 173.4

Isopropyl Alcohol UN1219

THIS PACKAGE CONFORMS TO 49 CER 173 4

Kit Includes:

2 x 25mL

DOT Info:

IS8018

The way in which light is absorbed or reflected by a crystalline material is determined by the structure of the crystal. This crystal structure can be changed by the application of heat. In some cases this change is reversible and the original structure, and therefore the original color, returns upon cooling. Two compounds are formed when metal salts are reacted with the [Hgl_]² Ion. Each compound is heated on a hot plate and a change in color is noted. The initial color of the compound returns upon cooling.

Determination of the Molecular Mass of a Liquid

stopper with a hole in it. The test tube is placed in a hot water bath. The

liquid vaporizes and excess vapors escape through the hole. The tube

is then placed in an ice bath to guickly cool the vapor and cause it to

condense. The mass of the tube is determined.

A volatile liquid is placed in a test tube and the tube is closed with a

3 x 25g 🛛 🖡	es: Mercury (II) Chloride UN1624 Potassium lodide Silver Nitrate UN1493
15g 2 x 25g	Copper (II) Chloride, Anhydrous UN2802
DOT Info:	

Small quantity exemption 173 4 THIS PACKAGE CONFORMS TO 49 CFR 173.4 for domestic highway or rail transport only

IS8019

Determination of the Hardness of Water

In this experiment the indicator eriochrome black T (EBT) is used to signal the presence of ions in the water sample. EBT binds with free metal ions in the water to form a pink complex. EDTA has a stronger affinity for the metal ions than EBT so when EDTA is added it replaces the EBT and the EBT returns to its blue, uncomplexed color. The blue color is used as the end point in the titration. A sample of tap water is treated with EBT indicator. If the indicator turns from blue to pink, metal ions such as calcium and magnesium are present. To determine the concentration of ions present, the sample is titrated with a known molar concentration of EDTA.

Kit Includes: 4 x 500mL EDTA Solution, 0.005M 2 x 200ml Buffer Solution 2 x 15mL EBT Solution (Eriochrome Black, 0.1%) UN1219

DOT Info⁻ Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173.4 for domestic highway or rail transport only



IS8020

\$31.00

Synthesis of Aspirin

In a reaction vessel salicylic acid, acetic anhydride and sulfuric acid are mixed. The exothermic reaction will cause the temperature to increase to 70-80°C. Once the reaction is complete the vessel is cooled in an ice bath and the acetylsalicylic acid crystallizes out. This material is re-crystallized in toluene to purify the product. The product identity is confirmed by melting point. The mass of the re-crystallized product is used to determine the yield of the synthesis. In this experiment, we will perform the synthesis of acetylsalicylic acid (aspirin), purification by re-crystallization and use melting point to confirm identity.

Kit Includes: 2 x 25mL Acetic Anhydride UN1715 Salicvlic Acid 15g 2 x 25mL Sulfuric Acid Concentrate UN1830 2 x 25mL Toluene UN1294

DOT Info: Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173.4 for domestic highway or rail transport only



WARNING: This product can expose you to chemicals including Strong inorganic acid mists containing sulfuric acid and Toluene, which are known to the State of California to cause cancer reproductive harm. For more information go to www.P65Warnings.ca.gov.

Grignard Synthesis of Benzoic Acid

Once the Grignard reagent is prepared, it can be used to prepare the target compound. In this experiment the target compound is a carboxylic acid. To prepare a carboxylic acid, the Grignard reagent is carboxylated in a reaction with dry ice. After the reaction is complete, hot water is added to evaporate any remaining solvent. Acid is added to dissolve the magnesium salts and liberate the carboxylic acid. Once the solution is cooled, the acid can be collected. In this experiment, we will prepare a carboxylic acid (benzoic acid) by the Grignard method.



IS8021

\$49.50



158001	I hermochemistry and Hess's Law		100012	
IS8002	Activity Series	WARNING: This product can expose you to chemicals including Lead and lead compounds, Potassium Bromate, Methanol/Methyl isobutyl ketone, Phenolophthalein, Strong inorganic acid mists containing sulfuric acid, Naphthalene, P-Dichlorobenzene, Toluene and Mercury/mercury compounds, which are known to the State of California to cause cancer and reproductive harm. For more information go to www.P65Warnings.ca.gov.	IS8013	Oxidation-Reduction Reactions
IS8003	Mole Ratio of Reactants		IS8014	Freezing Point Depression
IS8004	Le Chatelier's Principle		IS8015	Vapor Pressure
IS8005	Ester Formation		IS8016	Electrochemical Cells
IS8006	Acids and Bases		IS8017	Thin Layer Chromatography
IS8007	Reaction Kinetics		IS8018	Determination of the Molecular Mass of a Liquid
IS8008	Dissociation Constant of Weak Acids	Acide		Thermochromism
IS8009	Qualitative Analysis	5	IS8020	Determination of Water Hardness
	, ,		IS8021	Synthesis of Aspirin
IS8010	Beer Lambert Law		IS8022	Grignard Synthesis of Benzoic Acid
IS8011	Stoichiometry in the Synthesis of an	Ionic Compound	100022	Chighard Cynthesis of Denzolo / told

IS8011 Stoichiometry in the Synthesis of an Ionic Compound

www.aldon-chem.com



Determination of the Properties of Buffer Solutions

In this experiment, you will prepare three buffer solutions having different pH values and show that the pH of these solutions does not change significantly when small amounts of acids and bases are added. You will also show that when the same amounts of acids and bases are added to water and to a non-buffer solution (e.g. NaCl solution), the pH changes are large.

Kit Includes:

Kit includes.	
30g	Sodium Acetate, trihydrate
2 x 25g	Ammonium Chloride
30g	Sodium Bicarbonate
4 x 25g	Sodium Chloride
23mL	Acetic Acid EZ-Prep, to make 800mL of 0.5M Solution
27mL	Ammonium Hydroxide, EZ-Prep, to make 800mL of 0.5M Solution
15mL	Sodium Hydroxide, 1.0 N Solution UN1824
15mL	Hydrochloric Acid, 1.0 N Solution UN1789

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IS8024

\$69.90

The Molar Volume of a Gas

Students will determine the volume of one mole of gas. This is accomplished by generating a known mass of oxygen gas, measuring its temperature, volume and pressure, and then using the data to calculate the molar volume at STP. The oxygen is generated by the decomposition of potassium chlorate at high temperature.

Kit Includes: 2 x 25g Potassium Chlorate UN1485 2 x 25g Manganese Dioxide, 85% Native Powder UN1479 DOT Info: Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173.4 for domestic highway or rail transport only



Determination of the Empirical Formula of Magnesium Oxide

The quantitative stoichiometric relationships governing mass and amount will be studied using the combustion reaction of magnesium metal. Magnesium is reacted with oxygen from the air in a crucible, and the mass before and after the oxidation is measured. The resulting masses are used to calculate the experimental empirical formula of magnesium oxide, which is then compared to the theoretical empirical formula. A crucible and Bunsen burner will be used to heat magnesium metal to burning. This lab illustrates the (i) law of conservation of mass and (ii) the law of constant composition.

Kit Includes: 1 Roll Magnesium Metal, Ribbon UN1869

DOT Info:

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IS8025

\$29.31

Acid Base Indicators

pH is the measure of the concentration of hydrogen ions in a solution. As this concentration can extend over several orders of magnitude, it is convenient to express it by means of logarithms of base ten. Certain pigments change color with a change in pH. In this experiment we will extract pigments from various sources and determine if they are sensitive to changes in pH.

Kit Includes:

25mLHydrochloric Acid, 32% UN178910gSodium Hydroxide UN18232 x 30mLIsopropyl Alcohol, 99% UN121930mLAcetone UN1090

DOT Info:

Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173.4 for domestic highway or rail transport only



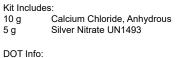
IS8027

\$55.01

Aldon

Gravimetric Analysis

Determine the identity of a metal carbonate by gravimetric analysis. Gravimetric analysis is a technique through which the amount of an analyte (the ion being analyzed) can be determined through the measurement of mass. You will complete the analyses comparing the masses of two compounds containing the analyte. The formula weight and the identity of the unknown are determined using gravimetric analysis.



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The Electrochemical Series

Students will study electrochemical series. The electrochemical series is built up by arranging various redox equilibria in order of their standard electrode potentials (redox potentials). When a strip of metal (an electrode) is placed in water the metal has a tendency to go into solution as ions, with a simultaneous build up of electrons on the metal strip. This process produces an electrical potential difference between the metal and solution which is called an electrode potential (E°).

Kit Includes: DOT Info: Copper Metal Strips 15 Small quantity exemption 173.4 15 Zinc Metal Strips THIS PACKAGE CONFORMS TO 15 Magnesium Metal Strips 49 CFR 173.4 for domestic highway 15 Iron Metal Strips or rail transport only 15 Aluminum Metal Strips 75 Filter Paper Strips EZ-Prep to make 1000mL of 0.1M Solution of: 1 ea Copper Sulfate Zinc Sulfate Magnesium Sulfate Iron Sulfate Aluminum Sulfate

Analysis of Potassium Aluminum Sulfate 12 Hydrate

After a compound has been synthesized, analytical tests should be carried out to confirm that the correct compound has been produced. In this lab we will perform tests to confirm that the crystals formed are in fact potassium aluminum sulfate 12 hydrate by comparing the melting point to published results, finding the number of waters of hydration, and determining the percent of sulfate in the compound.

Kit Includes:

IS8029

2 x 100g Aluminum Potassium Sulfate 4 x 25g Barium Nitrate UN1446

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\$28.00

Liquid (Column) Chromatography

The mixture to be analyzed by column chromatography is applied to the top of the column. The liquid solvent (the eluent) is passed through the column by gravity or by the application of air pressure. Equilibrium is established between the solute adsorbed on the adsorbent and the eluting solvent flowing down through the column. Because the different components in the mixture have different interactions with the stationary and mobile phases, they will be carried along with the mobile phase to varying degrees and a separation will be achieved. The individual components, or elutants, are collected as the solvent drips from the bottom of the column.

Kit Includes: 4X25mL Isopropyl Alcohol UN1219 100g Silica gel 60 2X15mL Dye Mixture 15 Pipettes Glass wool

WARNING: This product can expose you to chemicals including Glass wool fibers (inhalable and biopersistent), which are known to the State of California to cause cancer. For more information go to www.P65Warnings.ca.gov.





IS8030 \$97.66

IS8031

800-724-9877

\$76.50 **Aldon**

All Chemistry Demonstration kits come with enough reagents to complete at least 5 chemistry demonstrations. Complete with instructions.

Synthesis of Rayon

Demonstrate polymer chemistry. Newsprint is dissolved in a solution of ammonium hydroxide and cupric sulfate. The solution is then added to a sulfuric acid solution and dark strands are formed.

Ammonium Hydroxide, 28-30% Solution UN2672

Kit Includes:

125mL 175mL

200mL Sulfuric Acid, 0.5M/1.0N Solution UN2796

Copper (II) Sulfate, 25% Solution

UN2672, Ammonia solutions, 8, III, Ltd Qty UN2796, Sulfuric acid, 8, II, Ltd Qty

WARNING: This product can expose you to chemicals including Strong inorganic acid mists containing sulfuric acid, which are known to the State of California to cause cancer. For more information go to www.P65Warnings.ca.gov.

IS7001

\$30.82

Ampholytes Chemical Demo

Ampholytes are compounds which react with both acids and bases. When zinc chloride is reacted with sodium hydroxide the soluble zincate ion is formed. When zinc hydroxide is reacted with hydrochloric acid, the soluble zinc chloride and water are formed.



Endothermic Reactions

Study the process of endothermic reactions. The reaction is safely carried out in a plastic bag. The students can pass the bag around to feel the result of the reaction.

Kit includes: 5 bags x 25g Ammonium Nitrate UN1942 + 1 Tbsp Vermiculite

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IS7005

Aldon

\$27.76

Autocatalysis

Catalysis is a basic principle of chemistry and biochemistry. Some reactions are autocatylic; the product of the reaction actually catalyzes further reactions. The reaction is initiated with acid and produces acid to continue the reaction.



IS7002 \$29.31

Exothermic Reactions

Study the process of exothermic reactions. The reaction is safely carried out in a plastic bag. The students can pass the bag around to feel the result of the reaction.

Kit Includes: 5 bags of the following mixture: 25g Iron Metal Powder, 1g Sodium Chloride Crystals, 5g Calcium Chloride Flakes, 1 Tbsp Vermiculite





Exothermic/Endothermic Combination

\$46.24

Study both processes of endothermic and exothermic reactions. Reactions are safely carried out in plastic bags. The students can pass the bags around to feel the results of the reactions.

Combo kit includes both IS7004 and IS7005 kits.

DOT Info: Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173.4 for domestic highway or rail transport only



IS7006



Discover the principle of oxidation-reduction. Compounds that change color as a result of a change in their pH or oxidation state are called indicators. Illustrate this principle by mixing potassium hydroxide and dextrose with methylene blue and introducing oxygen by shaking the flask.

Kit Includes: 2 x 25mL 6mL 5 bags x 5g

Potassium Hydroxide, 5.0M Solution UN1814 Methylene Blue Chloride, 1% Solution Dextrose Monohydrate (Glucose) Powder

DOT Info: Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173.4 for domestic highway or rail transport only

IS7007

\$37.53

Thin Layer Chromatography

TLC is a technique of separation chemistry, which provides a simple, rapid method of separating small amounts of compounds. A mixture of three dyes is prepared and then separated on a thin layer of the absorbent silica gel with the appropriate solvent.



Synthetic Rubber

Show how the first synthetic rubber was made in United States by a reaction of sodium polysulfide with ethylene chloride. This produces a simple condensation polymer consisting of repeating units of ethane and polysulfide.

Kit Includes:

4 x 25mLSodium Polysulfide Solution UN176025mLEthylene Dichloride UN1184

DOT Info:



Nylon 6-10 Rope Trick

Show an example of polymer chemistry. Make your own nylon which is a synthetic compound of high molecular weight that consists of up to millions of repeated chemical units that are linked together. This polymer is referred to as Nylon 6-10.

Kit Includes:

25mL Solution A (1,6-Hexanediamine, Sodium Hydroxide, Water) UN1760 25mL Solution B (Sebacoyl Chloride, Hexane) UN1208

DOT Info: Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173.4 for domestic highway or rail transport only



chemicals including n-Hexane, which is known to the State of California to cause reproductive harm. For more information go to www. P65Warnings.ca.gov.

IS7010

\$29.99

Aldon

Formation of Silicate (Red & White Precipitates)

Show how silicates of divalent ions are insoluble in water whereas silicates of monovalent ions are soluble.

Kit Includes: 50mL 50mL 200ml

Calcium Chloride, 1.0 Molar Solution Calcium Chloride, 1.0 Molar Solution with Phenolphthalein Sodium Silicate, 15% Aqueous Solution

DOT Info: Non-Regulated

IS7011





Surface Tension of Water

Explain the principle of high surface tension as water acts like a membrane stretched across the surface preventing the sulfur particles from sinking. A wetting agent lowers the surface tension and allows the particles to fall through the surface to the bottom of the beaker.

Kit Includes: 1 x 15 g Sulfur NA1350 1 x 5 mL Dish Detergent, Green



DOT Info: Non-Regulated

IS7012

\$20.54

Eutectic Alloys

Form a metal alloy as bismuth and tin are heated to the eutectic point of the two metals which is lower than the melting point of either metal.





IS7014

Chemistry Demonstration Kits

Chemiluminescence

\$24.17

Explain and view chemiluminescence. Observe how luminol is converted to an excited state in the presence of an oxidizer such as hydrogen peroxide. As the excited state molecule decays or returns to the ground state, energy is produced as light.

Kit Includes: 100mL 50mL

Luminol Solution Hydrogen Peroxide, 6% Solution

DOT Info: Non-Regulated



IS7013



www.aldon-chem.com

\$34.93

Chemical Battery

Construct a simple wet cell battery. When complete it will generate 1.5 volts for 20-30 minutes. Several cells can be connected in series to increase the voltage or in parallel to increase the current.

Kit Includes: 100mL 200mL 2 x 6" long 2 x 3/4"x5" 2 x 15.9mm 25mL 1 each 1 each

Copper (II) Sulfate, 0.5M Solution Sodium Sulfate, 0.5M Solution Magnesium Metal Strips UN1869 Copper Metal Strips Dialysis Tubing Hydrochloric Acid, 3% Solution UN1789 Light Bulb & Wire Assembly Empty Jar with Lid



IS7015

\$31.50

DOT Info: Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173.4 for domestic highway or rail transport only

WARNING: This product can expose you to chemicals including Strong inorganic acid mists containing sulfuric acid, which are known to the State of California to cause cancer. For more information go to www.P65Warnings.ca.gov.

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Polyurethane Foam

Make your own foam and explain how when two viscous liquids are mixed together, they initiate a reaction producing a light weight polyurethane foam which expands to 30 times the original volume of the two liquids.



IS7018 \$73.25

Negative Coefficient of Solubility

Explain how there are exceptions to the rules of chemistry. Heat a solution and watch it form a precipitate. Watch it go back into solution as it cools.

Kit Includes: 500mL Calcium Acetate, Saturated Solution

DOT Info: Non-Regulated



Foam City - The Catalytic Decomposition of Hydrogen Peroxide

A large quantity of foam is produced when detergent and potassium iodide are added to hydrogen peroxide.

Kit Includes:

3 x 30mLHydrogen Peroxide, 30% UN201430mLDish Detergent, Green10gPotassium Iodide, Crystals

DOT Info:

Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173.4 for domestic highway or rail transport only



IS7019

\$26.71

The Dehydration of Sucrose

Sulfuric acid is poured on sucrose. The resultant reaction produces a large amount of black foam as well as steam and smoke.



IS7020

\$37.01

Oscillating Reactions - Yellow to Blue and Back

Prepare a flask containing a yellow solution, as you mix it on a magnetic stirrer the solution will turn from yellow to blue and back to yellow.



Oscillating Reactions - The Traffic Light

Sodium Hydroxide, Beads UN1823 Indigo Carmine, makes 50mL

A flask containing a yellow solution is gently swirled and the solution turns red. When the flask is shaken the solution turns green.

Kit Includes: 10g 0.5g 5 x 5g

DOT Info:

Dextrose (Glucose)

Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173.4 for domestic highway or rail transport only

IS7024

\$24.17



www.aldon-chem.com

The Volcano -Dehydration of Sucrose

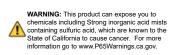
Create your own volcano for a science fair demonstration.

Kit Includes:

75g	Sucrose Crystals
25mL	Sulfuric Acid, Concentrate UN1830
50g	Sodium Carbonate, Anhydrous

DOT Info: Small quantity ex

Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173.4 for domestic highway or rail transport only







IS7021

\$36.50

Check out the Dehydration of Sucrose video at www.aldon-chem.com



Solid Fuel

Two solutions are poured together in a beaker and form a gel. The gel can be ignited with a match. Commercially this gel is sold as Sterno[®].

Kit Includes: 2 x 25mL Ethyl Alcohol, Denatured UN1170 2 x 25mL Calcium Acetate Solution

DOT Info: Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173.4 for domestic highway or rail transport only

IS7025



WARNING: This product can expose you to chemicals including Methanol, which is known to the State of California to cause reproductive harm. For more information go to www.P65Warnings.ca.gov



The Oxidation-Reduction Flag (Redox Flag)

You will be able to demonstrate the principles of a redox reaction. Paint the potassium thiocyanate solution and potassium ferrocyanide solution onto a filter paper. Once they are dry you can spray the filter paper with iron (II) chloride and watch blue and red colors appear.

Kit Includes: 3 x 25mL Iron (III) Chloride 0.1M Solution UN2582 30ml Potassium Thiocyanate 0.1M Solution

30mL Potassium Ferrocyanide 0.1M Solution

DOT Info: Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173.4 for domestic highway or rail transport only



IS7026

\$31.85

Oscillating Reactions - The REDOX Rainbow

For this demonstration, add hydrogen peroxide to a colorless solution of potassium iodide to produce a red color, or you can add the hydrogen peroxide to a purple solution of potassium permanganate to create a colorless solution.

Kit Includes: 100mL 200mL 200mL 10mL

Hydrogen Peroxide, 3% Solution Potassium Iodide, 0.1M Solution Potassium Permanganate, 0.02M Solution Sulfuric Acid, Concentrated UN1830

DOT Info: Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173.4 for domestic highway or rail transport only

WARNING: This product can expose you to chemicals including Strong inorganic acid mists containing sulfuric acid, which are known to the State of California to cause cancer. For more information go to www.P65Warnings.ca.gov.





\$26.71



Chemistry Demonstration Kits

Innovating Science[®]

Oxidation of Glycerin

A small amount of potassium permanganate is placed in a petri dish. A few drops of glycerin are added and after a few seconds, a puff of smoke and violent flames are produced.

Kit Includes: 30g Potassium Permanganate UN1490 25mL Glycerin

DOT Info: Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173.4 for domestic highway or rail transport only





IS7028

\$22.35

Underwater Fireworks - The Reaction of Acetylene and Chlorine

Calcium carbide is placed in a cylinder containing water. A reaction occurs producing acetylene gas bubbles. A plastic tube from a chlorine gas generator is placed in the cylinder. As the bubbles of acetylene and chlorine collide flashes of light are produced.

Kit Includes: 250mL 30mL 10g

Sodium Hypochlorite Hydrochloric Acid, 1.0M Solution UN1789 Calcium Carbide, Lumps UN1402





DOT Info: Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173.4 for domestic highway or rail transport only

IS7029

\$25.70

Oxidation-Reduction of Complex Ions

Hydrogen Peroxide, 3% Solution

Copper (II) Sulfate, 1.0M Solution

Potassium Sodium Tartrate, 0.3M Solution

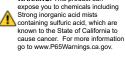
Hydrogen peroxide is added to a solution of sodium potassium tartrate and heated. Copper sulfate solution is added and the solution turns light blue. With continued heating the solution foams and turns orange-gold.

Kit Includes: 250mL 250mL 10ml

DOT Info: Non-Regulated







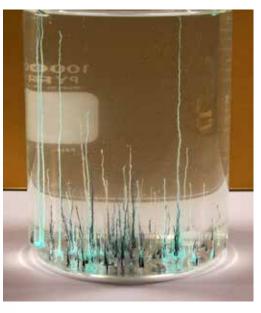
IS7030 \$29.31



The Silicate Garden

A few colored crystals are added to a clear solution in a glass jar or beaker. Within a few minutes large plant like structures extend from the crystals.

Kit Includes: 2 x 30mL Sodium Silicate Solution, 40% 20g Iron (III) Chloride, Hexahydrate UN1759 20g Copper (II) Chloride, Dihydrate UN2802 20g Zobalt Nitrate, Hexahydrate UN3085 20g Zinc Sulfate, Heptahydrate DOT Info: Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173.4 for domestic highway or rail transport only IS7031



\$43.18

Formation of Tin Wool

This is a single replacement reaction in which the zinc metal dissolves in solution and tin metal is formed.

Kit Includes: 2 x 700mL 140g

Tin (II) Chloride, 1.0 Solution UN1789 Zinc Metal Mossy

DOT Info: UN1760, Corrosive liquid, n.o.s., (Stannous Chloride, Hydrochloric Acid), 8, PGIII, Ltd. Qty

Up in Vapor! Hydrogen Peroxide Decomposition

In this experiment you will show the decomposition of the chemical hydrogen peroxide using sodium iodide.

Kit Includes: 350mL 7 x 4g

Hydrogen Peroxide, 30% Sodium Iodide

DOT Info: UN2014, Hydrogen peroxide, aqueous solutions, 5.1, (8), II UPS Hazard charge applies



IS7032

44

\$73.25

IS7033

\$36.50

800-724-9877



Le Chatelier's Principle: A Dynamic Demo on the Overhead

Use an overhead projector to study the reaction of Le Chatelier's principle.

Kit Includes:	
140mL	Potassium Thiocyanate, 0.002M Solution
10mL	Iron (III) Nitrate, 0.2M Solution
5g	Potassium Thiocyanate
5g	Sodium Phosphate, Monobasic, Anhydrous

DOT Info: Non-Regulated



Water to Wine

Study acid base indicators and complex ions. Mixing three colored chemical solutions results in another wine colored solution. Adding the fourth chemical of sodium fluoride turns the solution clear.

Kit Includes: 10mL 15mL 15mL 5g 3 x 25mL 10mL 10mL

Potassium Thiocyanate, 1.M Solution Thymolphthalein, 0.04% Solution UN1170 Phenolphthalein, 0.5% Solution UN1987 Iron (III) Chloride, 6-Hydrate UN1759 Sodium Fluoride, 1.0M Solution UN3415 Hydrochloric Acid, 32% UN1789 Ammonium Hydroxide, 14.8M, 28-30% Solution UN2672

DOT Info: Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173.4 for domestic highway or rail transport only





IS7034

\$32.88

IS7035

\$45.49

Chemistry Demonstration Kits

Forming Red, White & Blue

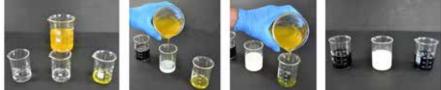
Introduce chemical reaction types. This is a patriotic demonstration that illustrates complex ion formation, double replacement while introducing chemical reaction types.

Kit Includes: 30mL 140mL 140mL 140mL

Iron (III) Chloride, 1.0M Solution UN2582 Silver Nitrate, 0.1M Solution Potassium Ferrocyanide, 0.1M Solution Potassium Thiocyanate, 0.002M Solution

DOT Info: Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173.4 for domestic highway or rail transport only





IS7036

\$35.00

Density Of Liquids: The Color Column

Several liquids of varying color are added to a cylinder. Due to differing densities of the liquids, a density gradient is formed resulting in a layered, multi-colored column. As an optional investigation, the relative density of small solid objects may be compared by dropping them in the density column.

Kit Includes: 450mL Glucose syrup 400mL Liquid detergent 400mL Vegetable oil 400mL Ethyl alcohol Blue food coloring 25ml 25mL Red food coloring Graduated cylinder 5 plastic cups DOT Info⁻ UN1170, Ethanol 3, II Ltd. Qty

> WARNING: This product can expose you to chemicals including Methanol, which is known to the State of California to cause reproductive harm. For more information go to www.P65Warnings.ca.gov



IS7037

\$71.65

Disappearing Rainbow

Three indicators are used in conjunction with acids and bases to create a rainbow of colors. In a basic solution, phenolphthalein is red, p-nitrophenol is yellow, and thymolphthalein is blue. These can be combined in various ways to create all the colors of the rainbow. If done correctly the students only see the acid or base being poured, which adds to the impressive display. There are enough materials to complete the demonstration 7 times.

Kit Includes: 25mL 25mL 25mL 25mL 25mL 25mL 25mL 2 X 30mL	Red Indicator UN1170 Orange Indicator UN1170 Yellow Indicator UN1170 Green Indicator UN1170 Blue Indicator UN1170 Violet Indicator UN1170 Hydrochloric Acid 1.0 N UN1789
2 X 30mL 30mL	Sodium Hydroxide 1.0 N UN1824 Sodium Hydroxide Sol 6.0M (6N) UN1824

DOT Info: Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173.4 for domestic highway or rail transport only

www.aldon-chem.com

WARNING: This product can expose you to chemicals including Phenolphthalein/Methanol/ Methyl isobutyl ketone, which are known to the State of California to cause cancer, reproductive harm. For more information go to www.P65Warnings.ca.gov.





IS7039

\$39.25

Chemistry Demonstration Kits

Innovating Science®

Thionin and Iron: A Light Induced Redox Reaction

In this lab a thionin-iron solution is prepared. Exposing the solution to light energy causes the colored thionin to be reduced by iron (II) ions, resulting in a clear solution. When half of the solution, in a single container, is exposed to light it will turn clear while the half not exposed to light will remain colored.

Kit Includes: 10g Iron(II) sulfate 2x25mL Sulfuric acid solution UN2796 Thionin Quick Solution

DOT Info: Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173.4 for domestic highway or rail transport only

MARNING: This product can expose you to chemicals including Strong inorganic acid mists containing suffuric acid, which are known to the State of California to cause cancer. For more information go to www. P65Warnings.ca.gov.

IS7040

\$28.28

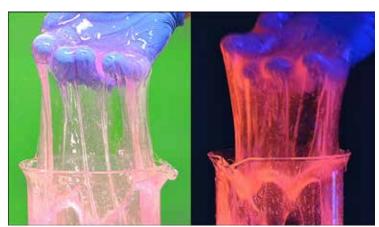
Fluorescent Slime Using Polyvinyl Alcohol

Create your own fluorescent polymers (Slime) in this great classroom demonstration. Mixture is pink under regular light and glows orange under a black light. Included pigment will make this polymer activity easy to see and memorable for your students!

Kit Includes: 100mL 500mL 5mL

Sodium Borate 4% Sol Polyvinyl Alcohol 3% Solution Fluorescent Pink Pigment Mixture

DOT Info: Non-Regulated





Instant Light Powder Chemiluminescence Demo

In this demonstration, students will observe an example of a chemiluminescence reaction. Instant light powder and water are placed in a beaker producing a blue light. This demonstration is designed to further help students understand the topics of chemiluminescence and kinetics. There is enough Instant Light Powder to complete 5 demonstrations.

15%

50

001

150

Kit Includes: 1 btl Instant Light Powder DOT Info: Non-Regulated





IS7042

\$13.35

Aldon

IS7041

\$23.75

Snow Polymer Demonstration

Demonstrate the wonderful, super absorbent properties of a polymer while making snow in your classroom. As the hydrophilic polymer instantly absorbs water, it will expand to over 40 times its original volume. The end result is fluffy flakes of fake snow. Performs 5 demonstrations. Complete with instructions.



War of the Acids

What does it mean to be a strong acid or a weak acid? This

strong and weak acids to demonstrate this concept. A low

demonstration kit uses visually stimulating comparisons between

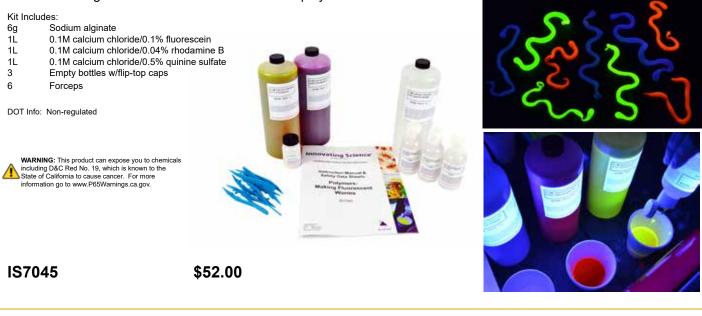
range pH indicator solution allows the audience to see how pH

is different between acids of equal concentration as well as the

impact of adding the conjugate base materials. The violence of

Polymers – Making Fluorescent Worms

Allow students to learn about polymers and cross-linking using this fun and colorful activity. Students add a solution of sodium alginate to three calcium chloride solutions, each containing a different fluorescent dye. After a few minutes the liquid sodium alginate solution forms into gel-like polymer "worms." Upon completion of the activity, the worms may be placed under a black light and the students will observe their worms glow red, blue, and green. Contains enough material to make several dozen polymer worms.



Rainbow Electrolysis

Study oxidation-reduction, electrolysis and acid-base reactions as Universal Indicator is added to an aqueous sodium sulfate solution. When two carbon electrodes are connected to a battery and placed in the solution, a rainbow of colors forms as the water is electrolyzed. There are enough materials to complete the demonstration 5 times.



Gold Nanoparticle Demonstration

Explore the interesting properties of nanomaterials by reducing a solution of gold (III) chloride to form colloidal gold nanoparticles. As the reaction occurs, the mixture undergoes multiple color changes before becoming the bright red color characteristic of small gold nanoparticles. There are enough materials to complete the demonstration 5 times.

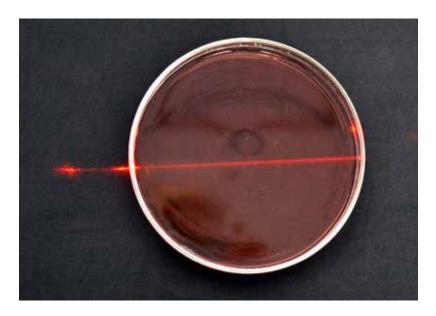
Kit Includes:

100 mL Gold (III) Chloride Solution, 1mM10 mL Sodium Citrate Solution, 1%50 mL Sodium Chloride Solution, 1M

DOT Info: Non-regulated

IS7047

\$41.10



Electrolysis of Tin

The electrolysis of a tin (II) chloride solution demonstrates the oxidation and reduction of Sn²⁺ ions as beautiful tin crystals grow and spread throughout the solution. There are enough materials to complete the demonstration 5 times.

Kit Includes: 2.5g Tin (II) Chloride 10 Zinc Electrodes 5 Petri Dishes 1 Red Connecting Wire 1 Black Connecting Wire

DOT Info: Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173.4 for domestic highway or rail transport only

IS7048

\$23.64



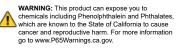
WARNING: This product can expose you to chemicals including Phthalates, which is known to the State of California to cause reproductive harm. For more information go to www.P65Warnings.ca.gov.

Electric Art

A copper "pen" will be used to electrolyze a solution of potassium iodide and draw on a piece of filter paper. The color of the "ink" will change depending on whether the copper is acting as the cathode or the anode in the electrolytic cell. There are enough materials to complete the demonstration 5 times.

Kit Includes:

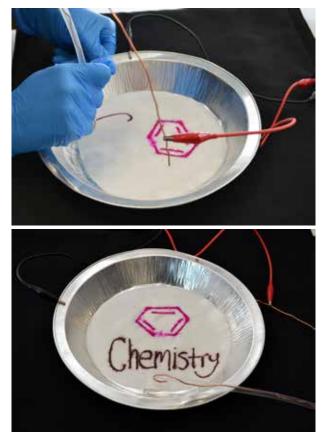
25 mL Phenolphthalein, 1% Solution
8g Potassium Iodide
2.5g Soluble Starch
1 Copper Wire
1 Aluminum Pan
1 Red connecting Wire
1 Black connecting Wire
10 Filter Paper Pieces
5 Plastic Straws



DOT Info: Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173.4 for domestic highway or rail transport only

www.aldon-chem.com

IS7049 \$41.10



800-724-9877



Green Chemistry

Innovating Science®

Green Chemistry: The Production of Biodiesel

In this activity, students will be performing a two-phase process to produce small batches of crude biodiesel. The crude biodiesel produced is of sufficient quality for use in the demonstration of the burning qualities of both biodiesel and vegetable oil. Included is an optional small-scale exercise where the students will use a washing procedure to experience the full process of producing biodiesel to meet quality levels necessary for use in vehicles. Kit contains enough materials for 15 groups of students. Teacher's Manual and Student Study Guide copymasters are included.

Kit Includes: 4 X 500mL 15 X 25ml 20g 2

Vegetable oil Potassium hydroxide

Optional:

Methyl alcohol Microburners

Containers with sealable caps large enough to hold 200mL of liquid (if performing the washing process) Distilled water (if performing the washing process)

DOT Info: Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173.4 for domestic highway or rail transport only

WARNING: This product can expose you to chemicals including Methanol, which is known to the State of California to cause reproductive harm. For more information go to www.P65Warnings.ca.gov.

IS9500

\$79.25

Electrochemical Remediation of Wastewater

Water pollution is one of the largest threats facing the global population. Water is a finite resource. Once polluted, it cannot be set aside in the hopes that the environment will "make" new, clean water. One procedure often employed to treat wastewater is coagulation/flocculation. While the coagulation/flocculation procedure in wastewater treatment is effective, it involves the addition of chemical components to the water being treated. Recently, a great deal of attention has been given to less traditional alternatives to the typical process of coagulation/ flocculation. One such approach receiving a good deal of attention is a process called electrocoagulation. Electrocoagulation is a coagulation process carried out by an electrical charge. Kit contains enough materials for 15 groups of students. Teacher's Manual and Student Study Guide copymasters are included.



Detergents and the Environment

Understand the consequences of cultural nutrient overloading on the aquatic environment. Recognize that there may be alternative, more environmentally-friendly alternatives in typical consumer chemical goods. Demonstrate and monitor the effects of two detergent builders on natural water samples. Visually quantify the differences between phosphate and nonphosphate detergent builders on aquatic organisms. Kit contains enough materials for 15 groups. Teacher's Manual and Student Study Guide copymasters are included.

Kit Includes:

50mL	Phosphate builder (10% sodium phosphate) solution,
2 x 25mL	Non-phosphate builder UN1760
	(15% sodium silicate/5% citric acid) solution
50mL	Control (deionized) water
45	Plastic cups
1 box	Microscope slides
1 box	Coverslips
Optional: Microscopes, Plastic wrap or similar	

Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173.4 for domestic highway or rail transport only



Aldon

www.aldon-chem.com

\$50.37

A Greener Synthesis of Acetylsalicylic Acid

Show students it is possible to produce acetylsalicylic acid from a naturally-occurring, renewable resource. Students will convert sodium salicylate to salicylic acid, collect and dry prepared salicylic acid. They may then use a quick confirmatory test to examine for the presence of salicylic acid. The prepared salicylic acid can then be used to synthesize acetylsalicylic acid. Kit contains enough materials for 15 groups. Teacher's Manual and Student Study Guide copymasters are included.

Kit Includes:6 x 25mLSodium Hydroxide 3M6 x 25mLHydrochloric Acid 3M25mLMethyl Salicylate2 x 25mLAcetic Anhydride5mLPhosphoric Acid 85%5mLFerric Nitrate15Pipettes, Disposable

DOT Info: Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173.4 for domestic highway or rail transport only

IS9503

\$66.50

Determining the Composition of an Unknown Mixture

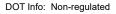
Often times, the composition of a mixture may contain a variety of unknown components. In some cases, the components of a mixture may be known but the exact amount of those components in the mixture is not. Analytical chemists often have a variety of tools and techniques to analyze unknown substances and arrive at conclusions with regards to the compounds/ percentages in the mixture. In this activity, students will determine the percent composition of sodium carbonate and sodium bicarbonate in an unknown sample. The mixture is heated vigorously until the sodium bicarbonate is completely decomposed to sodium carbonate. The only other products of the reaction are carbon dioxide and water. After performing the necessary calculations, students will determine the percentage of sodium bicarbonate that was present in their original sample. Kit contains enough materials for 15 groups. Teacher's Manual and Student Study Guide copymasters are included.



An Alternative lodine Clock Reaction

Students will learn the mechanisms and reactions involved in one type of clock reaction and understand how a clock reaction may provide insight into reaction kinetics. After assembling, performing, and obtaining data from several clock reactions students will alter experimental conditions and investigate the effects on clock reaction data. Determination of the effects of concentration and temperature on chemical kinetics will be investigated. Kit contains enough materials for 15 groups. Teacher's Manual and Student Study Guide copymasters are included.

Kit Includes: 3 X 100mL 3 X 100mL 400mL 3g 30 30 60 60	Ascorbic Acid Quick Solution, makes 100mL of 1% solution IKI Dilute Lugol's Solution Hydrogen Peroxide 3% Starch, Soluble Plastic Cups Stir Sticks Measuring Cups, Disposable Pipettes, Disposable
DOT Info: Non-r	egulated
IS9505 \$58.00	



Sodium carbonate

Sodium bicarbonate

Pre-labeled bottles (to prepare unknowns)

Kit Includes: 200g

200g

5



www.aldon-chem.com

IS9504

800-724-9877

Green Chemistry

Innovating Science®

Green Nanochemistry: Synthesis of Silver Nanoparticles

Nanotechnology is a field of technology that is described as the control and manipulation of matter in the 1 to 100 nanometer (nm) range. A nanometer is one billionth of a meter, or one millionth of a millimeter. To put it in perspective, an average sheet of copier paper is about 100,000 nanometers thick so a nanometer is 100,000th the thickness of a piece of copier paper. Certain materials, when prepared at nanoscale levels, display different physical and chemical properties than those of the same materials prepared on a larger scale. In this activity, students synthesize silver nanoparticles from silver nitrate. The particles are synthesized in glucose and starch, as opposed to the more hazardous reagents typically employed. Kit contains enough materials for 15 groups. Teacher's Manual and Student Study Guide copymasters are included.

Kit Includes:

The molado	5.
1mL	Silver nitrate, 0.1M
0.09g	Dextrose to make 5 mL of 0.1M Glucose
0.4g	Soluble starch (to make 200mL of 0.2% solution)
5mĹ	Sodium Hydroxide 0.1M

DOT Info: Non-Regulated

IS9506

\$61.66

The Hydrogen Fuel Cell Demonstration

In this demonstration, energy will be produced from combining of hydrogen and oxygen to form water. Platinum will serve as the catalyst and electrodes will be prepared by coating metal mesh with platinum. The hydrogen and oxygen will come from electrolysis. After the cell is set up, a brief current is applied (with a 9-volt battery) causing the formation of hydrogen gas bubbles on one electrode and oxygen gas bubbles on the other. Using a voltmeter, electricity produced by the recombining of hydrogen and oxygen, facilitated by the platinum metal catalyst, can be observed. Kit contains enough materials for 5 demos.



Green Fuel Cell: Energy From Yeast

The transfer of electrons is part of the metabolic process in living organisms. However, what if it were possible to capture, remove, and use some of these electrons? Could living organisms generate an electrical current? Students will set up a simple cell and using the simple-to-grow and environmentally tolerant organism yeast, as well as a special dye capable of entering yeast cells and collecting electrons, determine if the harvested electrons are capable of producing current in the cell and if so, how much current. This fun activity also serves as a great tool to stimulate discussion with regard to alternate energy sources. Kit contains enough materials for 15 groups. Teacher's Manual and Student Study Guide copymasters are included.



Aldon

Environmental Chemistry: Water Treatment and Filtration

Students will develop a knowledge of the processes performed at a water treatment plant and understand the reasons for each process. They will perform, on a small-scale, several of the procedures that occur in a water treatment plant on "polluted" water. They will examine the changes in the water after each treatment step is performed. They will also observe physical characteristics of water, such as clarity, color, odor, and how they are affected from the beginning of the treatment process until the end. Kit contains enough materials for 15 groups of students. Teacher's Manual and Student Study Guide copymasters are included.

Kit Includes:		
1500g	Fine sand	
1500g	Coarse sand	
1500g	Fine gravel	
300g	Activated charcoal	
25g	Potassium aluminum sulfate (alum)	
25g	Calcium oxide (lime)	
5g	Kaolin (clay) powder	
5mL	Green food coloring	
25mL	White Vinegar	
75	Plastic cups	
15	Stirring sticks	

DOT Info: Non-Regulated

WARNING: This product can expose you to chemicals including Respirable crystalline silica, which is known to the State of California to cause cancer. For more information go to www.P65Warnings.ca.gov.

\$78.99

IS9700

Environmental Chemistry: Nitrates, Phosphates, and **Eutrophication**

Understand the importance and value of fresh water and the need to protect and conserve this valuable resource. Realize that a variety of factors, including natural ones, contribute to the overall problem of water pollution. After completing this lab, students should understand the difference between point source and non-point source pollution and understand the role of nitrates and phosphates in the process of eutrophication. They will observe and examine the effects of nitrates, the effects of phosphates, and the effects of a combination of nitrates and phosphates in miniature "ponds." Kit contains enough materials for 15 groups. Teacher's Manual and Student Study Guide copymasters are included.



Environmental Chemistry: Acid Rain, Weathering, and Erosion

Acid rain is a term used generically to describe any type of acidic moisture, be it rain, snow, or fog. Acid rain can have devastating effects on not only aquatic ecosystems but also terrestrial areas. Acid rain not only affects naturally-occurring surfaces, such as exposed rocky surfaces of mountainous regions, but also human-made surfaces as well. Different stone and metal substances used in the construction of buildings, statues, monuments, etc. may all be affected by acid rain. The rates of erosion and weathering may increase rapidly depending on the degree of acidity in the precipitation. In this activity, students expose many rock materials and metals, both naturally-occurring and common in construction, to an acidic environment and examine the reaction of these materials in contrast to the same materials exposed to "normal" rain (tap water). Kit contains enough materials for 15 groups. Teacher's Manual and Student Study Guide copymasters are included.

Kit Include	s:
1 bag	Marble chips
1 bag	Brick chips
1 bag	Granite chips
1 bag	Limestone chips
1 bag	Sandstone chips
1 bag	Steel shot
1 bag	Copper shot
1 bag	Zinc shot
3 x 25mL	Sulfuric acid concentrate (to make 500mL each)
120	Measuring cups, disposable
DOT Info:	
	ity exemption 173.4
THIS PACKA	AGE CONFORMS TO 49 CFR 173.4
for domestic	highway or rail transport only
cher silica cont	RNING: This product can expose you to micals including Respirable crystalline a and Strong inorganic acid mists aining suffuric acid, which are known to the e of California to cause cancer. For more mation go to www.P66Warnings.ca.gov.

800-724-9877

60

Acid Rain and the Environment: Acidity and Plant Growth

The problem of acid rain is quite often associated with its effects on aquatic systems. Unfortunately, acid rain can also have devastating effects on terrestrial environments as well. In this activity, students will examine the detrimental effects of acidic conditions on plants. Plants will be grown under normal soil conditions, mildly acidic soil conditions, and very acidic soil conditions. Through physical observation, students will determine if the acidity has any impact on the growth of the plant. Kit contains enough materials for 15 groups.

Kit includes: 4 X 30mL 1 Bag 45 45 45 45 Pcs 1 Pkg Hydrochloric Acid 12M UN1789 Potting Soil 5 oz Translucent cup 7 oz Translucent cup Absorbent Wick Pad Radish Seeds



DOT Info: Small quantity exemption 173.4 This package conforms to 49 CFR 173.4 for domestic highway or rail transport only

IS9704

\$79.99

Oil Spill Cleanup: Biological vs. Physical

In this activity, students will compare two methods of oil spill cleanup: biological and physical. Applying both a special blend of oil degrading microbes and a hydrocarbon encapsulating polymer to oil and examining the results, students will draw conclusions with regard to the effectiveness of each approach. The kit includes a specially prepared stained vegetable oil to simulate crude oil, eliminating any of the hazards and associated disposal costs of the real thing while still allowing the students to perform the procedures using real oil. Kit contains enough materials for 15 groups. Teacher's Manual and Student Study Guide copymasters are included.



Effect of Salinity on Seawater

The world's ocean system contains a dynamic thermohaline current system that is responsible for many aspects of our environment. The experiments in this kit have been designed to help visualize the impact solutions of different density have on each other when mixed. Students will measure the salinity of three different water samples representing three major salt water bodies, as well as perform an experiment in which they will observe the flow patterns when different density solutions are mixed. This kit has enough materials for 15 groups. Teacher's Guide and Student Study Guide Copymasters included.



Determination of Dissolved Oxygen Concentration

This kit has been designed to allow the analyst to perform a very simple titration to determine the dissolved oxygen concentration in their fresh water sample. The titration uses an easily identifiable color transition to indicate the end of the analysis. The kit contains everything an analyst would need to perform a water analysis either in the laboratory setting or out in the field. Kit contains enough materials to perform 40 tests.

Kit Include	s:
4	Plastic Pipettes
3	Reaction Vessel
3	Collection Bottle
1	Measuring Tube
2	Titrators
30mL	Manganese Chloride Reagent
30mL	Alkaline Iodide Reagent
30mL	50% Sulfuric Acid
60mL	Sodium Thiosulfate Titrant
30mL	Starch Indicator Solution

DOT Info: Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173.4 for domestic highway or rail transport only



WARNING: This product can expose you to chemicals including Strong inorganic acid mists containing sulfuric acid, which are known to the State of California to cause cancer. For more information go to www.P65Warnings.ca.gov.

IS9750

\$38.50

Determination of Alkalinity of Water

This kit has been designed to allow the analyst to perform a very simple titration to determine the alkalinity of their fresh water sample. This kit allows the analyst to determine both the phenolphthalein alkalinity as well as the total alkalinity. The titration uses an easily identifiable color transition to indicate the end of the analysis. The kit contains everything an analyst would need to perform a water analysis either in the laboratory setting or out in the field. Kit contains enough materials to perform 40 tests.



Determination of Water Hardness

This kit has been designed to allow the analyst to perform a very simple titration to determine the hardness of their fresh water sample. The titration uses an easily identifiable color transition to indicate the end of the analysis. The kit contains everything an analyst would need to perform a water analysis either in the laboratory setting or out in the field. Kit contains enough materials to perform 40 tests.

Kit Includes:

6	Plastic Pipettes
3	Reaction Vessel
3	Collection Bottle
1	Measuring Tube
2	Titrator 3 mL
180mL	0.005 M EDTA Titrant
60mL	pH 10 Buffer Solution
15mL	Eriochrome Black T Indicator Solution

DOT Info: Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173.4 for domestic highway or rail transport only



IS9751

\$47.25

Determination of the Nitrite and Nitrate Concentration in Water

This kit is designed for an analyst to perform a simple colorimetric test to determine the nitrite and nitrate concentrations of fresh water samples. The reaction yields an easily identifiable color transition that can be compared with the colors on the color chart determining the concentrations. Identified nitrite and nitrate values are 0.0, 0.25, 0.5, 1.0, 2.0, 4.0, 6.0, 8.0 & 10.0 PPM. This kit contains everything an analyst would need to perform a water test either in the laboratory setting or out in the field. Kit contains enough materials to perform 40 tests.

Kit Includes: 60mL Griess Reagent 6 Plastic Pipettes 3 Collection Bottle

Wooden Dowels

 0.5 g
 Cadmium Metal Powder

 3
 Reaction Vessel

 1
 Measuring Tube

 1
 Nitrite & Nitrate Color Chart

DOT Info: UN2924, Flammable Liquids, Corrosive, n.o.s, (Ethanol, Phosphoric acid), 3 (B), III, Ltd Qty



WARNING: This product can expose you to chemicals including Methanol, Cadmium and cadmium compounds which are known to the State of California to cause cancer, reproductive harm. For more information go to www.P65Warnings.ca.gov.

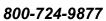


Aldon

56

IS9753

1 pkg



Determination of the Ammonia Nitrogen Concentration in Water

This kit has been designed to allow the analyst to perform a very simple colorimetric analysis to determine the ammonia nitrogen concentration of their fresh water sample. The analysis yields an easily identifiable color transition that can be compared with the included color key to determine the concentration in your sample. The kit contains everything an analyst would need to perform a water analysis either in the laboratory setting or out in the field. Kit contains enough materials to perform 40 tests.

Concentration Reader Range:

0, 0.25, 0.5, 0.75, 1.0, 1.5 and 2.0 mg NH /L Higher concentrations can be tested using a dilution of the water sample with DI water (not supplied)

Kit Includes: Plastic Pipettes 12 Innovating Sci 3 Reaction Vessel 3 **Collection Bottle** Measuring Tube ation of th 15mL Ammonia Test Solution #1 a biite Ammonia Test Solution #2 o Wate 15mL Ammonia Test Solution #3 15mL Color Code Indicator Chart DOT Info: Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173.4 for domestic highway or rail transport only

IS9754

\$47.25

Determination of Salinity

This kit has been designed to allow the analyst to perform a very simple titration to determine the salinity of their water sample. The titration uses an easily identifiable color transition to indicate the end of the analysis. This kit contains enough materials to perform 40 tests.



Ocean Acidification

Changes to the carbon cycle and climate due to human activities have had detrimental effects on the environment. One example of this is known as ocean acidification, which is the changing of the ocean's chemistry due to increased carbon dioxide in the atmosphere and ocean. In this activity, students will learn about the carbon cycle and how it relates to ocean acidification. Through three different activities, students will investigate the cause of ocean acidification, and the effects it can have on the ecosystem and marine life. Kit contains enough materials for 15 groups of students. Teacher's Manual and Student Study Guide Copymasters are included.



Aldon

Determination Of Dissolved Carbon Dioxide Concentration

The proper amount of dissolved carbon dioxide in water is critical for maintaining a healthy aquatic ecosystem. All aquatic life has an impact on and is in turn impacted by the dissolved carbon dioxide concentration in a system, so tracking the concentration and changes in concentration can help scientists better understand the health of an aquatic system and the organisms that exist within it. Kit contains enough materials to perform 40 tests.

Kit Contents:

3	Plastic Pipettes	
3	Reaction Vessel	
3	Collection Bottle	
1	Measuring Tube	
1	Titrator	
15mL	Phenolphthalein Indicator	
50mL	Potassium Hydroxide Reagent	

DOT Info:

Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173.4 for domestic highway or rail transport only

WARNING: This product can expose you to chemicals including phenolphthalein, which is known to the State of California to cause cancer. For more information go to http://www.P65Warnings.ca.gov.



IS9757

\$36.99

Plant Tissue Macronutrients

Plants need a variety of nutrients and minerals to grow and maintain overall health. The three main nutrients that plants require to survive and thrive are nitrogen, phosphorus, and potassium (also known as NPK). In this activity, students will extract nutrients from plants of your choice and perform analyses to determine if nitrogen, phosphorus, and potassium concentration is abundant, adequate, or deficient. Kit contains enough materials to perform 50 tests.

Kit includes:	
1 X 500mL	Nutrient Extractor Solution
1 X 30mL	Nitrogen Reagent #1
1 X 30g	Nitrate Reagent #2
1 X 30g	Phosphorus Reagent #1
1 X 30mL	Phosphorus Reagent #2
1 X 30g	Potassium Reagent #1
1 X 120mL	Potassium Reagent #2
Pk/100	Filter Paper
4	Collection Bottles
1	Spot Plate
2	Reagent Scoops
1	Stir Stick
5	Pipettes

DOT Info: UN1170 Ethanol, 3, II Ltd. Qty



The Carbon Cycle In Soil

Changes in the carbon cycle and climate due to human activities have had detrimental effects on the environment. In this activity, students will learn about the carbon cycle in soil and how soil impacts global warming. Through 3 different activities, students will investigate one cause of global warming, and the effects of a positive feedback loop. Kit contains enough materials for 15 groups of students. Teacher's manual and student study guide copymasters are included. Kit includes enough materials for 15 groups of students.

Kit Includes:

- 30 Nylon Tubing
- 30 One-hole stoppers
- 30 Soil respiration chambers2L Limewater
- 1kg Topsoil
- DOT Info: Non-regulated

IS9759

\$116.00



800-724-9877

Estuary Monitoring Water Test Kit

Test a water sample of your choice for coliform bacteria, chloride, dissolved oxygen, biochemical oxygen demand, nitrate, pH, phosphate, and temperature. Kit contains enough materials to perform 30 tests.

3

5

2

Reaction Vessels

5g Coliform Test Powder

Titrators

Color Charts

Kit Includes: Water Collection Bottles 3 Pipettes 15 3 Measuring Tubes 30 Phosphate Powder Pillows Toothpicks 1pkg pH Strips 1pka 2 x 30mL Starch Indicator Solution 2 x 30mL Manganese Chloride Reagent 2 x 30mL Alkaline Iodide Reagent 2 x 30mL 50% Sulfuric Acid Reagent 2 x 60mL Sodium Thiosulfate Titrant Griess Reagent 60mL 0.5g Cadmium Metal Powder 3x100mL Silver Nitrate Titrant 3 x 15mL Salinity Indicator Solution

DOT Info:

UN2924, Flammable liquids, corrosive, n.o.s (Ethanol, Phosphoric Acid), 3(8), III, LTD $\ensuremath{\mathsf{QTY}}$

Soil Macronutrients

Plants need a variety of nutrients and minerals to grow and maintain overall health. The three main nutrients that plants require to survive and thrive are nitrogen, phosphorus, and potassium (also known as NPK). In this activity, students will extract nutrients from a soil sample and perform analyses to determine if nitrogen, phosphorus, and potassium concentration is abundant, adequate, or deficient. Kit contains enough materials to perform 50 tests.

۲it	includes:	
	Soil (۰,

5	Soil Collection Bottles
5	Plastic Scoops
5	Pipettes
1	Color Chart
120mL	Soil pH Indicator
120mL	Nitrogen Extraction Solution
30g	Nitrogen Indicator Powder
120mL	Phosphorus Extraction Solution
30mL	Phosphorus Reagent
15g	Phosphorus Indicator Powder
120mL	Potassium Reagent
15g	Potassium Indicator Powder

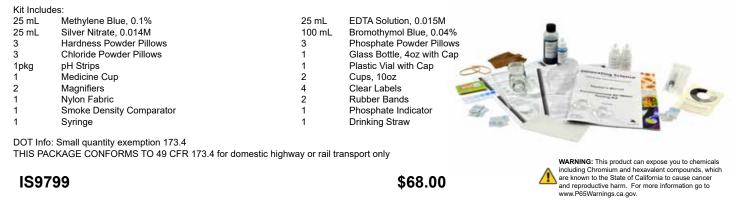
15g Potassium Indicator Powder

DOT Info: UN1170, Ethanol, 3, II, LTD QTY



Environmental Air/Water Test Kit

Learn how environmental scientists study pollution by testing air and water samples from your local area! This kit includes nine different tests to determine the air and water quality of samples you collect from your community. Investigate your water quality by testing for dissolved oxygen levels, water hardness, pH, and phosphate and chloride content. Then, test for particulate and chemical pollutants in the air, measure smoke density, and learn how excess carbon dioxide in air can affect your environment. Teacher's Guide and Student Analysis copymasters are included. There are enough materials for a class of 30 working in groups.



Aldon

Urban Water Testing Kit

Test water sample for chloride, copper, iron, hardness, nitrate, pH, phosphates, and temperature. This kit will allow students to learn about how water quality affects urban areas. Kit contains enough materials to perform 10 tests.

Kit Includes:	
3	Water Collection Bottles
6	Reaction Vessels
6	Measuring Tubes
15	Pipettes
6	Titrators, 1mL
2	Titrators, 3mL
3	Color Charts
1pkg	Toothpicks
10	Phosphate Powder Pillows
1pkg	pH Strips
10	Iron Test Tablets
10	Copper Test Tablets
5g	Coliform Test Powder
60mL	pH Buffer 10 Solution
15mL	Eriochrome Black T Solution
60mL	Griess Reagent
0.5g	Cadmium Metal Powder
3 x 100mL	Silver Nitrate Titrant
3 x 15mL	Salinity Indicator Solution
3 x 60mL	EDTA Titrant
1pkg 10 1pkg 10 5g 60mL 15mL 60mL 0.5g 3 x 100mL 3 x 15mL	Color Charts Toothpicks Phosphate Powder Pillows pH Strips Iron Test Tablets Copper Test Tablets Coliform Test Powder pH Buffer 10 Solution Eriochrome Black T Solution Griess Reagent Cadmium Metal Powder Silver Nitrate Titrant Salinity Indicator Solution

DOT Info:

UN2924, Flammable liquids, corrosive, n.o.s (Ethanol, Phosphoric Acid), 3(8), III, LTD QTY

WARNING: This product can expose you to chemicals including methanol, cadmium and cadmium compounds, and chromium hexavalent compounds, which are known to the State of California to cause cancer and reproductive harm. For more information go to http://www.P65Warnings.ca.gov.



IS9762

\$140.00

Total Water Investigation Kit

Test water sample for alkalinity, ammonia, biochemical oxygen demand, chloride, coliform bacteria, copper, hardness, iron, nitrate, dissolved oxygen, pH, and phosphate. Kit contains enough materials to perform 10 tests.

Kit Includes:			
3	Water Collection Bottles		
6	Reaction Vessels		
6	Measuring Tubes		
15	Pipettes		
6	Titrators,1 mL		
2	Titrators, 3 mL		
4	Color Charts		
1pkg	Toothpicks		
10	Phosphate Powder Pillows		
1pkg	pH Strips		
10	Iron Test Tablets		
10	Copper Test Tablets		
5g	Coliform Test Powder		
60mL	pH Buffer 10 Solution		
15mL	Eriochrome Black T Solution		
60mL	Griess Reagent		
0.5g	Cadmium Metal Powder		
15mL	Total Alkalinity Indicator		
15mL	· · · · · · · · · · · · · · · · · · ·		
15mL			
	Ammonia Test Solution 2		
15mL	Ammonia Test Solution 3		
3 x 60n			
3 x 100			
3 x 15n	nL Salinity Indicator Solution		
3 x 60n	nL EDTA Titrant		
2 x 30n	nL Starch Indicator Solution		
2 x 30n	5 - 5		
2 x 30n	·- · · · · · · · · · · · · · · · · · ·		
2 x 30n	nL 50% Sulfuric Acid		
2 x 60n	nL Sodium Thiosulfate Titrant		

DOT Info:

UN2924, Flammable liquids, corrosive, n.o.s (Ethanol, Phosphoric Acid), 3(8), III, LTD QTY

WARNING: This product can expose you to chemicals including Phenolphthalein, chromium hexavalent compounds, strong inorganic mists containing sulfuric acid, methanol, cadmium and cadmium compounds, which are known to the State of California to cause cancer and reproductive harm. For more information go to http://www.P65Warnings.ca.gov.



IS9763

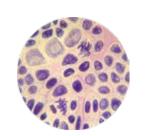
\$225.00

Aldon

DNA/Chromosome Staining

Prepare your own squashed stained slide and be able to identify the phases of plant mitosis and chromosomal development. Students will be able to prepare, stain and mount slides using specially prepared onion root tips. Kit contains enough material for 15 groups. Teacher's Manual and Student Study Guide copymasters are included.

Kit Includes: 1 vial Preserved root tips 2 x 25mL 6M Hydrochloric acid 2 x 25mL Toluidine blue 1.0% Solution 1 box Microscope slides 1 box Coverslips 30 Aluminum dishes 30 Forceps



DOT Info: Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173.4 for domestic highway or rail transport only

IS3000

\$71.99

Osmosis and Diffusion Lab

This lab allows you to learn about two forms of passive transport: diffusion and osmosis. You will compare and contrast similarities and differences in the processes of diffusion and osmosis. Use a colorimetric test to demonstrate the movement of a solute across a semi-permeable membrane. Set up an environment likely to facilitate osmosis and gather data to determine whether or not osmosis may have occurred. Kit contains enough materials for 15 groups. Teacher's Manual and Student Study Guide copymasters are included.

Kit Includes:		
30 pc.	Dialysis tubing	
1 cpsl	Starch (to make 100mL of 1.0% solution)	
1 btl.	Sucrose (to make 100mL of 0.5M solution)	
1 btl.	lodine/potassium iodide solution, 15mL	
30	Plastic cups	

DOT Info: Non-regulated

DNA Extraction

WARNING: This product can expose you to chemicals including Methanol/Methyl isobutyl ketone which is known to the State of California to cause cancer and reproductive harm. For more information

go to www.P65Warnings.ca.gov.

In this lab you will learn the history of the discovery of DNA and DNA structure. Understand the nature of genetic inheritance and the role of DNA and proteins in genetic expression while using biological detergents, enzymes, and ethanol to isolate DNA from plant material. You need to supply the plant material. Kit contains enough materials for 15 groups. Teacher's Manual and Student Study Guide copymasters are included.

Kit Includes:

- 2 x 250mL 7.5% SDS/1.5% NaCl
- Pepsin (to make 25mL 0.5% solution) 2 x 25mL 95% ethanol
- 20 Zipper bags
- 15 Filters
- 15 Plastic tubes
- 30 Graduated pipettes
- 15 Stirrers

DOT Info: Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173.4 for domestic highway or rail transport only



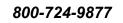
IS3001

\$62.69

the State of California to cause cancer reproductive harm. For more information go to www.P65Warnings.ca.gov.

\$45.00

Aldon



Innovating Science[®]

Diffusion and Cell Size

Why are cells microscopic? The answer relates to the needs for the cell to effectively move materials in and remove waste. In this activity, students will create simulated cells (agar blocks) of different sizes and examine how effectively a substance is able to diffuse into the cell in a set period of time. A special indicator in the cells will allow students to visualize the degree of diffusion. The results will clearly display the fact that a smaller volume creates a more favorable condition for the exchange of material across a cell membrane. Kit contains enough materials for 15 groups of students. Teacher's Manual and Student Study Guide copymasters are included.



Anesthefly Kit

Anesthetize Drosophila melanogaster (fruit fly) and other small insects for at least 40-50 minutes without killing or sterilizing. Includes instructions.



Kidneys and Blood Filtration

Learn the role of the kidney in blood filtration and waste removal along with the many functional tasks performed by nephrons, as well as nephron structure. Students will create an artificial kidney model to filter simulated blood. This will allow them to visually determine if filtration of the simulated blood may or may not have occurred. Chemically test the resulting filtrate to detect any possible waste material that may have been removed by the kidney. Kit contains enough materials for 15 groups. Teacher's Manual and Student Study Guide copymasters are included.



Urinalysis Using Simulated Urine

Urinalysis, one of the oldest medical diagnostic tests performed, is to this day still one of the most common. In this activity, students will use simulated urine to avoid the unpleasantness of using the real thing while still performing actual tests used on real urine samples. Students will examine the simulated urine for factors such as pH, color, clarity, as well as test for the presence or absence of proteins, glucose, and calcium. The students will then examine the samples microscopically to determine if crystals may be present in any of the samples. Kit contains enough materials for 15 groups of students. Teacher's Manual and Student Study Guide copymasters are included. Needed but not supplied are a hot water bath, glass test tubes, and compound microscopes (4X/10X/40X).

Kit Include	s.			ups.
4	Simulated urine samples, 250m Patient X	Leach	-	ide c
2 x 25mL 2 x 25mL 2 x 25mL 1 pkg. 1 box 1 pkg. 60 60	Patient X Patient Y Patient Z Control Benedict's Qualitative Biuret Reagent Sulkowitch reagent pH test strips, 100/vial Microscope slides Coverslips Graduated plastic cups, 30mL Graduated pipettes, 1mL		15 15 15 15 15	Info: 1
DOT Info:	Small quantity exemption 173.4			-

This package conforms to 49 CFR 173.4 for domestic highway or rail transport only

			 1000
IS3008	\$75.00		
IS3008-REF - Replacement Urine for IS3008	\$36.50	IS3010	\$85.00

Enzymes and the Process of Digestion

All the food in the world is of no use if the human body does not have the ability to extract necessary nutrients from it. With this activity, students will be able to expose three nutrients (carbohydrates, proteins, and lipids) to different digestive enzymes. These samples will be compared to nutrients to which no enzymes are added and chemical tests will be used to determine if the enzymes were effective in digesting the compounds. Upon completion, students will not only understand the importance of the digestive system but also the vital role enzymes play in releasing nutrients from food and converting them to a form usable by the body. Kit contains enough materials for 15 groups. Teacher's Manual and Student Study Guide copymasters are included.



Introduction to **Mendelian Genetics**

In this activity, students can simulate Mendel's work

and determine patterns of inheritance. Using special chips and Innovating Science's exclusive "double dice," students will be able to simulate both monohybrid and dihybrid crosses. After the crosses, students will be able to determine genotypic and phenotypic ratios for select traits and compare their values to the theoretical "ideal" values as put forth by Mendel. Kit contains enough materials for 15 Teacher's Manual and Student Study copymasters are included.

- nohybrid chips, female (red/yellow)
- nohybrid chips, male (red/white)
- ybrid double dice, female (colored) vbrid double dice, male (clear)
- stic shaker cups

Non-regulated



Best

Seller

Electrophoresis: Agarose Gel Separation of Dyes

Introduce your students to this valuable separation science in a safe and colorful manner. Unlike DNA and other molecules which cannot be seen during electrophoresis, this activity uses dyes that can be observed during the actual procedure, providing visual reinforcement of the forces driving molecular movement and separation in the electrophoresis process. Kit contains enough materials to run ten 20mL agarose gels (actual number of runs may vary based on your equipment). Teacher's Manual and Student Study Guide copymasters are included. Not included but required are agarose electrophoresis chambers, electrophoresis power supplies and micropipettes capable of measuring 10µl.



WARNING: This product can expose you to chemicals including Crystal Violet, which is known to the State of California to cause cancer. For more information go to www.P65Warnings.ca.gov.

IS3011 \$57.50

Testing Food For Nutrients

This experiment will help students understand the importance of proteins, carbohydrates, and lipids in living organisms. They will learn to identify a positive test result for proteins using biuret reagent and examine the reaction between Benedict's reagent and a simple sugar. Using iodine/potassium iodide they will test for the presence of starch and test for the presence of lipids using a fat-soluble dye.



Cellular Respiration: What Sugar Does Yeast Like Best?

In this experiment you will expose living yeast cells to three different sugars. The sugars used will be glucose, sucrose, and lactose. When living yeast cells are exposed to these sugars, the cells should begin to utilize the sugars as a food source if they are capable of metabolizing them. Upon using the sugars, the yeast cells will begin to engage in aerobic respiration and/or fermentation. This lab allows students to understand that yeast may use different options for energy production. Students will expose living yeast cells to three different potential food sources and use a pH indicator to indirectly determine the effectiveness of three different sugars as a food source for yeast. Kit contains enough materials for 15 groups.

> Litmus blue 0.5% Yeast instant dry Dextrose (d-glucose) Sucrose Lactose monohydrate Sodium hydroxide 0.1N 0.1M



\$42.15

IS3012

DOT Info: Non-regulated

Kit Includes:

50mL

5g

50g

50g

50g

25ml

Introduction To Microbiology: Bacterial Growth And Staining

Bacteria, good and bad, is all around us. In this activity students will collect and grow bacteria and then learn techniques for studying the bacteria they have grown. They will perform a simple staining technique on bacterial cells to study the morphology. They will also perform a differential staining technique, Gram staining, on bacterial cells and determine if the collected bacteria are Gram-

positive, Gram-negative, or both.

	Kit Includes: 20 Sterile Petri dishes 2 x 200mL Prepared nutrient agar 20 Sterile cotton swabs, pk/2 30mL Methylene blue 30mL Crystal violet 30mL Gram's iodine 30mL Safranin O 4 x 30mL 95% Ethanol 1 Microscope slides, pk/72
Transfer Sciences Transfer Secure May Freed for Mandesates	DOT Info: Small quantity exemption 173.4 This package conforms to 49 CFR 173.4 for domestic highway or rail transport only
<u> </u>	WARNING: This product can expose you to chemicals including Crystal Violet and Methano/Methyl isobutyl ketone, which are

Methanol/Methyl isobutyl ketone, which are known to the State of California to cause cancer, reproductive harm. For more information go to www.P65Warnings.ca.gov.

IS3014

\$87.00

Aldon

800-724-9877

ATP Muscle Set

Students can observe and measure how muscles contract. The muscles are shipped in glycerol which extracts most of the water soluble material. This creates muscle tissue that can be identified and studied using a microscope. The muscle can also be stimulated to contract using the ATP chemical solution. There are enough materials for 15 groups. Perishable materials must be refrigerated until use.

Kit Includes: 1 pack Microscope slides 1 pack Glass covers 25mL Rabbit Psoas Muscle Solution A 25mL Rabbit Psoas Muscle Solution C Pack of 3 Rabbit Psoas Muscle in Glycerin DOT Info: Non-regulated

IS3015

\$145.00

Plant Food – Nutrient Deficiency in Plants

Using the materials provided, students will be able to examine the effects of nutrient deficiency on plants. Specially prepared nutrient solutions (included in the kit) will allow students to deprive each plant of one of seven specific vital nutrients. Over time, students will determine the effects, if any, of the nutrient deprivation through physical observation of the plant's growth. Kit contains enough materials for 3 complete set-ups. Teacher's Manual and Student Study Guide copymasters are included.

Kit Includes: Solution A - 200mL Solution B - 200mL	Calcium Nitrate 1M Potassium Nitrate 1M
Solution C - 100mL Solution D - 50mL	Magnesium Sulfate 1M Potassium Phosphate 1M
Solution E - 25mL	Sodium Phosphate 1M
Solution F - 100mL	Sodium Nitrate 1M
Solution G - 25mL	Magnesium Chloride 1M
Solution H - 25mL	Sodium Sulfate 1M
Solution I - 50mL Solution J - 50mL	Calcium Chloride 1M Potassium Chloride 1M
Solution 5 - 50mL	Micronutrient Solution
1 bag	Vermiculite
24	Cup 5oz. Clear Plastic
24	Cup 7oz. Clear
24 Pieces	Absorbent Wick Pad
1 Pkg	Radish Seeds
DOT Info - Non - Reg	
Let me from frog	
IS3050	\$72.00

Deluxe Owl Pellet Dissection Lab Activity

Owls have a specialized digestive tract which helps them expel the indigestible parts of their prey in pellet form. Dissecting an owl pellet can teach students the concepts of food webs and food chains while identifying the small animals that are prey for the owls. The pellets are heat sterilized and individually wrapped. Kit contains enough materials for 15 groups. Teacher's Manual and Student Study Guide copymasters are included.

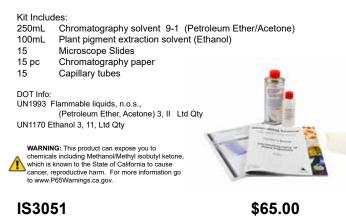
Kit Inc	ludes:	
15	Owl Pellets	Small parts.
30	Wooden probe	Not for children under 3 yrs.
15 60 15 15	Forceps Gloves Magnifier Small ruler	WARNING: This product can expose you to chemicals including Lead and lead compounds, which are known to the State of California to cause cancer, reproductive harm. For more information go to www.P65Warnings.ca.gov.
DOT In	fo: Non-regulated	

IS3016

\$225.00

Chromatography of Plant Pigments

Chlorophyll is the most prevalent and well-known plant pigment related to photosynthesis. It is not, however, the only plant pigment necessary for photosynthesis to occur. Other pigments are involved in the process. These pigments are often overlooked as they tend to be masked by the abundance of the green pigment chlorophyll. In this activity, students will extract the various pigments from green plant material and separate the pigment using chromatography. Students will not only confirm the presence and learn the role of these "hidden" pigments but also learn about chromatography as a technique for separating molecules. Kit contains enough materials for 15 groups and includes Teacher's Manual and Student Study Guide copymasters.



Innovating Science[®]

Cells - It's Alive! A Guided Inquiry

In this experiment, students will plan and execute their own investigation in which they will observe a variety of samples under the microscope to explore the various parts of a cell and distinguish between living versus non-living substances, animal cells versus plant cells, and unicellular versus multicellular organisms.

Kit Includes: **Prepared Slides:** Innovating Science Human Hair Thread Sand Teacher's Manual **Onion Bulb Epidermis Onion Root Tip** Cells - It's Alive! Elodea Yeast A Guided Inquiry Euglena Paramecia 193052 Human Cheek Cells Human Nerve Cells DOT Info: Non-regulated IS3052 \$84.11

Antimicrobial Properties of Essential Oils

Essential oils have been used for a variety of purposes throughout history for their scents, flavors, and pharmacological properties. Some oils are commonly used in cleaning products, but which ones are most effective? In this activity, students will perform the Kirby-Bauer test on a variety of essential oils to determine their antimicrobial efficacy.

Kit Includes:

11

10mL	Peppermint Oil, with Dropper	3 x 9mL
10mL	Lavender Oil, with Dropper	2 x 200mL
10mL	Cinnamon Oil, with Dropper	
10mL	Lemon Oil, with Dropper	2
10mL	Balsam Oil, with Dropper	5
15	Sterile Cotton Swabs, 2-pack	20

Nutrient Broth Tubes Nutrient Agar B. Subtillis Freeze Dried Vial 50 pack, Blank Sterile Disks Forceps Petri Dishes

DOT Info⁻

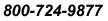
Small quantity exemption 173.4 This package conforms to 49 CFR 173.4 for domestic highway or rail transport only

WARNING: This product can expose you to a chemical Safrole which is known to the State of California to cause cancer. For more information go to www.P65Warnings.ca.gov.

IS3055

\$129.29





Life Science

Innovating Science®

Simulated ABO Blood Typing

The first blood typing system discovered, the ABO system, is the most important and widely used. In this activity, students will determine the ABO blood type of four unknown samples. Utilizing Innovating Science's new simulated blood, students will come to understand the nature and importance of antigen-antibody reactions. The most realistic simulated blood available, this activity provides the most realistic simulation of the actual blood typing procedure. Kit contains enough materials for 10 groups. Teacher's Manual and Student Study Guide copymasters are included.

Kit Includes:

4	Simulated Blood Samples		
	Donor #1		
	Donor #3		
1 btl	Simulated anti-A serum		
1 btl	Simulated anti-B serum		
40	Blood typing trays		
1pkg	Toothpicks		

DOT Info - Non-Regulated

IS3100	\$45.00
IS3100-REF - ABO Blood Typing Refill	\$27.00

Donor #2 Donor #4

Simulated ABO/Rh Blood Typing

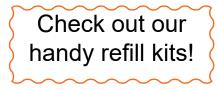
This activity provides the most procedurally accurate simulation of the blood typing technique available. Students test and determine the ABO/Rh blood types of four different simulated blood samples. Using Innovating Science's new simulated blood, the students combine blood samples and antisera, gently agitate the blood typing tray, and observe the results. No toothpicks, no stirring, and no waiting for results required. Kit contains enough materials for 10 groups. Teacher's Manual and Student Study Guide copymasters are included.

Kit Include	es:
4	Simulated Blood Samples
	Donor #1
	Donor #2
	Donor #3
	Donor #4
1 btl	Simulated anti-A serum
1 btl	Simulated anti-B serum
1 btl	Simulated anti-Rh serum
40	Blood typing trays
1pkg	Toothpicks
DOT Info	- Non-Regulated
	644
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-	
	Innovating Science



IS3101 \$55.00 IS3101-REF ABO/Rh Blood Typing Refill \$31.00





Genetics of Blood Types (Simulated)

Blood type, an inherited characteristic, has use in everything from forensic investigations to medical procedures. In this activity, students will learn about the genetics that determine blood type and the possible inheritance patterns and how they express themselves. Students will use Innovating Science's new simulated blood to determine the blood type of four unknown samples and use their results to assist in the resolution of a fictional paternity dispute. Kit contains enough materials for 10 groups. Teacher's Manual and Student Study Guide copymasters are included.

es:	
Simulated Blood Sample	s
Mother	
Child	
Sample X	
Sample Y	
Simulated anti-A serum	
Simulated anti-B serum	B
Blood typing trays	and the second sec
Toothpicks	
- Non-Regulated	
	H
	Child Sample X Sample Y Simulated anti-A serum Simulated anti-B serum Blood typing trays



IS3102
IS3102-REF Genetics of Blood Refill

\$51.38 \$27.76

Understanding Blood Type Interactions through Simulated Blood Typing

The earliest historical attempts at blood transfusions often had lethal results. These results led to the investigation and discovery of blood types, as well as a deeper understanding of the importance of antigen/antibody interactions. In this investigation, students will utilize Innovating Science's new simulated blood to determine the ABO/Rh blood type of four individuals, one in need of a transfusion. Based on the results, students will then determine which of three potential donors would provide the best match for the patient in need. Kit contains enough materials for 10 groups. Teacher's Manual and Student Study Guide copymasters are included.

Kit Include: 40 4	s: Blood typing trays Simulated Blood Samples Donor #1 Donor #2 Donor #3 Patient
1 btl	Simulated anti-A serum
1 btl	Simulated anti-B serum
1 btl	Simulated anti-Rh serum
1pkg	Toothpicks

DOT Info - Non - Regulated





IS3104

IS3104-REF

Forensics Using Simulated Blood

Though the use of blood type in a forensic investigation is not enough to prove guilt, it may aid in exonerating a potential suspect. In this activity, students act as lab technicians and assist investigators by examining evidence collected at a crime scene. Students first use a presumptive blood test to determine if a substance on a stained piece of cloth may be blood, and then determine the blood type of a sample collected at the scene. Students then compare their results to those of samples provided from two suspects. Kit contains enough materials for 10 groups. Teacher's Manual and Student Study Guide copymasters are included.

Kit Include 40 10 1 pc 1 pkg 1 btl 1 btl 4 1 btl 4	es: Blood typing trays Cotton swabs Cloth Toothpicks 70% ethanol Blood detection reagent #1 Blood detection reagent #2 Simulated blood samples Victim Suspect #2 Simulated anti-A serum	Suspect #1 Crime Scene Evidence	
1 btl 1 btl	Simulated anti-B serum Simulated anti-Rh serum		WARNING: This product can expose you to chemicals including Phenolphthalein and Methanol/Methyl isobutyl ketone, which are
	tity exemption 173.4 ge conforms to 49 CFR 173.4 for domes	tic highway or rail transport only	known to the State of California to cause cancer, reproductive harm. For more information go to www.P65Warnings.ca.gov.

Determination of Cholesterol Using Simulated Blood

Cholesterol levels can have a major impact on your health and your risk of heart disease. In this activity, students will measure the total cholesterol level for four patients using test strips and simulated blood. After analyzing their results and the lifestyles of each patient, students will recommend treatment plans to help the patients lower their cholesterol. The cholesterol levels for each patient will be tested again, and students will determine whether or not their treatment plans were successful. Kit contains enough materials for 15 groups. Teacher's Manual and Student Study Guide copy masters are included.

Kit Includes: 8 Simulated Blood Samples Patient #1 Pre Treatment Patient #1 Post Treatment Patient #2 Pre Treatment Patient #2 Post Treatment Patient #3 Pre Treatment Patient #3 Post Treatment Patient #4 Pre Treatment Patient #4 Post Treatment 15 Spot Plates 120 pH Strips 15 Cholesterol Charts IS3105 \$110.00 DOT Info: Non-regulated



68

www.aldon-chem.com

800-724-9877

Aldon

\$70.43

\$52.99

Innovating Science[®]

Diagnosis of White Blood Cell Counts

While white blood cells, or leukocytes, only account for approximately 1% of the total number of cells found in blood, they are a critical part of the immune system, protecting the body from foreign invaders such as bacteria and viruses. White blood cells counts are often performed as part of a medical diagnosis to determine if the level of white blood cells in a patient's blood is within or outside the normal range. In this activity students will perform white blood cell counts on five different patient samples and suggest possible diagnoses based on the white blood cell levels. The samples contain safe, non-biological simulated cells that do not require staining. Requires microscopes and hemocytometers (not included). Contains enough material for 15 groups and includes Teacher's Guide and Student Study copymasters.

Kit Includes: **5** Simulated Blood Samples Patient #1 Patient #2 Patient #3 Patient #4 Patient #5

DOT Info: Non-regulated

IS3106 \$39.05

Blood Typing Sets

For blood typing using real human blood. 5 mL vials contain enough antisera for 75 students. Sample will keep over one year with proper refrigeration.

FOR EDUCATIONAL USE ONLY. NOT FDA APPROVED FOR CLINICAL USE.

IS3150	Blood Typing Anti-Sera: Anti-A and Anti-B	\$34.25
IS3151	Blood Typing Anti-Sera Type A, B and Rh	\$64.25
IS3152	Blood Typing Anti-Sera: Anti Rh	\$32.50
IS3153	Blood Typing Anti-Sera: Anti-H Lectin	\$49.99



ABO/Rh Blood Typing Tray

These styrene trays are washable and reusable. They contain depression wells to perform ABO and Rh blood-typing. Package of 100.



DOT: Non-regulated

IS3155

Aldon

\$53.46



Freeze Dried ABO Blood Typing Sets

Use for blood typing of real human blood. Five milliliter vial contains enough antisera for 75 students. No refrigeration needed for freeze dried samples.

nnovating Science

IS3170 Blood Typing Anti-Sera: Anti-A and Anti-B, Freeze Dried - \$40.50 IS3171 Blood Typing Anti-Sera Type A, B and Rh, Freeze Dried - \$72.50 IS3172 Blood Typing Anti-Sera: Anti Rh, Freeze Dried - \$36.25 DOT: Non-regulated



Erycard[™] Blood Typing Card

Determine ABO or ABO/Rh blood type of real human blood in under five minutes.

A small amount of blood is applied to a series of wells containing lyophilized blood typing antibodies, followed by the addition of a wash buffer that flows across each well. Red blood cells that agglutinate in the presence of a particular antibody will not be rinsed from the sample well and the well will remain red, indicating a positive result. Red blood cells

that have not undergone agglutination will be rinsed from the well and the well will revert to white, indicating a negative result. Each card contains a negative control well to ensure accurate results. **For Educational Use Only

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10.0		•			
	in	in		10	ĥ

DOT Info: Non-regulated	
IS3180 - Erycard ABO Blood Typing card Pack of 24	\$78.00
IS3181 - Erycard ABO/Rh Blood Typing card Pack of 24	\$89.00

Erycard[™] Blood Typing Single Test Kit

Determine ABO/Rh blood type of real human blood in under five minutes.

A small amount of blood is applied to a series of wells containing lyophilized blood typing antibodies, followed by the addition of a wash buffer that flows across each well. Red blood cells that agglutinate in the presence of a particular antibody will not be rinsed from the sample well and the well will remain red, indicating a positive result. Red blood cells that have not undergone agglutination will be rinsed from the well and the well will revert to white, indicating a negative result. Each card contains a negative control well to ensure accurate results. Kit contains enough materials for 1 test. Instructions are included.

Kit Includes: 6 mL Buffer Solution 1 Alcohol Wipe 4 Blood Collection Sticks

1 Erycard™ 1 Lancet

DOT Info: Non-regulated

This test kit is designed for educational/instructional use only. This test kit is not designed for clinical/medical/diagnostic use.



IS3185

\$11.50

Determination of Blood Type Using Real Blood and Saliva

This kit has been designed to allow students to test their own saliva to determine their ABO blood type. Students that are part of the ~80% of the human population that secrete their blood antigens will perform titers to understand which blood type they belong too. These titers will be performed using supplied screened/typed human blood samples allowing the students to better understand how real blood typing works and what would happen if the wrong blood was given to a patient needing a transfusion. The initial box contains all of the non-perishable supplies including a disinfecting solution spray as well as a coupon that is to be redeemed for the perishable components near the time of use. This kit has all of the materials necessary for 30 students working in groups of three. Teacher's Manual and Student Study Guide copymasters are included.



800-724-9877



Photosynthesis: The Hill Reaction

The ability to convert light energy into usable chemical energy (photosynthesis) makes life possible. The process of photosynthesis can be broken down into two parts known as light and dark cycles. The activities in this kit focus on understanding the function of the light cycle by running experiments based on the Hill reaction. The instructions in this kit are written for both visual analysis and/or UV-Vis analysis. Kit contains enough materials for 6 groups. Teacher's manual and Student Study Guide copymasters are included.

Kit Includes:		
40mL	10% Propylene Glycol Solution	
500mL	Plant Extract Buffer	
.05g	0.05% Dichloroindophenol makes 100mL solution	
18g	0.5 M Sucrose makes 100mL solution	
0.25g	Sodium Hydrosulfite	
2	Filters	
30	Graduated Pipettes	
0.25g 2	Sodium Hydrosulfite Filters	

DOT Info: Small quantity exemption 173.4 This package conforms to 49 CFR 173.4 for domestic highway or rail transport only



IS3500

\$78.00

Introduction to Enzyme Catalyzed Reactions

Living organisms require enzymes to regulate the chemical reactions necessary to maintain life. The activities in this kit are designed to introduce students to the concepts of enzymes. Students perform a series of four experiments involving invertase, amylase and papain, demonstrating different enzyme functions, activities and specificities. Kit contains enough materials for 10 groups. Teacher's manual and Student Study Guide copymasters are included.



Introduction to Basic, Selective, and Differential Media

In this activity, students will learn about the different types of media that can be used to grow bacteria. They will predict which media will support the growth of different bacteria, then streak the different media with bacteria cultures to observe their growth. The bacteria growth will be evaluated to determine if students' predictions were correct. Kit contains enough material for 10 groups.



ife Science

Aldon

Introduction to Streak Isolation and Aseptic Technique

In this activity, students will be introduced to the skills and techniques that microbiologists use on a daily basis. Students will practice these techniques by performing a streak isolation with the provided bacterial cultures, as well as samples they collect from their classroom, to learn about different types of media and how bacterium reproduce. Kit contains enough material for 10 groups.

Kit Includes: 2x200mL Prepared tryptic soy agar 10 Lyophilized cultures of Serratia marcescens 20 Sterile petri dishes 10 Tryptic soy broth tubes 10 Inoculation loops 10 Sterile pipettors 10 Sterile cotton swabs, pk/2

DOT Info: Non-regulated

IS3611

\$95.60

Freeze Dried Bacteria Cultures

All cultures are freeze dried into a lyophilized pellet packaged in a sterile plastic vial. The culture includes one tube of rehydrating broth and one tube agar slant, sterile swab and instructions.

IS3901 Bacillus subtilis - \$19.99
IS3902 E. coli - \$19.99
IS3903 Micrococcus luteus - \$19.99
IS3904 Serratia marcescens - \$19.99
IS3906 Staphylococcus epidermidis - \$19.99
IS3907 E. coli jm101 freeze dried - \$19.99
IS3908 Bacillis cereus freeze dried - \$19.99
IS3909 Enterobacter aerogenes, freeze dried - \$19.99

DOT Info: Non-regulated



Student Bacteria Science Kit



Antibiotic Discs

Each cartridge contains 50 disks of antibiotics with a reclosable cap. Disks are 1/4 inch in diameter.

Available Antibiotics and part number:

XD1361	\$12.95	
XD1362	\$12.95	
XD1363	\$12.95	
XD1364	\$12.95	
XD1365	\$12.95	
XD1366	\$12.95	
XD1368	\$12.95	
XD1360	\$5.66	
XD1397	\$12.95	
XD1398	\$12.95	
	XD1361 XD1362 XD1363 XD1364 XD1365 XD1366 XD1368 XD1360 XD1397	

DOT Info: Non-regulated





Aldon

Bacteria are often associated with various diseases, infections, and other sicknesses. Certain bacteria can have negative effects on other living organisms, but in reality the vast majority of bacteria range from harmless to incredibly beneficial. This kit contains materials for students to use in investigating bacteria in their environment. While learning proper microbiological techniques, you will test hands, mouths, and your classroom for bacteria, before and after cleaning.

Kit Includes:2 x 200mLNutrient Agar20Sterile Petri Dishes20Sterile Cotton Swabs

DOT Info: Non-regulated

IS5100

\$53.44



Dehydrated Microbiology Media We can help you with your microbiology needs!

We can help you with your microbiology needs! Check our comprehensive list below or contact us for custom agar. DOT Info: Non-regulated



ltem #	Description	Amount Needed per Liter	Price
IS5110	Nutrient Agar, 1kg	23g/L	\$205.00
IS5111	Nutrient Broth, 1kg	8g/L	\$205.00
IS5112	Tryptic Soy Agar, 1kg	40g/L	\$243.64
IS5113	Tryptic Soy Broth, 1kg	25g/L	\$201.49
IS5114	Luria Agar (Miller's), 1kg	40g/L	\$166.54
IS5115	Luria Broth (Miller's), 1kg	25g/L	\$143.90
IS5116	Sabouraud-Dextrose Agar, 1kg	65g/L	\$143.90
IS5117	Sabouraud-Dextrose Broth, 1kg	50g/L	\$135.70
IS5118	EMB(Levine) Agar, 1kg	36g/L	\$193.26
IS5119	MacConkey Agar, 1kg WARNING: This product can expose you to chemicals including Crystal Violet, which is known to the State of California to cause cancer. For more information go to www.P65Warnings.ca.gov.	50g/L	\$201.49
IS5130	Nutrient Agar, 100g	23g/L	\$27.25
IS5131	Nutrient Broth, 100g	8g/L	\$31.36
IS5132	Tryptic Soy Agar, 100g	40g/L	\$29.79
IS5133	Tryptic Soy Broth, 100g	25g/L	\$27.25
IS5134	Luria Agar (Miller's), 100g	40g/L	\$23.14
IS5135	Luria Broth (Miller's), 100g	25g/L	\$19.79
IS5136	Sabouraud-Dextrose Agar, 100g	65g/L	\$19.27
IS5137	Sabouraud-Dextrose Broth, 100g	50g/L	\$18.65
IS5138	EMB(Levine) Agar, 100g	36g/L	\$25.20
IS5139	MacConkey Agar, 100g WARNING: This product can expose you to chemicals including Crystal Violet, which is known to the State of California to cause cancer. For more information go to www.P65Warnings.ca.gov.	50g/L	\$26.73
IS5150	Nutrient Agar, 500g	23g/L	\$110.00
IS5151	Nutrient Broth, 500g	8g/L	\$110.00
IS5152	Tryptic Soy Agar, 500g	40g/L	\$125.42
IS5153	Tryptic Soy Broth, 500g	25g/L	\$103.83
IS5154	Luria Agar (Miller's), 500g	40g/L	\$84.82
IS5155	Luria Broth (Miller's), 500g	25g/L	\$74.17
IS5156	Sabouraud-Dextrose Agar, 500g	65g/L	\$74.17
IS5157	Sabouraud-Dextrose Broth, 500g	50g/L	\$69.90
IS5158	EMB(Levine) Agar, 500g	36g/L	\$98.69
IS5159	MacConkey Agar, 500g Marning: This product can expose you to chemicals including Crystal Violet, which is known to the State of California to cause cancer. For more information go to www.P65Warnings.ca.gov.	50g/L	\$103.83

ltem #	Description	Amount Needed per Liter	Price
IS5160	Phenol Red Lactose Broth, 100g	21g/L	\$26.73
IS5161	Phenol Red Lactose Broth, 500g	21g/L	\$104.86
IS5162	Phenol Red Sucrose Broth, 100g	21g/L	\$26.73
IS5163	Phenol Red Sucrose Broth, 500g	21g/L	\$99.72
IS5164	Phenol Red Dextrose Broth, 100g	21g/L	\$27.76
IS5165	Phenol Red Dextrose Broth, 500g	21g/L	\$92.50
IS5166	Starch Agar, 100g	25g/L	\$55.51
IS5167	Starch Agar, 500g	25g/L	\$135.70
IS5168	Simmons Citrate Agar, 100g	24g/L	\$80.18
IS5169	Simmons Citrate Agar, 500g	24g/L	\$111.55
IS5170	SIM Medium, 100g	30g/L	\$42.41
IS5171	SIM Medium, 500g	30g/L	\$105.88
IS5172	Potato Dextrose Agar, 100g	39g/L	\$21.85
IS5173	Potato Dextrose Agar, 500g	39g/L	\$78.13
IS5174	Potato Dextrose Broth, 100g	24g/L	\$17.48
IS5175	Potato Dextrose Broth, 500g	39g/L	\$61.18
IS5176	Mueller-Hinton Agar 100g	38g/L	\$42.41
IS5177	Mueller-Hinton Agar 500g	38g/L	\$104.86

DOT Info: Non-regulated

Prepared Microbiology Media Packs

Our agar and broth media now comes in packs of 12 tubes!

ltem #	Description	Price
IS5360	Tryptic Soy Agar Slant Tubes pk/12	\$35.00
IS5361	Tryptic Soy Broth Tubes pk/12	\$35.00
IS5362	Nutrient Agar Tubes pk/12	\$35.00
IS5363	Nutrient BrothTubes pk/12	\$33.92
IS5364	Luria Agar (Miller's)Tubes pk/12	\$33.92
IS5365	Luria Broth(Miller's) pk/12	\$35.00
IS5366	Sabouraud-Dextrose Agar Tubes pk/12	\$35.00
IS5367	Sabouraud-Dextrose Broth Tubes pk/12	\$35.00
IS5368	Potato Dextrose Agar Tubes pk/12	\$34.93
IS5369	Potato Dextrose Broth Tubes pk/12	\$35.00



DOT Info: Non-regulated



Innovating Science[®] Prepared Microbiology Media

Our agar and broth media comes in bottles and tubes! We are now manufacturing sterile microbiological media for the lab and education market. Our product will bring quality back to the lab market, as we are manufacturing our tubes with the classic black rubber lined phenolic caps to ensure a sterile seal. Great quality and competitive pricing will help you outsource or upgrade your current product line. Let us help you be successful in the microbiology market!

ltem #	Description	Price
MD0100-CS/24	Nutrient Agar, 200mL	\$189.00
MD0101-CS/24	Nutrient Agar, 125mL	\$152.00
MD0102-CS/24	Tryptic Soy Agar, 200mL	\$227.00
MD0103-CS/24	Tryptic Soy Agar, 125mL	\$199.00
MD0104-CS/24	EMB(Levine) Agar, 200mL	\$199.00
MD0105-CS/24	EMB(Levine) Agar, 125mL	\$199.00
MD0106-CS/24	Luria Agar (Miller's), 200mL	\$232.00
MD0107-CS/24	Luria Agar (Miller's), 125mL	\$212.00
MD0108-CS/24	Sabouraud-Dextrose Agar, 200mL	\$217.00
MD0109-CS/24	Sabouraud-Dextrose Agar, 125mL	\$194.00
MD0110-CS/24	MacConkey Agar, 200mL	\$199.00
MD0111-CS/24	MacConkey Agar, 125mL	\$176.00
MD0300-CS/100	Nutrient Agar Tube, Slant, 6mL	\$299.00
MD0301-CS/100	Nutrient Broth Tube, 9mL	\$299.00
MD0302-CS/100	Tryptic Soy Agar Tube, Slant 6mL	\$299.00
MD0303-CS/100	Tryptic Soy Broth Tube, 9mL	\$299.00
MD0306-CS/100	Luria Agar (Miller's) Tube, Slant 6mL	\$299.00
MD0307-CS/100	Luria Broth (Miller's) Tube, 9mL	\$299.00
MD0308-CS/100	Sabouraud-Dextrose Agar Tube, Slant 6mL	\$299.00
MD0309-CS/100	Sabouraud-Dextrose Broth Tube, 9mL	\$299.00

DOT Info: Non-regulated



STEM Investigations: Create Your Own Power

The voltaic cell is an electrochemical cell that produces electrical energy through a chemical reaction, specifically an oxidation-reduction reaction. In this activity, students will set up electrochemical cells, with each group using a different combination of materials for the anode and the cathode of the cell. Students will also compare two different styles of cell, one using a salt bridge and one using a semi-permeable membrane. After determining the voltage output of their cells and comparing their values to the rest of the class, students will be challenged to design an electrochemical cell that can run a small motor. Kit contains enough materials for 15 groups. Teacher's manual and student study guide copymasters are included.

- Kit Includes: 1L Copper Chloride 1.0M 1L Iron Chloride 1.0M 1L Zinc Chloride 1.0M 1L Potassium Chloride 1.0M 125 pieces Tin Foil Strips 125 pieces Aluminum Metal Strips 125 pieces Magnesium Metal Ribbon Strips 15ft Dialysis Tubing 15 Black Connecting Wires with Alligator Clips 60 Pipettes
- 1L Magnesium Chloride 1.0M 1L Tin (II) Chloride 1.0M 1L Aluminum Chloride 1.0M 125 pieces Copper Metal Foil Strips 125 pieces Iron Metal Strips 125 pieces Zinc Strips 60 Cups, 5oz 15 Red Connecting Wires with Alligator Clips 20 pieces Filter Paper 15 3 volt Motors

P65Warnings.ca.gov.



DOT Info:

UN3082, RQ, Environmentally hazardous substance, liquid, n.o.s., (Cupric chloride solution), 9, III, Ltd Qty UN1760, Corrosive liquids, n.o.s., (Stannous chloride, Hydrochloric acid), 8, III, Ltd Qty NA1760, Ferrous chloride solution, 8, III, Ltd Qty UN2581, Aluminum chloride solution, 8, III, Ltd Qty I

IS3450

\$535.00

STEM Investigations: Fingerprint Classification

In this lab activity students will learn about the history of fingerprinting as a tool for identification. They then will analyze fingerprints for common characteristics and create a classification system using those characteristics. Once they have created a system, they will analyze class data using the system that is used worldwide for classification. This kit includes an optional activity where students can practice lifting a fingerprint and comparing it to a "crime scene" print. Kit contains enough materials for 15 groups. Teacher's Guide and Student Study Guide copymasters are included.

Kit includes:

- 15 Hand magnifiers
- 1 Ink pad
- 1 btl Fingerprinting powder
- 15 Fingerprinting brushes
- 30 Acetate sheets
 - Fingerprint Record Sheet

DOT Info: Non-regulated



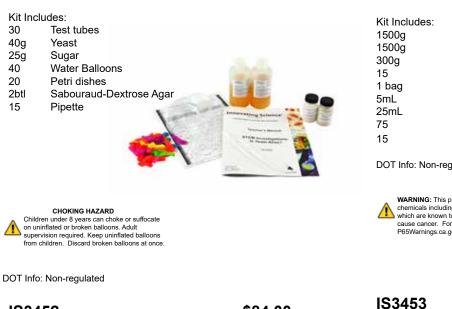
IS3451

\$95.00



STEM Investigations: Is Yeast Alive?

How do scientists know if something is alive? There are certain characteristics that the scientific community uses to determine whether something could be considered living. On first observation, yeast does not look like a typical living thing. In this activity, students will hypothesize about yeast and then test to see if their hypothesis is correct. Using this information they will be able to come to a conclusion to the question of whether yeast is alive. Kit contains enough materials for 15 groups. Teacher's Guide and Student Study Guide copymasters are included.



STEM Investigations: Design a Water Filter

How is the water that comes out of our faucets made safe for consumption? Students will learn about the processes performed at water treatment facilities, and then engineer their own small scale filtration plant. Once they design their filtering system, they will observe and evaluate the quality of the water they produce. Kit contains enough materials for 15 groups of students. Teacher's Manual and Student Study Guide copymasters are included.

Kit Includes:	
1500g	Fine Sand
1500g	Gravel
300g	Activated Charcoal
15	Coffee Filters
1 bag	Cotton Balls
5mL	Green Food Colorir
25mL	White Vinegar
75	Plastic Cups
15	Stirring Sticks

ng

DOT Info: Non-regulated

WARNING: This product can expose you to chemicals including Respirable crystalline silica which are known to the State of California to cause cancer. For more information go to www. P65Warnings.ca.gov.





\$84.00

\$73.99

32-35

STEM Investigations: Make Your Own Liquid Crystal Thermometer

Students explore materials science and engineering while learning about an interesting state of matter: liquid crystals. Liquid crystals have an ordered structure like a solid, but are also fluid like a liquid. The molecular-level structure of

a liquid crystal can be affected by changes in its environment, such as electric field or temperature, which can lead to changes in its macroscopic properties, such as color. In this experiment, students mix three chemicals in different ratios to form four different liquid crystal mixtures that are sensitive to temperature. Working in groups, students will determine the working temperature range for each of the four liquid crystal mixtures. As an engineering challenge, the students will then use the information they have gathered to design and construct a thermometer using their liquid crystals. Through this kit, students will gain an understanding of liquid crystals and how they can be used in a variety of everyday technologies.

Kit Includes: 5g Cholesteryl Oleyl Carbonate 5g Cholesteryl Pelargonate 2q Cholesteryl Benzoate 8 Dram Vials with Caps 90 Clear Labels 8 Plastic Forceps 16 Black Index Card, 4"x 6"

DOT Info: Non-regulated

IS3454 \$208.00

Aldon

www.aldon-chem.com



17-23

17-23

STEM Investigations: Design a Biodegradable Plastic

Hundreds of millions of tons of plastic are produced each year and used for a variety of products, from packaging to automobiles. The majority of these products are disposable items, leading to excessive amounts of plastic being discarded into landfills and the oceans. Due to the major threat this poses to the environment, scientists and engineers are developing bioplastics that break down more quickly and produce less harmful byproducts than traditional plastics. In this activity, students will make samples of different bioplastics to study how their composition affects their physical properties. They will then use what they have learned to create a bioplastic that could be used to replace the PET plastic in disposable water bottles. Additionally, students will design an experiment to compare the biodegradability of their bioplastic samples to a traditional plastic. Kit contains enough materials for 15 groups. Teacher's Manual and Student Study Guide copy masters are included.

Kit Includes: 25mL Glycerin 75mL Vinegar 175g Gelatin 125g Starch 45 Petri dishes 30 Pipettes

DOT Info: Non-regulated

IS3455

\$103.00



Introduction to Materials Science

In this activity, students will be introduced to the field of materials science as they explore the four main types of materials: metals, polymers, ceramics, and composites. They will test the physical and mechanical properties of different material samples, then categorize each material based on their results. Kit contains enough material for 15 groups. Teacher's Guide and Student Study Guide copymasters included.



Separation of a Mixture: Physical Properties of Matter

After a class discussion of matter and the fact that matter may differ from other matter in a variety of ways, students will examine three different solids individually and observe some differences in physical properties. Sand, iron filings and salt will be observed and tested using water and magnets. Based on what they find individually they will come up with a method (as a class or as groups) to separate a mixture of all three. The activity contains enough materials for 6 groups of students.

Kit Includes: 50g Sand 50g Salt 50g iron filings 200g 3 part mixture Magnets Plastic cups Magnifiers

DOT: Non-regulated







WARNING: This product can expose you to chemicals including Respirable crystalline silica, which is known to the State of California to cause cancer. For more information go to www.P65Warnings.ca.gov.

Acids and Bases

Students will learn about acids and bases through a class discussion and then test a dilute acid and dilute base with neutral litmus paper to learn how the paper indicates something is acidic or basic. Once that skill is learned they can move on to test some common items that would be found around their house to determine if they are acids or bases. Optional: Extra litmus paper is included to test more items either from home or around the school building for a greater understanding of acids and bases. The activity contains enough materials for 6 groups of students.

Kit Includes: 30mL 0.01M HCI 30mL 0.01M NaOH 30mL Distilled water 30mL Vinegar 30ml Dilute ammonia 30ml Liquid soap 30mL Lemon juice Neutral litmus paper Spot plates



WARNING: CHOKING HAZARD Small parts. Not for children under 3 yi

IS1001

DOT Info:

Non-regulated



Playing with Polymers

Not for children under 3 vrs

\$61.50

Teacher leads a class discussion about polymers and their importance in our everyday lives (specifically plastics). After the discussion students will be able to make two different polymers – worms and slime. The activity contains enough materials for 6 groups of students.





Elementary Explorations

Innovating Science[®]

Properties of Matter: What is Oobleck?

This kit has been designed to introduce young students (K-4) to the concepts of matter in a fun, safe, hands-on way. Students are given samples of both a solid and a liquid so that they can perform some simple experiments that demonstrate the differences. The students are then given a sample of "Our Friend Ooblek" (a non-Newtonian fluid) to use in the same experiments and are asked to determine if Oobleck is a solid or a liquid. This kit contains everything needed for 30 students to perform the experiments while working in groups of 3.



Seed Germination and Plant Structure

Designed for young elementary students to help them examine and understand the needs of plants for growth and survival. This kit contains materials for 15 groups of two to dissect and germinate seeds, and compare their plant's growth to other types of plants grown in the classroom.



Healthy Plants

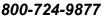
Designed for older elementary students to help them examine and understand the needs of plants for growth and survival. For the sake of time, the class can be split in thirds, with some students conducting a light/dark experiment, some studying plant nutritional needs, and some studying effects of acid rain. Results will be shared at the end. This kit contains materials for 15 groups of two.



IS1005

\$56.00







Soap Power!

Why are raindrops the shape that they are? How do detergents work? Simply put - surface tension. This kit contains materials to demonstrate the concept of surface tension to your classroom using soap as a power source! Kit contains enough materials for 6 groups of 4.



Kit Includes:

Elephant Toothpaste

changes and catalysts.

Check out this fun demonstration that is classroom safe

and shows the effects of a catalyst on a chemical reaction.

Discussion topics can include basic chemical reactions, phase

Teach students about the importance of fossils and

how they form with this fun hands-on activity! This

Kit contains enough materials for 15 groups of

kit simulates the formation of cast-and-mold fossils.

Density: Lava Lamp

What is density? This concept can be difficult to explain to your elementary students. Help them visualize and understand density and introduce the idea of polarity as well with this demonstration for the whole classroom.



Innovating Science[®]

Small Group Learning kits from Innovating Science

These kits cross different curriculums and are designed for smaller class settings, distance learning, after school programs and home schools. Most kits are designed for 5 groups and include a teacher's manual and student guide to complete the lab. These kits provide the same results as our larger traditional classroom kits but allow for situations where a smaller group is desired.

Small Group Learning: **Introduction to Chemical Equations**

Chemical reactions take place all around us. Being able to identify reactions and communicate what is happening is vital to the scientific community. For full kit description see IS2514 on page 107. Kit contains enough materials for 5 groups. Teachers Manual and Student Study Guide copymasters are included.

Kit Includes:

20mL Cobalt Nitrate 0.1M 20ml Cupric Nitrate 0.1M 20mL Ferric Nitrate 0.1M 20mL Nickel Nitrate 0.1M Aluminum Nitrate 0.1M 20ml 4x25mL Sodium Hydroxide 0.2M 2x25ml Hydrochloric Acid 2.0M 2x25mL Silver Nitrate 0.1M 3 Zinc Metal pieces Copper Wire pieces 5 1 pack Wooden Dowels 12 Graduated Pipettes



10mL 0.1 M Sodium Chloride

10mL 0.1 M Sodium Iodide

10mL 0.1 M Sodium Sulfate

10mL 0.1 M Cadmium Nitrate

10mL 0.1 M Chromium Nitrate

5 Acetate Sheets (Preprinted)

10mL 0.1M Silver Nitrate

10mL 0.1 M Cobalt Nitrate

10mL 0.1 M Barium Nitrate

10mL 0.1 M Zinc Nitrate

10mL 0.1 M Sodium Ferrocyanide

10mL 0.1 M Sodium Phosphate

DOT Info:

Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173.4 for domestic highway or rail transport only

WARNING: This product can expose you to chemicals including Nickel (soluble compounds). which are known to the State of California to cause cancer, reproductive harm. For more information go to www.P65Warnings.ca.gov.

IS2514-SGL \$52.75

Small Group Learning: Introduction to Ionic Reactions

This kit is designed to introduce students to the concept of ionic reactions. For full kit description see IS2518 on page 108. This kit has all of the materials needed for 5 groups. Teachers Manual and Student Study Guide copymasters are included.

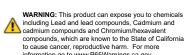
Kit Includes:

10mL 0.1 M Sodium Acetate 10mL 0.1 M Sodium Carbonate 10mL 0.1 M Sodium Hydroxide 10mL 0.1 M Sodium Oxalate 10mL 0.1 M Sodium Silicate 10mL 0.1 M Lead Nitrate 10mL 0.1 M Cupric Nitrate 10mL 0.1 M Strontium Nitrate 10mL 0.1 M Calcium Nitrate 10mL 0.1 M Aluminum Nitrate 10mL 0.1 M Ferric Nitrate 1 box Toothpicks

DOT Info⁻ Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173.4 for domestic highway or rail transport only

IS2518-SGL

\$58.60



information go to www.P65Warnings.ca.gov.

Small Group Learning: A Safer Flame Test: Identification of Metal Lab

The flame test is an analytical technique often used for the identification of certain elements, primarily metal ions. For full kit description see IS2520 on page 109. This kit contains enough material for 5 groups. Teacher's Manual and Student Study Guide copymasters are included.

Kit Includes: 2mL 10% Boric Acid in Methanol 250a Sand **Color Flame Candles** 1 pkg Alcohol Burner 1 5 **Diffraction Slides** 1 Spectroscope 12 Small Trays munin DOT Info Small quantity exemption 173.4 this package conforms to 49 CFR 173.4 for domestic highway or rail transport only WARNING: This product can expose you to chemicals including Methanol and Respirable crystalline silica, which are known to the State of California to cause cancer, reproductive harm. For more information go to www.P65Warnings.ca.gov.

IS2520-SGL

\$55.00

82

www.aldon-chem.com

800-724-9877



Small Group Learning

Innovating Science®

Small Group Learning: Fractional Distillation

In this activity, students will complete a fractional distillation experiment to simulate the fractionation of crude oil. For full kit description see IS2521 on page 109. This kit contains enough material for 5 groups. Teacher's Manual and Student Study Guide copymasters are included.



Small Group Learning: Chromatography of Amino Acids

In this activity, students will perform a paper chromatography experiment on three known amino acids, and then use their results to identify the components of an unknown amino acid mixture. For full kit description see IS2522 on page 109. This kit contains enough material for 5 groups. Teacher's Manual and Student Study Guide copymasters are included.

WARNING: This product can expose you to chemicals including Methanol/Methyl isobutyl Kit Includes: 0.5g Ninhydrin Powder ketone, which is known to the State of California 4x25mL 95% Ethanol Solution to cause cancer, reproductive harm. For more information go to www.P65Warnings.ca.gov 2x25mL Chromatography Solvent 10mL Alanine Control 10ml Glycine Control 10mL Leucine Control 10mL Unknown Amino Acid Mixture 4 **Capillary Tubes** 7 Chromatography Paper DOT Info:

 $${\rm Small}$$ quantity exemption 173.4 this package conforms to 49 CFR 173.4 for domestic highway or rail transport only

IS2522-SGL \$29.95

Small Group Learning: Separation of a Mixture of Solids

Performing liquid-liquid extractions allows students to apply their knowledge of the properties of molecules. Students will learn how the properties of solubility can be utilized to separate two solids in a mixture. For full kit description see IS2523 on page 109. This kit contains enough material for 5 groups. Teacher's Manual and Student Study Guide copymasters are included.

Kit Includes:

Benzoic Acid/Acetanilide 1:1 Mixture 15a Benzoic Acid/Acetanilide 2:1 Mixture 15q 15g Benzoic Acid/Acetanilide 3:1 Mixture 1q Methyl Violet 135mL Ethvl Acetate 75mL Sodium Bicarbonate 100mL Hydrochloric Acid 1.0N 12 Pipettes DOT Info:

DOT Info: UN1173 Ethyl acetate, 3, II, Ltd Qty UN1789 Hydrochoric acid, 8, III, Ltd Qty

IS2523-SGL

Aldon



www.aldon-chem.com

\$38.00

800-724-9877

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Small Group Learning: DNA/Chromosome Staining

Prepare your own squashed stained slide and be able to identify the phases of plant mitosis and chromosomal development. Students will be able prepare, stain and mount slides using specially prepared onion root tips. Each kit contains enough material for 5 groups.

Kit Includes: 1 vial 20mL 20mL 1 box 1 box 5	Preserved Root Tips Hydrochloric Acid, 6M Toluidine Blue 1.0% Solution Microscope Slides Coverslips Aluminum Dishes		Valling Science
5	Forceps	CHAI	Late Activity
DOT Info:		(3)	and the second s
Small quantity ex THIS PACKAGE	cemption 173.3 CONFORMS TO 49CFR 173.4		. 🙆
for domestic high	way or rail transport only		Alteral Card
WARNING:This product can expose you to chemicals including Methanol/Methyl isobutyl ketone, which is known to the State of California to cause cancer, reproductive harm. For more information go to http://www. P65Warnings.ca.gov.		IS3000-SGL	\$36.50

Small Group Learning: DNA Extraction

In this lab you will learn the history of the discovery of DNA and DNA structure. Understand the nature of genetic inheritance and the role of DNA and proteins in genetic expression while using biological detergents, enzymes, and ethanol to isolate DNA from plant material. You need to supply the plant material. There are enough materials for 5 groups.



IS3002-SGL

\$22.00

Small Group Learning: Diffusion and Cell Size

Why are cells microscopic? The answer relates to the need for the cell to effectively bring materials in and remove waste. In this activity, students will create simulated cells (agar blocks) of different sizes and examine how effectively a substance is able to diffuse into the cell in a set period of time. A special indicator in the cells will allow students to visualize the degree of diffusion. The results will clearly display the fact that a smaller volume creates a more favorable condition for the exchange of material across a cell membrane. Kit contains enough materials for 5 groups of students. Teacher's Manual and Student Study Guide copymasters are included.

Kit Includes: 25g Agar 25mL Hydrochloric acid, 2.0M 5 Plastic cups 5 Plastic stirrers

10mL Bromothymol blue concentrate 6 Agar block casting trays 5 Plastic knives

DOT Info: Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173.4 for domestic highway or rail transport only

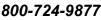
IS3003-SGL \$21.33



to chemicals including Methanol/Methyl bisobutyl ketone, which is known to the State of California to cause cancer and reproductive harm. For more information on to

Aldon

http://www.P65Warnings.ca.gov.



Small Group Learning

Innovating Science®

Small Group Learning: Enzymes and the Process of Digestion

Upon completion of this lab, students will not only understand the importance of the digestive system but also the vital role enzymes play in releasing nutrients from food and converting them to a form usable by the body. For full kit description see IS3009 on page 55. Kit contains enough materials for 5 groups. Teacher's Manual and Student Study Guide copymasters are included.

WARNING: This product can expose you to chemicals including Phenolphthalein, which is known to the State of California to cause cancer. For more information go to www.P65Warnings.ca.gov.

Kit Includes:

Pancreatin, Bile Salts	1g
Starch, Soluble	350
Amylase Bacteriological Pwd	10n
Hydrochloric Acid 0.1M	10n
Phenolphthalein 1% IPA	10n
Olive Oil, Pure	4
	Starch, Soluble Amylase Bacteriological Pwd Hydrochloric Acid 0.1M Phenolphthalein 1% IPA

1g 350mg rd 10mL 10mL 10mL 4

Albumin Egg Pwd Pepsin 1:10,000 Sodium Hydroxide 0.1M Biuret For Protein Test IKI Dilute Lugols Solution Pipette,plastic,graduated

DOT Info: Small quantity exemption 173.4

This package conforms to 49CFR 173.4 for domestic highway or rail transport only

IS3009-SGL

\$35.50

Small Group Learning: Introduction to Mendelian Genetics

In this activity, students can simulate Mendel's work and determine patterns of inheritance using special chips and Innovating Science's exclusive "double dice." For full kit description see IS3010 on page 55. Kit contains enough materials for 5 groups. Teacher's Manual and Student Study Guide copymasters are included.

Kit Includes:

- 5 Monohybrid chips, female (red/yellow)
- 5 Monohybrid chips, male (red/white)
- 5 Dihybrid double dice, female (colored)
- 5 Dihybrid double dice, male (clear)
- 5 Plastic shaker cups



IS3010-SGL



Small Group Learning: Deluxe Owl Pellet Dissection Lab Activity

In this activity, students will dissect an owl pellet and discuss food chains. For full kit description see IS3016 on page 57. This kit contains enough material for 5 groups. Teacher's Manual and Student Study Guide copymasters are included.

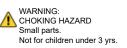
Kit includes: 5 Owl Pellets 10 Wooden probe 5 Forceps 20 Gloves 5 Magnifier 5 Small ruler

5 Smail ruler

DOT Info: Non-regulated

IS3016-SGL

Aldon



\$76.00



WARNING: This product can expose you to chemicals including Lead and lead compounds, which are known to the State of California to cause cancer, reproductive harm. For more information go to www.P65Warnings.ca.gov.

Small Group Learning: Introduction to Microbiology -Bacterial Growth and Staining

Bacteria, good and bad, is all around us. In this activity students will collect and grow bacteria and then learn techniques for studying the bacteria they have grown. They will perform a simple staining technique on bacterial cells to study the morphology. They will also perform a differential staining technique, Gram staining, on bacterial cells and determine if the collected bacteria are Gram-positive, Gram-negative, or both. Kit contains enough materials for 5 groups of students. Teacher's Manual and Student Study Guide copymasters are included.

Kit Includes: 50mL Prepared nutrient agar 10mL Methylene blue 10mL Crystal violet 10mL Gram's iodine 10mL Safranin O

30mL Ethanol, 95% 5 Sterile petri dishes 5 Sterile cotton swabs, pk/2 Microscope slides, pk/72 Coverslips

DOT Info: Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173.4 for domestic highway or rail transport only

WARNING: This product can expose you to chemicals including Crystal Violet and Methanol/Methyl isobutyl ketone, which are known to the State of California to cause cancer and reproductive harm. For more information go to www.P65Warnings.ca.gov.

IS3014-SGL

\$42.50



Small Group Learning: Genetics of Blood Types (Simulated)

In this activity, students will learn about the genetics that determine blood type and the possible inheritance patterns and how they express themselves. For full kit description see IS3102 on page 58. Kit contains enough materials for 5 groups. Teacher's Manual and Student Study Guide copymasters are included.

Kit	Include	s:
	molado	۰.

4	Simulated Blood Samples
	Mother
	Sample X
1 btl	Simulated anti-A serum
1 btl	Simulated anti-B serum
20	Blood typing trays
1pkg	Toothpicks

DOT Info: Non - Regulated



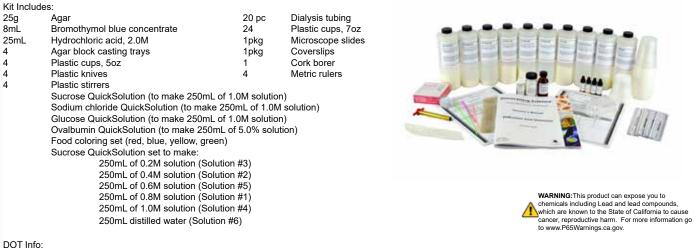
IS3102-SGL

\$30.82

Small Group Learning: Diffusion and Osmosis

Child Sample Y

Students will study the movement of water and nutrients across a cell membrane and observe osmosis in living tissue, along with many other related topics. For full kit description see IS3704 on page 14. This kit contains enough materials for 4 groups. Teacher's manual and Student Study Guide copymasters are included.



Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173.4 for domestic highway or rail transport only

IS3704-SGL

\$95.00

Small Group Learning: Oxidation-Reduction Reactions

Three experiments will be run where a compound, which is colorless in solution when reduced, is converted to a deeply colored solution when oxidized. For full kit description see IS8013 on page 27. This kit contains enough materials for 5 groups. Teacher's manual and Student Study Guide copymasters are included.

Kit Includes: Ferrous ammonium sulfate 4q 30ml Sulfuric acid 6 0M 60mL Potassium thiocyanate 1.0M 10mL Potassium permanganate 10mL Hydrogen peroxide, 3% 100mL Stannous chloride 0.1M 5ml Methylene blue 1% 250ml Potassium hydroxide 1.0M Dextrose 50q

DOT Info: UN1814, Potassium hydroxide, solution, 8, PG II, Ltd Qty UN2796, Sulfuric acid, 8, PG II, Ltd Qty

IS8013-SGL \$





WARNING: This product can expose you to chemicals including Strong inorganic acid mists containing sulfuric acid, which are known to the State of California to cause cancer. For more information go to www.P65Warnings.ca.gov.



Innovating Science[®]

Small Group Learning: The Electrochemical Series

Students will study electrochemical series which is built up by arranging various redox equilibria in order of their standard electrode potentials (redox potentials). For full kit description see IS8030 on page 31. This kit contains enough materials for 5 groups. Teacher's manual and Student Study and Analysis copymasters are included.

Kit Inclu	des:		
5	Copper Metal Strips		
5	Zinc Metal Strips		
5	Magnesium Metal Strips		
5	Iron Metal Strips		
5	Aluminum Metal Strips		innovating Science
8	Filter Paper Strips		Preschar's Manuel
1 ea	EZ-Prep to make 500mL of 0.1M Solution of:		Electrochumical Barles
	Copper Sulfate		1000 No. 100
	Zinc Sulfate		
	Magnesium Sulfate		
	Iron Sulfate		
	Aluminum Sulfate		
			DOT Info: Small quantity examption 172.4
IS8030-SGL		\$49.33	Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173.4

IS8030-SGL

\$49.33

for domestic highway or rail transport only

Small Group Learning: Formation of a Native Copper Mineral Deposit

In this activity, students will create specific environment conditions and examine the formation of native copper deposits over several days. For full kit description see IS8701 on page 82. Kit contains enough materials for 5 groups. Teacher's Guide and Student Study and Analysis copymasters are included.



Small Group Learning: Soil Analysis

Our soil analysis kit will teach students the skills needed to properly conduct soil composition tests as well as allow them to collect and evaluate local soil samples. For full kit description see IS8702 on page 82. Kit contains enough materials for 5 groups. Teacher's Guide and Student Study and Analysis copymasters are included.

5 x 1.5mLUniv	ersal Indicator Solution	5	Magnifiers
5	Universal Indicator Solution Color Chart	5	Forceps
28mL	Bromoform	10	Polarizing filters
28mL	Bromobenzene	25	Drying Dishes
5 x 6mL	Lead Test Reagent	10	Sieve Screen Small
5 x 6mL	0.5 M Hydrochloric Acid	10	Sieve Screen Large
5	Soil Sample A, 12g	1pkg	Glass Slides
5	Soil Sample B, 12g	5	Plastic Scoops
5	Soil Sample C, 12g	5	Soil Sample D, 12g
4	Test Tubes with Caps for Density Column Tubes		

DOT: Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173.4for domestic highway or rail transport only



\$195.30



WARNING: This product can expose you to chemicals including Phenolphthalein/Methanol/ Methyl isobutyl ketone, Bromoform, and Respirable crystalline silica, which are known to the State of California to cause cancer and reproductive harm. For more information go to http://www.P65Warnings.ca.gov.

Innovating Science[®]

Small Group Learning: **Forensic Chemistry of Unknown Substances**

10g

10g 10g

10g

In this experiment, students will use their observation skills, senses, and chemical tests on a series of known substances as well as two unknown substances. For full kit description see IS9006 on page 85. Kit contains enough material for 5 groups. Teacher's Manual and Student Study Guide copymasters are included.

Kit Includes:		
10	Reaction Plates	
10g	Baking Soda	
10g	Plaster	
10g	Gelatin	
10g	Mystery Substance #2	
10mL	Dilute Lugol's Iodine	

Baking Powder Corn Starch Salt Mystery Substance #1 10mL **Biuret Reagent** 10mL Acetic Acid

DOT Info:Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173.4 for domestic highway or rail transport only



IS9006-SGL

\$42.67

Small Group Learning: **Forensic Chemistry of Hair Analysis**

Discover how forensic scientists use hair to assist in solving crimes. For full kit description see IS9007 on page 85. Kit contains enough material for 5 groups. Teacher's Manual and Student Study Guide copymasters are included.

www.aldon-chem.com

Kit Includes:

1 box	Microscope Slides
1 pkg	Coverslips
	Deer Hair Sample
	Cat Hair Sample
4	Human Hair Samples
5	Forceps

DOT Info: Non-Regulated

Small Group Learning: **Properties of Toothpaste**

In this activity, students will investigate several properties of toothpaste samples provided in the kit. For full kit description see IS9350 on page 94. Kit contains enough materials for 5 groups. Teacher's Manual and Student Study Guide copymasters are included.

Kit Includes:	
10mL	Hydrochloric Acid UN1789
50mL	Glycerin
10g	Sodium Lauryl Sulfate UN1325
10g	Gum, Arabic
50g	Calcium Carbonate
3 tubes	Toothpaste
1 Pkg/20	Universal Indicator Strips
1 Pkg/20	Fluoride Test Strips
5	Acetate Sheets
15	Polystyrene Test Tubes
15	Graduated Plastic Pipettes

DOT Info: Small quantity exemption 173.4 This package conforms to 49 CFR 173.4 for domestic highway or rail transport only



800-724-9877

Small Group Learning

Innovating Science[®]

Small Group Learning: Properties of Soaps and Detergents

Students will examine the similarities and differences in the properties of soap, hand dishwashing detergent, and machine dishwashing detergent, all provided in the kit. For full kit description see IS9351 on page 94. Kit contains enough materials for 5 groups. Teacher's Manual and Student Study Guide copymasters are included.

Kit Includes: 10mL 10mL 20mL 20mL 20 20 20 20 20 20 1 Pkg/50	Liquid Soap Dishwashing Detergent (Hand) Dishwashing Detergent (Machine Vegetable Oil Calcium Chloride, 5% Solution Disposable Medicine Cups Polystyrene Test Tubes Capillary Tubes Graduated Plastic Pipettes Universal Indicator Strips)	Kit Includes: 50mL 3 x 30mL 10mL 1 Pkg/18 1 Pkg/5 15 1 Pkg/25 1 Pkg/20 1 Pkg/15 1 Pkg/5	Hydrochloric Acid, 1M Solu Sodium Hydroxide, 1M Sol Phenolphthalein, 1% in Eth Universal Indicator Strips Graduated Plastic Pipettes Disposable Medicine Cups Aspirin, Regular Aspirin, Ruffered Aspirin, Enteric Coated Aspirin, Extra Strength	ution UN1824 anol UN1170
DOT Info: Non-Regulate	ed		DOT Info: Small quantity exemption This package conforms t	173.4 o 49 CFR 173.4 for domestic high	way or rail transport only
			chemicals including A Phenolphthalein/Meth which are known to th	uct can expose you to cetyl Salicylic acid/Aspirin and anol/Methyl isobutyl ketone, e State of California to cause arm. For more information go ca.gov.	
IS9351-SGL		\$46.50	IS9354-SGL	\$4	6.00

Small Group Learning:

Properties of Aspirin

Guide copymasters are included.

Students will examine the solubility of aspirin in varying

ingredients in regular and extra-strength aspirin, and the

differences between regular and buffered aspirin. For full

kit description see IS9354 on page 96. Kit contains enough

materials for 5 groups. Teacher's Manual and Student Study

gastrointestinal environments, a comparison of active

Small Group Learning: Science in the Kitchen

This comprehensive kit incorporates a variety of scientific techniques all themed around common kitchen materials. For full kit description see IS9355 on page 96. Kit contains enough materials for 5 groups. Teacher's Manual and Student Study Guide copymasters are included.

ldon	www.aldon-chem	.com	800-724-9877	89
IS9355-SGL			\$118.00	
DOT Info: Non-Regulated				
Kit Includes:0.5mLFood Color, Red0.5mLFood Color, Green10gBaking Powder10gCorn Starch10mLAcetic Acid, 5% Solution5gPapain1 capsuleAscorbic Acid10mLStarch Indicator Solution50mLDetergent (Machine)10mLCalcium Chloride, 5% Solution5Graduated Plastic Pipettes25Polystyrene Test Tubes1 boxToothpicks	0.5mLFood Color, Blue0.5mLFood Color, Yellow10gBaking Soda10g "Unknown powder"10mLIodine Solution5gGelatin30mLIodine Potassium Iodic50mLLiquid Soap, 5%1 Pkg/50Universal Indicator Str30Disposable Medicine O1 Pkg/4Capillary Tubes5Spot Plates	ps		

www.aldon-chem.com



Small Group Learning: Green Chemistry: The Production of Biodiesel

In this activity, students will be performing a two-phase process to produce small batches of crude biodiesel. For full kit description see IS9500 on page 46. Kit contains enough materials for 5 groups of students. Teacher's Manual and Student Study Guide copymasters are included.

Kit Includes: 2 X 500mL 5 X 25mL 10g 2

Optional:

Vegetable oil Methyl alcohol Potassium hydroxide Microburners

Containers with sealable caps large enough to hold 200mL of liquid (if performing the washing process) Distilled water (if performing the washing process)

DOT Info: Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173.4 for domestic highway or rail transport only



WARNING: This product can expose you to chemicals including Methanol, which is known to the State of California to cause reproductive harm. For more information go to www.P65Warnings.ca.gov.

IS9500-SGL

\$47.29

Small Group Learning: Environmental Chemistry: Water Treatment and Filtration

Students will develop a knowledge of the processes performed at a water treatment plant and understand the reasons for each process. For full kit description see IS9700 on page 49. Kit contains enough materials for 5 groups of students. Teacher's Manual and Student Study Guide copymasters are included.



Small Group Learning: Environmental Chemistry: Nitrates, Phosphates, and Eutrophication

After completing this lab, students will realize that a variety of factors, including natural ones, contribute to the overall problem of water pollution. For full kit description see IS9701 on page 49. Kit contains enough materials for 5 groups. Teacher's Manual and Student Study Guide copymasters are included.

 Kit includes:

 20mL
 Nitrate solution

 20mL
 Phosphate solution

 20mL
 Nitrate/Phosphate solution

 20mL
 Control (deionized) water

 20
 Plastic cups

 1pkg
 Microscope slides

 1pkg
 Coverslips

DOT Info: Non-regulated

IS9701-SGL

\$44.00



90

800-724-9877



Formation of a Native Copper Mineral Deposit

Of all the naturally occurring metals, copper is one of the most useful and versatile. Evidence suggests copper has been used by humans for at least 10,000 years. Copper is used in architecture, electrical wiring, water pipes, and plumbing fixtures. It is found in automobiles and a variety of electronics and electronic devices. It is used in electrical motors, generators, and turbines. It is an important part of heat exchangers found in refrigeration and air conditioning units, and is found on circuit boards and in electromagnets. While copper is typically extracted from copper-containing mineral ores that must be treated and refined to obtain the pure metal, copper also exists in nature in its native elemental form. This is the rarest and most valuable form of copper as it is the easiest to process to obtain pure metal. In this activity, students will create specific environment conditions and examine the formation of native copper deposits over several days. Kit contains enough materials for 15 groups. Teacher's Guide and Student Study and Analysis copymasters are included.

Kit Includes: Copper Sulfate 200a 200g Sodium Chloride 30 Iron nails 15 Plastic vials 50 Filter paper discs, 25mm 15 Hand magnifiers IS8701 \$68.00 DOT Info: Non-regulated

Soil Analysis

Soil composition has been a keystone experiment in earth science classrooms for many years. Our soil analysis kit will teach students the skills needed to properly conduct soil composition tests as well as allow them to collect and evaluate local soil samples. By bringing in local soil samples the students can build a stronger connection to their own local environment while still building on a knowledge base that can be applied to any other environment. Kit contains enough materials for 10 groups. Teacher's Guide and Student Study and Analysis copymasters are included.

10

10

20

50

20

20

10

1pkg

Kit Includes:	
10 x 1.5mL	Universal Indicator Solution
10	Universal Indicator Solution Color Chart
28mL	Bromoform
28mL	Bromobenzene
10 x 6mL 0.5M Lead	Test Reagent
10 x 6mL 0.5 M Hydr	ochloric Acid
10 x 12g	Soil Sample A
10 x 12g	Soil Sample B
10 x 12g	Soil Sample C
10 x 12g	Soil Sample D
-	

DOT: Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173.4 for domestic highway or rail transport only

IS8702

Mohs Hardness

Mohs hardness scale was created by Fredrich Mohs in 1812. Since then, the Mohs hardness scale has assisted geologists in determining the identity of minerals. The activities in this kit will familiarize students with Mohs hardness scale and help them develop vital skills they will use throughout their studies in earth science. This kit also provides an opportunity to take samples from the local environment and investigate the relative hardness of the minerals around them. Kit contains enough materials for 15 groups. Teacher's Guide and Student Study and Analysis copymasters are included.

\$153.17

Kit Includes:

1 bagSample A, 15 Pieces1 bagSample B, 15 Pieces1 bagSample C, 15 Pieces1 bagSample D, 15 Pieces35Pieces of Aluminum Foil35Scratch Plate35Streak Plates

DOT Info: Non-regulated

IS8703



expose you to chemicals including Respirable crystalline silica, which is known to the State of California to cause cancer. For more information go to www.P65Warnings.ca.gov.

WARNING: This product can expose you to chemicals including Phenolophthalein/Methanol/Methyl isobutyl ketone, Bromoform, and Respirable crystalline silica, which are known to the State of California to cause cancer and reproductive harm. For more information go to http://www.P65Warnings.ca.gov.

\$272.42

IS8750

Magnifiers

Polarizing filters

Sieve Screen Small

Sieve Screen Large Glass Slides

Drying Dishes

Plastic Scoops

Forceps

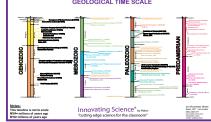
Geology Time Scale Poster

The Innovating Science Geological Time Scale shows major evolutionary events from 4.5 Billion Years Ago to present. Students can visualize the milestones of geologic history. The events are color coded to the Period or Epoch in which they occurred.

\$11.29

Geology Time Scale, Laminated; Measures 11" by 17" IS8751 \$22.60

Geology Time Scale; Measures 21" x 34"



Aldon

Chemical and Mechanical Erosion of Rocks

The rock cycle has been the subject of study for many years. A key component to the rock cycle is the mechanical and chemical erosion of rocks. In this kit students will learn about the rock cycle and the significance of certain types of erosion. This kit also provides an opportunity for students to investigate acid rain and other environmental factors that can be influenced by human actions. Also included is a demonstration that will help students visualize the freezing and thawing action of water on rocks. With multiple activities, this is a kit that is sure to keep students engaged throughout the whole experiment. Kit contains enough materials for 15 groups. Teacher's Guide and Student Study and Analysis copymasters are included.

Kit Includes: Plaster of Paris 350g 30mL 1M Hydrochloric Acid 1 Bag Limestone, 50 Pieces Granite, 50 Pieces 1 Bag 1 Bag Sandstone, 50 Pieces Plastic Cups 4 Plastic Bottles with Caps 50 2 Balloons 20 Disposable pipettes Plastic Trays 50



CHOKING HAZARD

balloons at once.

Children under 8 years can choke or suffocate on uninflated or

broken balloons. Adult supervision

required. Keep uninflated balloons from children. Discard broken

DOT: Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173.4 for domestic highway or rail transport only

WARNING: This product can expose you to chemicals including Respirable crystalline silica, which is known to the State of California to cause cancer. For more information go to www.P65Warnings.ca.gov.

IS8704 \$84.30

Crystal Growing

This kit offers a chance for students to view the formation of colorful and unique shaped crystals. Over the period of a week, students will be able to observe natural formation of crystals. This kit also provides teachers with the opportunity to discuss the formation of crystals and how they play a role in the rock cycle. Students can also investigate different types of crystals that they encounter every day and view the different forms crystals can make. Kit contains enough materials for 15 groups. Teacher's Guide and Student Study and Analysis copymasters



www.aldon-chem.com

IS8705

\$79.25

Exploring Meteorology

Weather impacts everyone, everyday in some way and the meteorologist that study weather patterns are our best source for information. This experiment provides students with the experience of being a meteorologist. It includes studies of pressure changes and creating their own barometer. Studies of wind speed and creating their own anemometer and, the skills to confidently read a weather map. *Bonus* this experiment includes a demonstration of how clouds form by allowing the instructor to form a vapor cloud. Experiment includes enough material for 15 groups of 2 to perform the tasks. The demonstration portion has enough material to perform the demonstration 5 times.

Kit Includes: 10mL Isopropyl alcohol 99% 15 Cork stoppers, small 60 3 oz cups 15 Stir sticks 15 Balloons 15 Rubber bands 15 Push pins

15 Test tubes 15 Bottles 30 5 oz cups 30 Straws 1 Dispenser 1 Glass bottle 15 Practice Weather Maps

DOT Info: Small quantity exemption 173.4 This package conforms to 49 CFR 173.4 for domestic highway or rail transport only

CHOKING HAZARD Children under 8 years can choke or suffocate on uninflated or broken balloons. Adult supervision required Keep uninflated balloons from Iren. Discard broken balloons at once.

\$69.00 IS8706



Porosity and Permeability Lab Activity

While sources of fresh water are usually associated with visible bodies of water such as lakes, rivers, ponds, and streams, the majority of fresh water on Earth is actually underground, collectively referred to as groundwater. Like fresh water found on the surface, groundwater can flow beneath the surface of the planet. Porosity and permeability are two factors that affect both how much groundwater may be found in a given area and the ability of this water to move beneath the surface as well as affect the degree in which certain environmental issues, such as water pollution, may impact different areas. In this activity, students will examine three different substrates and determine the porosity of each. After finding the volume of water each substrate can hold, students will then compare the permeability of the different substrates in order to determine how each may affect the ability of water to move beneath the surface. Kit contains enough materials for 15 groups. Teacher's Manual and Student Study Guide copymasters are included.

Kit Include	s: Marble chips Medium gravel Coarse sand	WARNING: This product can expose you to chemicals including Respirable crystalline silica, which is known to the State of California to cause cancer. For more information go to www.P65Warnings.ca.gov.
60	Plastic cups	
15	Nails	
	Modeling clay	
25mL	Food coloring	
DOT Info:	Non-Regulated	
IS870	7 \$77.00	

800-724-9877

Master set of 12 Forensic kits

IS9000 \$780.00

See individual kit listings for detailed information

Gun Shot Residue Presumptive Test Kit

This is a two part test to determine whether a surface has been exposed to a discharged firearm. A rapid color change takes place to verify the presence of nitrates and lead. Kit contains instructions and enough materials for 30 tests.



Presumptive Blood Test Kit

Test for the presence of blood on materials using phenolphthalein. The test will not distinguish between animal and human blood. Further serology tests are needed. Kit contains instructions, blood standard and reagents to complete 30 tests.



Forensic Chemistry of Document Analysis

The school library's computers have been stolen. Left behind was a ransom note demanding money. Help solve the crime using thin-layer chromatography to separate the ink on the ransom note and ink found in markers tied to possible suspects. It may be possible to provide evidence as to whether or not the ransom note could have been written with a particular marker. Kit contains enough material for 15 groups. Teacher's Manual and Student Study Guide copymasters are included.



Physical Properties of Glass

Often times, during a criminal investigation, police and crime scene investigators must use all available tools and pieces of evidence to work backwards and create the most likely scenario as to what might have occurred. Different types of evidence provide different pieces to the puzzle. Learn about the different chemical and physical characteristics and properties of glass. Find how forensic scientists use these differences to help provide evidence to solve crimes. Kit contains enough material for 15 groups. Teacher's Manual and Student Study Guide copymasters are included.

Kit Includes:			
1 bottle	Borosilicate Glass Beads		
1 bottle	Flint Glass Beads		
1 bottle	Soda-lime Glass Beads		
1 set	Refractive Index Solutions		
15	Magnifiers		

Optional: UV Light Source

DOT Info: Non-Regulated

IS9005

\$135.00

Forensic Chemistry of Unknown Substances

Often times, when collecting evidence at a crime scene, investigators may recover substances they are unable to identify in the field. Along with evidence such as fingerprints, hair, fibers, etc., there may be traces of unknown chemicals or powders left behind by the perpetrator or perpetrators. Evidence of this nature is sent to the crime lab for identification. In this experiment, students will use observation skills, senses, and chemical tests on a series of known substances as well as two unknown substances. They will then attempt to identify the mystery substances based on observations and recorded data. Kit contains enough material for 15 groups. Teacher's Manual and Student Study Guide copymasters are included.

Forensic Chemistry of Hair Analysis

IS9005-REF

Properties of Glass Refill Kit

\$61.00

Discover how forensic scientists use hair to assist in solving crimes. Students will discover the differences between human and animal hair as well as differences among different types of human hair. In the second part of the activity, students will try to determine the origin of a hair sample from a crime scene in relation to hair samples from four known suspects. Kit contains enough material for 15 groups. Teacher's Manual and Student Study Guide copymasters are included.

Kit Includes:

1 box	Microscope Slides
1 pkg	Coverslips
	Deer Hair Sample
	Cat Hair Sample
4	Human Hair Samples
15	Forceps

DOT Info: Non-Regulated

800-724-9877



30 **Reaction Plates** 30g **Baking Powder** 30g Baking Soda 30g Corn Starch 30g Plaster 30g Salt 30g Gelatin 30g Mystery Substance #1 30g Mystery Substance #2 30mL **Biuret Reagent** 30mL Dilute Lugol's lodine Acetic Acid 30ml

DOT Info:Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173.4 for domestic highway or rail transport only

IS9006

Kit Includes:

\$94.58

Aldon

94

Forensic Chemistry of Blood Types

Blood typing is a method of classifying blood based on the presence or absence of specific proteins, called antigens, on the surface of red blood cells. Blood type, an inherited characteristic, is valuable to know in that it affects medical procedures, such as surgery and transfusions, paternity testing, as well as serving as evidence in criminal investigations. Determining blood type can help provide supporting evidence or eliminate a possible suspect's involvement in a crime. Kit contains enough material for 15 groups. Teacher's Manual and Student Study Guide copymasters are included.



IS9008-REF

Refill kit for Forensic Chemistry of Blood Typing **\$27.25**

Contains: Simulated Blood Samples Victim Suspect #1 Suspect #2 Crime Scene ABO/Rh Blood Typing Anti-sera

IS9008 \$60.65 Note: This activity uses Innovating Science Simulated Blood and is safe for classroom use.

Chemiluminescence in Blood Stain Detection

Crime scene investigators are called to the scene of a possible violent crime. They examine the scene for evidence, such as fingerprints, hair, fibers, etc. After collecting the evidence, they notice there is no visible blood. Someone takes out a spray bottle and begins to spray the area with a liquid. After the area is covered with spray, they turn out the lights. A strange, faint glow appears in certain areas of the scene. Learn how luminol is used in scenes like this everyday. The special luminol formulation does not require a separate hydrogen peroxide catalyst. Simply re-hydrate and use with the simulated blood hemoglobin to show your class. This activity includes an Instruction Manual with suggested activities. Kit contains enough material for several demonstrations.



Forensic Chemistry of Drug Detection

Everyone who ate the school cafeteria's chili became ill. Could someone have tainted the chili? You are a forensic toxicologist. It is you and your classmates' task to determine if any of the chili ingredients from the cafeteria could have been substituted with aspirin, which appears to have been stolen from the nurse's office. You will perform a series of chemical tests, including tests on control acetylsalicylic acid, the chemical name of aspirin, in the lab. Kit contains enough material for 15 groups. Teacher's Manual and Student Study Guide copymasters are included.

	1 bottle	Control Acetylsalicylic Acid
	3 bottles	Chili Ingredients
	1 bottle	Ferric Nitrate Solution 0.2M
	1 bottle	Dilute Lugol's lodine
	15	Microreaction Plates
	1 btl	Sodium Hydroxide 1.0N UN1824
		,
		ty exemption 173.4 WARNING: This product can expose you to chemicals including Acetyl Salicylic Acid/Aspirin, which is known to the State of California to
		ye conforms to 49 CFR 173.4 cause reproductive harm. For more information go to www.P65Warnings.ca.gov.
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7		
		Innovating Science
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	\$51.3	4
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Aldon

www.aldon-chem.com

Kit Includes:

Crime Scene Investigation Lab

Use your forensic techniques to solve the crime of the missing frogs from the biology classroom. Four possible suspects have been identified by the authorities. Use fingerprints, hair examination, and chemical analysis of ink by thin-layer chromatography to help determine the most likely culprit. Kit contains enough material for 6 groups. Teacher's Manual and Student Study Guide copymasters are included.

Forensic Chemistry of Dusting for Fingerprints

Learn to identify and classify different types of fingerprints. Students will learn how to identify different types of fingerprints and distinguishing characteristics, as well as dusting for fingerprints, the oldest and most commonly used method of fingerprint detection. Kit contains enough material for 15 groups. Teacher's Manual and Student Study Guide copymasters are included.



Forensic Chemistry: Chemical Detection of Fingerprints

Utilize alternative methods for detecting fingerprints. Examine some possible methods of gathering evidence when dusting for fingerprints is not effective. Learn to identify fingerprint types, a method of fuming for fingerprints, and a technique of chemically-developing fingerprints. The three activities include fingerprint analysis, ninhydrin development, and cyanoacrylate fuming. Kit contains enough material for 15 groups. Teacher's Manual and Student Study Guide copymasters are included.



\$47.29

Blood Spatter Lab Activity

Bloodstain patterns at a crime scene can often yield a wealth of information. Part observational skills, part physics, and part mathematics, bloodstain pattern analysis may be used to help reconstruct as well as assist in supporting or refuting suspect/victim or other eyewitness accounts of the crime. Through a series of stations, students will examine the effects of several factors such as height, angle of impact, surface texture, and velocity before impact as they relate to the physical appearance of bloodstains.Kit contains enough materials for an entire class as well as Teacher's Manual and Student Study Guide copymasters.

200mLSimulated spatter blood100Small index cards, 3" X 5"100Large index cards, 5" X 8"6Rulers3Protractors3Clipboards3Dispensers, 3cc30Tongue depressors3Toothbrushes15Acetate sheets15Cloth squares6Spoons
--



DOT: Non-regulated

IS9014

\$154.00

Forensics of Soil and Mineral Analysis

A crime has been committed and it's up to you to help prove who did it. This kit has been designed to introduce students to the field of forensic soil analysis by providing them with a set of soil and glass samples that have been recovered from a crime scene to compare with a set of samples taken from various suspects. Students will compare samples based on physical observations using polarized, UV and standard light both under ambient conditions and after heating the samples. The properties of the samples will be tested by measuring particle size distribution, material acidity, particle density gradient and mineral composition. This kit contains 10 unique sets of materials that will accommodate up to 30 students when working in groups of 3. Teacher's Manual and Student Study Guide copymasters are included.

Kit Includes:

15mL	Universal Indicator
30mL	Bromobenzene
60mL	Hydrochloric Acid 0.5 M
30g	Soil Sample B
30g	Soil Sample D
10bags	Glass Sample A
10bags	Crime Scene Glass Sample
10	Forceps
20	Polarizing Filters
10	Sieve Screen, Small
72	Glass Slides

DOT Info: Small quantity exemption 173.4 This package conforms to 49 CFR 173.4 for domestic highway or rail transport only

-WARNING: This product can expose you to chemicals including Phenolphthalein/Methanol/Methyl isobutyl ketone, Bromoform, and Respirable crystalline silica, which are known to the State of California to cause cancer and reproductive harm. For more information go to http://www.P65Warnings.ca.gov.

30mL	Bromoform
60mL	Lead Test Reagent
30g	Soil Sample A
30g	Soil Sample C
30g	Crime Scene Soil Sample
10bags	Glass Sample B
10	Magnifying glass
5	Capped Test Tubes
50	Drying Dishes
10	Sieve Screen, Large
10	Plastic Scoops



Blood IS9075 1L \$62.00

Simulated Spatter

Forensics

IS9077 250mL **\$18.99**

DOT: Non-regulated



IS9015 Refill Kit

15mL 30mL 30mL 60mL 30g 30g 30g 30g 30g	Universal Indicator Bromoform Bromobenzene Lead Test Reagent Hydrochloric Acid 0.5 M Soil Sample A Soil Sample B Soil Sample D Crime Scene Soil Sample
30g	Crime Scene Soil Sample

IS9015-Refill \$179.90





\$269.00

Aldon

Forensic Drug and Poison Analysis: Chemistry and Toxicology

In this lab, the student learns the difference between forensic chemistry and forensic toxicology. They gain knowledge in understanding how over-the-counter drugs, controlled drugs, and illegal drugs may act as toxins. Students will have the opportunity to perform presumptive chemical tests on drugs and poisons, identify unknown samples based on chemical test results and examine urine for evidence of heavy metal poisoning using chromatography. Teacher's Manual and Student Study Guide copymasters are included. This kit contains enough material for 15 groups.

Kit Include	es:	
10g	Aspirin	10g
10g	Sodium Bicarbonate Antacid	10g
10g	Unknown #1	10g
30mL	Universal Indicator Solution	30mL
30mL	Ferric Nitrate	1g
10g	Simulated LSD	10g
30mL	Simulated Scott Reagent	30mL
30mL	Simulated Lead Solution	30mL
30mL	Unknown Poison	30mL
30mL	Simulated Urine Sample	30mL
30mL	Glycine Control	1
4 x 25mL	95% Ethanol	15 x 2
15	Universal Indicator Color Charts	6
15pc	Chromatography Sheets	30
8	Scoops	1pkg

g Acetaminophen g Effervescent Antacid g Unknown #2 mL 0.5M Hydrochloric Acid Simulated Marijuana g Simulated Cocaine mL Simulated Van Urk Reagent mL Simulated Mercury Solution mL Simulated Potassium Chromate mL Alanine Control Spray Bottle w/ 0.5g Ninhydrin Powder x 25ML Chromatography Solvent Capillary Tubes Spot Plates kg Toothpicks



WARNING: This product can expose you to chemicals including Phenoiphthalein/Methanoli Methyl isobutyl ketone, which are known to the State of California to cause cancer, reproductive harm. For more information go to www.P65Warnings.ca.gov.

DOT Info: Small quantity exemption 173.4 This package conforms to 49 CFR 173.4 for domestic highway or rail transport only

IS9016 \$204.57

Document Analysis: Comprehensive Lab Activities

This kit has been designed to introduce students to the field of document analysis. Experiments in this kit cover paper analysis, hand writing analysis, detection of a forgery, recovery of mechanical/chemical erasures, hidden message discovery and ink analysis. Students will use microscopy, chromatography and fuming techniques to complete the activities. Kit contains 10 unique sets of materials for 10 groups. Teacher's Manual and Student Study Guide copymasters are included.

Kit Includes: 10 x 20mL 10 x 10mL 1 x 500mL 10 pcs 10 pcs 10 pcs 10 pcs 10 x 10g 10 2 2 2 2	Ink Remover Phenolphthalein Chromatography Solvent Chromatography Paper Paper A Paper C Paper E Transfer Clay Magnifying glass Marker A Marker C #1 Pencil	10 x 5mL 10 x 5mL 1 x 55mL 100 pcs 10 pcs 10 pcs 10 pcs 10 pcs 2 2 2 2	1.0 M Sodium Hydroxide Iodine Solution Paper Slips Paper B Paper D Paper F Forceps Cotton Swabs Marker B Marker D #2 Pencil
10	#3 Pencil	2	Yellow Highlighter
DOT Info: UN1170,Ethanol,3,II,Ltd	t can expose you to		
chemicals including Phe Methyl isobutyl ketone,	enolphthalein/Methanol/		



IS9017

to the State of California to cause cancer, reproductive harm. For more information go to

ww.P65Warnings.ca.gov.

\$184.00

800-724-9877

Forensic Analysis of Fabric and Fibers

This kit has been designed to introduce students to the concepts of fabric and fiber analysis. Experiments in this kit cover the areas of cloth weave identification, microscopic fiber analysis, burn testing and chemical analysis. Students will first analyze a group of known materials to learn the various techniques. They will then be given a crime scene sample and need to test materials taken from four suspects to try and identify who might be involved. Kit contains enough materials for 10 groups. Teachers manual and Student Study guide copymasters are included.

Kit Includes: 4 x 25mL Acetone 4 x 25mL Hydrochloric Acid 1 x 25mL Mounting Medium 1 Pack Microscope Slides 1 Pack Lead Acetate Paper Cotton Fabric Sample 1 Acetate Fabric Sample 1 Polyester Fabric Sample Suspect #1 Sample Suspect #3 Sample 1

4 x 25mL Sodium Hypochlorite (Bleach) 4 x 25mL Sodium Hydroxide 1 Pack pH Strips 2 Packs Microscope Cover Slips 10 Metal Forceps Wool Fabric Sample Polypropylene Fabric Sample Crime Scene Sample Suspect #2 Sample Suspect #4 Sample



DOT Info⁻

Small quantity exemption 173.4 This package conforms to 49 CFR 173.4 for domestic highway or rail transport only

IS9018

\$152.00

chemicals including Ethylbenzene which is know

to the State of California to cause cancer. For nore information go to www.P65Warnings.ca.gov

Who Stole the Gold? A Comprehensive Forensic Analysis

Your forensic team is needed to help figure out which of four suspects is responsible for stealing the gold. Evidence collected from the scene, as well as samples from the four suspects need to be analyzed using techniques from soil and mineral analysis, document analysis, fabric and fiber analysis, blood analysis, fingerprint analysis and hair analysis. This kit contains enough materials for 15 groups. Teacher's Guide and Student Study Guide copymasters included.

1

1

1

Kit Includes:

3 x 3mL	Crime Scene Blood Sample	3 x 3mL	Suspect #1 Blood Sample
3 x 3mL	Suspect #2 Blood Sample	3 x 3mL	Suspect #3 Blood Sample
3 x 3mL	Suspect #4 Blood Sample	3 x 5mL	Simulated Anti-B Serum
3 x 5mL	Simulated Anti-A Serum	3 x 3mL	0.5M Hydrochloric Acid
3 x 2mL	Universal Indicator	24g	Crime Scene Soil Sample
24g	Suspect #1 Soil Sample	24g	Suspect #2 Soil Sample
24g	Suspect #3 Soil Sample	24g	Suspect #4 Soil Sample
3	Universal Indicator Color Charts	5	Plastic Scoops
3	Small Mesh Screens	3	Large Mesh Screens
15	Blood Typing Trays	2 Boxes	Microscope Slides
1 Pack	Microscope Slide Cover Slips	3	Hand Magnifiers
1 Pack	pH Strips	1 Pack	Lead Acetate Paper
3	Metal Forceps	1	Crime Scene Fabric Sample
1	Suspect #1 Fabric Sample	1	Suspect #2 Fabric Sample
1	Suspect #3 Fabric Sample	1	Suspect #4 Fabric Sample
3	Crime Scene Note	3	Suspect #1 Writing Sample
3	Suspect #2 Writing Sample	3	Suspect #3 Writing Sample
3	Suspect #4 Writing Sample	1	Crime Scene Hair Sample
1	Suspect #1 Hair Sample	1	Suspect #2 Hair Sample
1	Suspect #3 Hair Sample	1	Suspect #4 Hair Sample

DOT Info

Small quantity exemption 173.4

WARNING: This product can expose you to chemicals including Respirable crystalline silica and Phenolphthalein/Methanol/Methyl isobutyl ketone, which are known to the State of California to cancer, reproductive harm. For more information go to www.P65Warnings.ca.gov.



IS9019

\$176.00

Forensics Mastery

This kit pack includes the necessary materials to introduce students to the forensic fields of hair analysis, fingerprint analysis, soil and mineral analysis, drug and poison analysis and document analysis. Each kit contains enough material for 30 students when working in groups of 2 to 3. Teacher 's Guide and Student Study Guide copymasters included.

Kit Includes:

- IS9007 Forensic Chemistry of Hair Analysis Kit
- IS9013 Forensic Chemistry: Chemical Detection of Fingerprints Kit
- **IS9015** Forensics of Soil and Mineral Analysis Kit
- IS9016 Forensic Drug and Poison Analysis Lab Activity Kit
- IS9017 Document Analysis: Comprehensive Lab Activities Kit

See individual kit listings for more detailed information

DOT Info: UN1170, Ethanol, 3, II, Ltd Qty

IS9020

\$668.00







Forensic Dental Analysis

Forensic dentistry is defined as the branch of forensic medicine that deals with identification through teeth, dental apparatus and their markings. The activities in this kit will introduce students to the various techniques used in forensic dentistry while they try and identify the suspect in the case. The eight different activities in this kit include the concepts of creating a dental chart, creating a dental casting, creating a bite impression, identification of a patient from dental records, metric analysis of castings and metric analysis of bite marks. This kit includes enough materials for 30 students. Teacher Manual and Student Study guide copymasters are included.



Forensic Toxicology

Your class will utilize different lab techniques to determine the presence of alcohol, illicit drugs and pain medications present in the simulated urine samples of suspected criminals. The various tests include the Marguis Test for morphine, Simon's test for MDMA, THC Immunoassay, Cocaine Immunoassay, Marquis Test for morphine, 6-Monoactylmorphine Test. This kit includes enough materials for 30 students. Teacher Manual and Student Study guide copy masters are included.

Kit Includ	es:			
100mL	Driver urine sample	100mL	Passenger #1 urine sample	
100mL	Passenger #2 urine sample	100mL	Passenger #3 urine sample	
100mL	Patient #1 urine sample	100mL	Patient #2 urine sample	
100mL	Patient #3 urine sample	100mL	Patient #4 urine sample	STARSEN STAR
100mL	Patient #5 urine sample	400mL	Patient #6 urine sample	
400mL	Patient #7 urine sample	400mL	Patient #8 urine sample	
400mL	Patient #9 urine sample	400mL	Patient #10 urine sample	
400mL	Patient #11 urine sample	200mL	Patient #12 urine sample	
200mL	Patient #13 urine sample	200mL	Patient #14 urine sample	
200mL	Patient #15 urine sample 1	20mL	Synthetic cocaine indicator	
30mL	Synthetic ethanol test reagent	60mL	Synthetic marquis reagent	·
30mL	Synthetic Simon's reagent	2mL	Ferric chloride 1.0M	
500mL	Synthetic THC indicator Solution 1	60mL	Synthetic THC indicator Solution 2	
30mL	Synthetic 6-monoacetylmorphine 15	Synthetic	marquis reagent color charts indicator	WARNING: This product can expose you to
				chemicals including Phenolphthalein/Methanol/
DOT Info:	ydrochloric acid, 8, III, Ltd Qty			Methyl isobutyl ketone, which are known to the State of California to cause cancer,
UN1769, H				reproductive harm. For more information go to www.P65Warnings.ca.gov.
IS902	22		\$154.00	

IS9022

Luminol Blend - for Forensic Detection of Blood

To utilize the luminol solution simply add 250 mL of distilled water to the dry luminol reagent. Spray the area of interest with the luminol solution and observe for blood evidence. This special luminol formulation does not require a separate hydrogen peroxide catalyst. Once hydrated the solution is good for up to 2 days if kept refrigerated.

15 g Luminol Reagent Powder

DOT Info: Non-Regulated

IS5040

\$12.50



100



Forensics

Innovating Science®

Forensic Drug Testing: A Simulated Immunoassay

Drug testing comes in two forms, presumptive and confirmatory. Because of the cost associated with confirmatory testing, a presumptive test is often employed first. While there are various forms of presumptive drug testing, two of the most common techniques are immunoassay testing and thin-layer chromatography. In this activity, students will perform a simulated immunoassay test on several urine samples for the detection of the cocaine metabolite benzoylecgonine. All urine samples and testing reagents are simulated to avoid the potential hazards of actual biological specimens while still providing realistic results. Kit contains enough materials for 15 groups. Teacher's Guide and Student Study Guide copymasters are included.

Kit Includes: 5mL Positive Control 5mL Negative Control 5mL Specimen 1 5mL Specimen 2 5mL Specimen 3 5mL Specimen 4 5mL Specimen 5 5mL Specimen 6 30mL Antibody Solution 30mL Labeled Metabolite 15 Spot Plates

ACE 20

DOT Info: Non-regulated

IS9023

\$65.29

Forensic Case Study: Drugs - Small Town, Big Problem

In the small community of Riverside, police have noticed a recent increase in drug-related activity. Under orders of the mayor, the police department was tasked with solving the problem. After extensive investigation the police determined the distributor of the drugs is most likely one of six suspects. The Riverside police have asked the police department from the neighboring town of Mapledale to assist in the investigation. In this activity your students work for the Mapledale Forensic Analysis Lab and have been asked to analyze a variety of forensic evidence, including fingerprints, unknown powders, urine specimens, money, and handwriting samples, to assist the Riverside police in determining which is the most likely suspect. Kit includes enough materials for 15 groups. Teacher's Guide and Student Analysis copymasters are included.

Kit Includes: 25g Suspect 1: White Powder Evidence 25g Suspect 2: White Powder Evidence 25g Suspect 3: White Powder Evidence 25g Suspect 4: White Powder Evidence WARNING: This product can expose you to chemicals including Phenolphthalein and Methanol/Methyl isobutyl ketone, which are known to the State of California 25g Suspect 5: White Powder Evidence 25g Suspect 6: White Powder Evidence 10mL Suspect 1: Urine Sample 10mL Suspect 2: Urine Sample to cause cancer and reproductive harm. For more information go to www.P65Warnings.ca.gov. 10mL Suspect 3: Urine Sample 10mL Suspect 4: Urine Sample 10mL Suspect 5: Urine Sample 10mL Suspect 6: Urine Sample 3 x 10mL Drug Detection Reagent 3 x 10mL Drug Extraction Reagent 3 x 10mL Drug Metabolite Detection Reagent 15 Spot Plates 15 strips Suspect 1: Money Evidence 15 strips Suspect 2: Money Evidence 15 strips Suspect 3: Money Evidence 15 strips Suspect 4: Money Evidence 15 strips Suspect 5: Money Evidence 15 strips Suspect 6: Money Evidence DOT Info: Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173.4 for domestic highway or rail transport only Check IS9024 \$87.000 out our Handy Refill Kit! IS 9024-REF \$50.35 Aldon 101 www.aldon-chem.com 800-724-9877

Forensic Case Study: Murder at Eagle Nest Harbor

A murder in a quiet house boat community sets the stage for you and your students to test their forensic knowledge. Students will take on the role of a forensic technician as they analyze hair, fiber, fingerprint, blood, and hand writing evidence to help pinpoint the killer. This kit provides a unique and complex set of evidence for students to analyze. The kit includes enough evidence and reagents for 15 groups to perform their analysis as well as teacher and student guide master copies. Does your class have what it takes to find the murderer?



Forensic Analysis of Narcotics

Forensic chemists are responsible for analyzing and identifying unknown substances, such as drugs and narcotics, that are found at a crime scene. In this activity, students will learn how forensic chemists can use simple chemical tests and reactions to help determine the identity of an unknown substance. Students will perform a variety of presumptive tests on an unknown substance along with a series of known controls. By comparing the test results of the unknown with those of the controls, the most likely composition of the unknown will be determined. Teacher's Manual and Student Study Guide copymasters are included. Kit contains enough materials for 15 groups.

Kit Includes:

10g	Aspirin
10g	Acetaminophen
10g	Simulated Oxycodone
10g	Simulated Ecstasy
10g	Unknown
30ml	Universal indicator solution
30ml	0.5M Hydrochloric acid
30ml	Ferric nitrate, 0.2 M
15	Universal indicator color charts
30	Spot plates
8	Scoops
1pkg	Toothpicks

DOT Info: Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173.4 for domestic highway or rail transport only

IS9026

\$89.00



Properties of Toothpaste

The history of teeth cleaning agents actually goes back thousands of years. Toothpaste as we know it today however is a comparatively recent development. In this activity, students will investigate several properties of toothpaste samples, provided in the kit, including presence of fluoride, pH, abrasiveness, and foaming ability. As a second activity, students will use the included materials to create their own toothpaste. Kit contains enough materials for fifteen groups. Teacher's Manual and Student Study Guide copymasters are included.

Kit Includes: Hydrochloric Acid UN1789 30mL 100mL Glycerin Sodium Lauryl Sulfate UN1325 25g 25g Gum, Arabic 150g Calcium Carbonate 3 tubes Toothpaste 1 Pkg/50 Universal Indicator Strips 1 Pkg/50 Fluoride Test Strips 1 Pkg/15 Acetate Sheets Polystyrene Test Tubes 45 45

45 Graduated Plastic Pipettes DOT Info: Small quantity exemption 173.4 This package conforms to 49 CFR 173.4

for domestic highway or rail transport only

www.P65Warnings.ca.gov

WARNING: This product can expose you to chemicals including Respirable crystalline silica, which is known to the State of California to cause cancer. For more information or

IS9350



\$115.00

Properties of Soaps and Detergents

Though similar in function, soaps and detergents differ in chemistry and performance. Students will examine the similarities and differences in the properties of soap, hand dishwashing detergent, and machine dishwashing detergent, all provided in the kit. Activities include testing pH, examining the effect of soap and detergent on the surface tension of water, foaming ability, fat emulsification, and the performance of soap and detergent in hard water. Kit contains enough materials for fifteen groups. Teacher's Manual and Student Study Guide copymasters are included.



IS9351

\$79.00

Aldon

Properties of Antacids

With the global population spending over a half a billion dollars a year on commercial antacids, several companies are out there competing for a part of the business. In this activity, students will learn about some of the more common active components in over-the-counter antacids as well as investigate the rate of acid neutralization of five different antacids. Students will also compare the buffering ability of three different calcium carbonate based antacids. All antacid samples are included and the kit contains enough materials for fifteen groups. Teacher's Manual and Student Study Guide copymasters are included.



Properties of Shampoo

Different shampoos are manufactured from a variety of ingredients to perform a variety of tasks. In this investigation, students will not only learn about the role of several shampoo ingredients, but also test several properties of different shampoos which are included in the kit. Students will determine the relative viscosity, pH, flash-foam formation and retention, oil emulsification, and contaminant-dispersion abilities of different shampoos. Kit contains enough materials for fifteen groups. Teacher's Manual and Student Study Guide copymasters are included.

IS9353-REF \$39.99 Refill pack for Properties of Shampoo

Kit Includes:				<u>.</u>
30mL	Shampoo #1			
30mL	Shampoo #2			
30mL	Shampoo #3			
30mL	Ink, Black, 10%			
30mL	Vegetable Oil			
1 Pkg/50	Universal Indicator Strips			
1 Pkg/45	Graduated Plastic Pipettes			
1 Pkg/25	Copper Balls			
45	Disposable Medicine Cups			
45	Polystyrene Test Tubes		Innovating Science	Second States
3	Plastic Centrifuge Tube			And And
	Ū.		Contraction of the Contraction o	
DOT Info:			Proparties of Strampos	
Non-Regulated		and a	in the second se	
Non-Regulated			and the second s	
			2015 X	

IS9353

\$58.75

Consumer Chemistry

Innovating Science®

Properties of Aspirin

Aspirin is to this day the most widely used painkiller and largest selling non-prescription medicine in the world. Learn about the history of the development acetylsalicylic acid (aspirin) and test the performance of several different types of aspirin. Students will examine the solubility of aspirin in varying gastrointestinal environments, a comparison of active ingredients in regular and extra-strength aspirin, and the differences between regular and buffered aspirin. All aspirin samples are included and the kit contains enough materials for fifteen groups. Teacher's Manual and Student Study Guide copymasters are included.

Kit Includes:	
5 x 30mL	Hydrochloric Acid, 1M Solution UN1789
8 x 30mL	Sodium Hydroxide, 1M Solution UN1824
30mL	Phenolphthalein, 1% in Ethanol UN1170
1 Pkg/50	Universal Indicator Strips
1 Pkg/15	Graduated Plastic Pipettes
45	Disposable Medicine Cups
1 Pkg/75	Aspirin, Regular
1 Pkg/60	Aspirin, Buffered
1 Pkg/45	Aspirin, Enteric Coated
1 Pkg/15	Aspirin, Extra Strength
-	

DOT Info: Small quantity exemption 173.4 This package conforms to 49 CFR 173.4 for domestic highway or rail transport only

WARNING: This product can expose you to chemicals including Acetyl Salicylic acid/Aspirin and Phenolphthalein/Methanol/ Methyl isobutyl kelone, which are known to the State of California to cause cancer, reproductive harm. For more information go to www.P65Warnings.ca.gov.



IS9354

\$86.00

IS9355

\$184.00

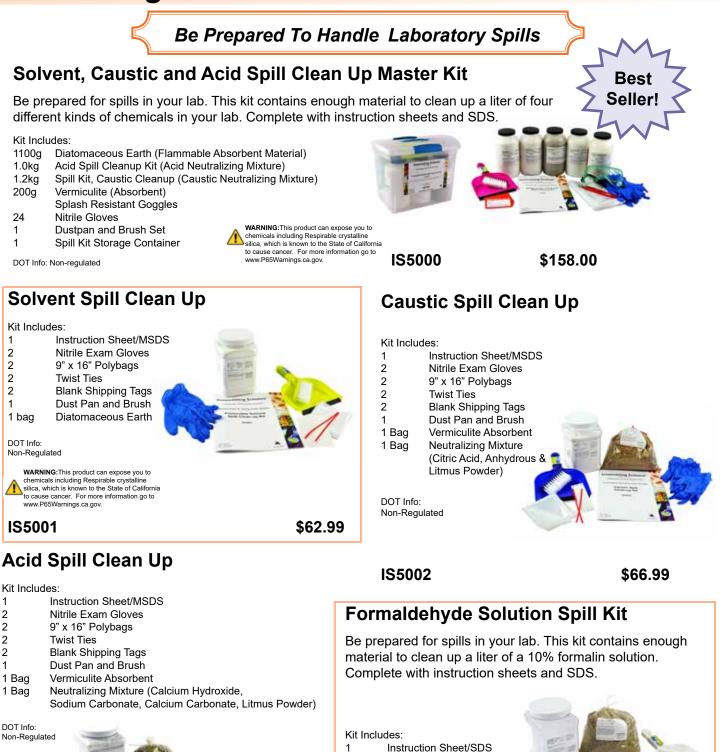
Science in the Kitchen

This comprehensive kit incorporates a variety of scientific techniques all themed around common kitchen materials. Students will perform chromatography on commercial food colors, use chemical tests to identify an unknown cooking ingredient, examine the protein digesting ability of a common meat tenderizing enzyme, use titration to quantify vitamin C levels and then test an unknown juice or soda (not provided), and lastly examine some of the differences and similarities in the materials used to clean up (soap, hand dishwashing detergent, and machine dishwashing detergent). Kit contains enough materials for fifteen groups. Teacher's Manual and Student Study Guide copymasters are included.

Kit Includes: 0.5mL 0.5mL 0.5mL 0.5mL 25q 25g 25g 25g 30mL 30mL 5g 15g 1 capsule 3 x 30mL 30mL 100mL 100mL 100mL 30mL 1 Pkg/50 1 Pkg/15 90 75 1 Pkg/4 1 Pkg/15 15 1 box

Food Color, Red Food Color, Blue Food Color, Green Food Color, Yellow Baking Powder Baking Soda Corn Starch "Unknown" Acetic Acid, 5% Solution **Iodine Solution** Papain Gelatin Ascorbic Acid Iodine Potassium Iodide Starch Indicator Solution Detergent (Hand) 5% Detergent (Machine) Liquid Soap, 5% Calcium Chloride, 5% Solution Universal Indicator Strips **Graduated Plastic Pipettes** Disposable Medicine Cups Polystyrene Test Tubes Capillary Tubes Chromatography Sheets Spot Plates Toothpicks

DOT Info: Non-Regulated



2

2

2

2

1 1 Bag

1kg

IS5035



IS5003

\$66.99

Nitrile Exam Gloves

Blank Shipping Tags Dust Pan and Brush

Vermiculite Absorbent

9" x 16" Polybags

UN3260, Corrosive solid, acidic, inorganic,

n.o.s., (Sodium bisulfate), 8, III, Ltd Qty

Twist Ties

\$65.00

Formaldehyde Neutralizing Agent (Sodium Bisulfite)

www.aldon-chem.com

800-724-9877



Universal Absorbent

Be prepared for spills in your lab. This universal absorbent eliminates the guesswork and is safe for use on all chemicals except hydrofluoric acid. This kit contains enough material to absorb 1 liter of solvent. Complete with instruction sheets and SDS.



chemicals including Respirable crystalline silica, which is known to the State of California to cause cancer. For more information go to www.P65Warnings.ca.gov.

IS5037

\$15.50

Universal Spill Kit

Be prepared for spills in your lab. This universal absorbent eliminates the guesswork and is safe for use on all chemicals except hydrofluoric acid. This kit contains enough material to absorb 10 liters of solvent. Complete with instruction sheets and SDS.



Lab Pillows

Each 4" x 14" x 1" lab pillow absorbs 0.22 gal (750 mL) of solvent. Not to be used on hydrofluoric acid. Pack of 18

DOT Info Non-Regulated

IS5034

\$190.00



Universal Mini Spill Kit

Be prepared for spills in your lab. This universal absorbent eliminates the guesswork and is safe for use on all chemicals except hydrofluoric acid. This kit contains enough material to absorb 1 liter of solvent. Complete with instruction sheets and SDS.



IS5038

\$32.00

Master Spill Kit

All your spill clean up needs in one spot!

- 3 dust pan and brush sets
- Poly bags and twist ties
- Blank tags
- Nitrile gloves
- Splash resistant goggles
- Diatomaceous earth x 2
- Vermiculite absorbent x 2
- Caustic neutralizing mixture x 2
- Acid neutralizing mixture x 2

WARNING: This product can expose you to chemicals including Respirable crystalline silica, which is known to the State of California to cause cancer. For more information go to www.P65Warnings.ca.gov.



DOT Info: Non-Regulated

IS5041 Master Spill Kit \$267.28

Aldon

www.aldon-chem.com

800-724-9877

Vital Stain Kit

Kit Includ	des:	
30mL	Bismarck Brown, 1% Alcohol Solution UN1170	
30mL	Brilliant Cresyl Blue, 1% Alcohol Solution UN1170	
30mL	Cupric Acetate, 3% Aqueous Solution	
30mL	Cupric Sulfate, 1% Aqueous Solution	
30mL	Janus Green, 1% Alcohol Solution UN1170	
30mL	Methylene Blue, 1% Alcohol Solution UN1170	
30mL	Neutral Red, 1% Alcohol Solution UN1170	
This packag	titly exemption 173.4 ge conforms to 49 CFR 173.4 c highway or rail transport only	
	VING: This product can expose you to cals including Methanol/Methyl isobutyl	

ketone and Strong inorganic acid mists containing sulfuric acid, which are known to the State of California to cause cancer, reproductive harm. For more information go to www.P65Warnings.ca.gov.

 $\bullet \bullet \bullet \bullet \bullet \bullet \bullet$

IS5004 \$31.00

Acid Fast Stain Chemicals Kit Includes



ution UN1219
N2924
1

DOT Info: Small quantity exemption 173.4 This package conforms to 49 CFR 173.4 for domestic highway or rail transport only

WARNING: This product can expose you to chemicals including Methanol/Methyl isobutyl ketone which are known to the State of California to cause cancer, reproductive harm. For more information go to www.P65Warnings.ca.gov.

IS5006

\$8.99

Bacteria Stain Chemicals

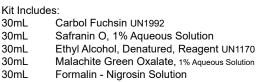
Kit Includes:

The more	465.		
30mL	Methylene Blue Loefflers Solution U	N1987	
30mL	Methylene Blue Saturated, 1% in IP	A/Water UN1219	
30mL	Carbol Fuchsin Solution UN1992		
30mL	Carbol Rose Bengal Solution UN19	92	1.1
30mL	Crystal Violet, 1% Alcohol UN1170	the second se	
This packa	tity exemption 173.4 ge conforms to 49 CFR 173.4 c highway or rail transport only		
A includir ketone	ING:This product can expose you to chemicals g Crystal Violet and Methanol/Methyl isobutyl which are known to the State of California to cause reproductive harm. For more information go to	IS5008	\$17.46

ketone which are known to the State of California to cause cancer, reproductive harm. For more information go to www.P65Warnings.ca.gov.







DOT Info⁻ Small quantity exemption 173.4 This package conforms to 49 CFR 173.4 for domestic highway or rail transport only



Microcrystal Growth Kit

Discover the differences in the crystalline properties of different chemicals. Students will use solutions to create different crystals and study them under a microscope or magnifier (not included).

Kit Inclu	ides:
60mL	Ammonium Phosphate
60mL	Copper Sulfate
60mL	Potassium Chromate
60mL	Potassium Ferricyanide
60mL	Sodium Chloride
60mL	Sodium Nitrate

DOT Info: Non-Regulated

Negative Stain Set

Often used to determine cell size and examine cell arrangement, negative staining stains the background while leaving bacterial cells untouched. The result is clear cells against a dark background. Negative staining does not require the bacterial cells to be heat-fixed an is especially useful for cells that are susceptible to damage during the heat fixation process.

Kit Includes: 30mL Formaline-Nigrosin Solution 30mL 1% Hydrochloric Acid 30mL 1% Congo Red

DOT Info: Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173.4 for domestic highway or rail transport only



WARNING: This product can expose you to chemicals including Formaldehyde, Methanol and Benzidine-based dyes, which are known to the State of California to cause cancer, reproductive harm. For

more information go to www.P65Warnings.ca.gov

IS5014

\$14.34

Gram's Stain

Small Kit Includes: 4 x 25 mL 30mL 30mL 30mL

Ethyl Alcohol, Denatured, 95% UN1170 Crystal Violet Ammonium Oxalate Safranin O, 1% Aqueous Solution Iodine, Potassium Iodide Solution

IS5011 - Small



WARNING: This product can expose you to chemicals including Crystal Violet and Methanol/Methyl isobutyl ketone, which are known to the State of California to cause cancer, reproductive harm. For more information go to www.P65Warnings.ca.gov.



Large Kit Includes:

500mL **Dilute Lugols** 500ml Crystal Violet Safranin O 1% Aqueous 500mL 21 Ethanol 95%, UN1170

IS5017 - Large \$79.16

DOT Info⁻ Small quantity exemption 173.4 This package conforms to 49 CFR 173.4 for domestic highway or rail transport only



Microscope Slide Making Kit

This basic microscope slide-making kit will get you started mounting your own slides.

Kit Includes:

25mL Synthetic Balsam 1 Microscope Slide, Pack/72 1

- Coverslips, Pk/100
- Forceps

DOT Info:

1

Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173 4 for domestic highway or rail transport only



Wide Range pH Water Test Kit

Test water pH within 2-10 range with this wide range pH testing kit. Kit includes enough materials to repeat the procedure over 50 times.

This test kit is designed for educational/ instructional use only.



DOT Info: Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173.4 for domestic highway or rail transport only WARNING: This product can expose you to

Phenolphthalein, Methanol and Methyl isobutyl ketone, which are known to the State of California to cause cancer and birth defects or other reproductive harm. For more information go to www.P65Warnings.ca.gov.

IS5016



Aldon

www.aldon-chem.com

800-724-9877

Complete Stain Kit

Innovating Science's Complete Stain Kit contains everything you need for a variety of staining techniques!

Kit Includes: IS5004 Vital Stain Kit IS5008 Bacteria Stain Chemicals Set IS5011 Gram's Stain Kit

DOT Info: Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173.4 for domestic highway or rail transport only



IS5018

\$78.00

Make Your Own Bromine Water

Prepare fresh bromine water in a couple of minutes! Save on hazardous transport costs and avoid storing elemental bromine in your storage room.

Kit Includes:

2.15gPart 1 Potassium Bromide0.6gPart 2 Potassium Bromate25mLPart 3 Sulfuric Acid 1 M50mLPart 4 Bromine Water-Btl/Label Only

DOT: Small quantity exemption 173.4 This package conforms to 49 CFR 173.4 for domestic highway or rail transport only





Aldon's Innovating Science cereal grass media is used for culturing protozoa (Rhizopods, Choanoflagellates, Ciliates, and Flagellates). It contains dehydrated cereal grass leaves with natural vitamins A,B,C,K. Our cereal grass media is recommended by King's lab at The University of California at Berkeley.

DOT Info: Non-regulated



IS5020	25grams	\$10.26
IS5021	100grams	\$19.03
IS5022	500grams	\$26.22

Microbiology Science Fair Kit

Students can grow their own bacteria with this handson, easy-to-use kit. They will be able to swab various surfaces around the classroom such as door handles and sinks to inoculate agar plates. Students will also investigate the levels and importance of bacteria in everyday situations.

Kit Includes:

- 2g Nutrient Agar (makes 100mL 2% Agar)
- 6 Petri Dishes
- 6 Sterile Swabs

DOT: Non-regulated





\$18.99

IS5030

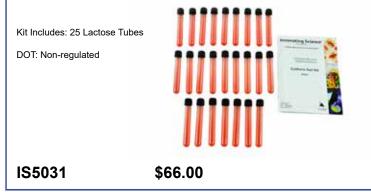
800-724-9877

\$19.51 **Aldon**

110

Qualitative Coliform Test Kit

Coliforms are a broad group bacteria mostly found in the intestines of warm blooded animals. While these organisms are not typically associated with serious illness themselves, they are often used as indicator organisms. The presence of coliforms may indicate a possible presence of other pathogenic or disease-causing organisms. With this easy to use and economical test kit, students simply add a small amount of collected water to a sample tube and incubate for 24-48 hours. A distinct color change indicates that coliforms are present. The kit contains instructions enough tubes to perform 25 tests.



Instant Drosophila Medium, White

Just add water to easily produce the medium, no need to sterilize. ½ liter will make 30-40 standard cultures. Inquire for larger sizes or bulk quotes.

\$6.32

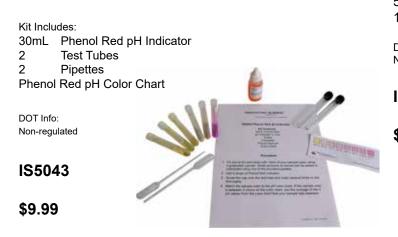
Transpilla Mecks

Phenol Red pH Testing Kit

DOT Info: Non-Regulated

IS5036

Test water pH samples within pH range of 6.8-8.2 with this phenol red pH testing kit. The kit includes phenol red pH indicator, pipettes, reaction vessels, a color chart to compare your results to, and instructions on how to use the kit. Enough materials to repeat the procedure over 50 times!



Acid Rain Science Fair Kit

The Acid rain test kit is great to use in the classroom or for science fair projects to determine if you have acid rain in your area. Collect water samples and use a low range pH indicator to determine the pH of your water sample with a colorimetric comparison chart.



IS5032

\$12.99

pH Buffer Solution

5L of our best selling buffer solutions:

IS5051 5L Buffer pH 4 **IS5052** 5L Buffer pH 7 **IS5053** 5L Buffer pH 10



DOT Info: Non-Regulated

Buffer Calibration Kit

This kit includes 500mL of each buffer solution pH 7, pH 4, and pH 10. It also includes 100mL of electrode storage solution.



Aldon

Coliform Powder Test Kit

Simply add a small amount of collected water to the sample tube and incubate for 24-48 hours. A distinct color change from purple to yellow indicates that coliforms are present. Available as a single test or pack of 5 tests.

Kit Includes: 1 tube Coliform test powder

DOT Info: Non-regulated



IS5055 IS5056 pk/5 \$5.99 \$20.50

Sterile Ringer's Solutions

200mL bottle.

IS5065Ringers, Mammal	\$12.32
IS5066Ringers, Frog	\$12.32
IS5067Ringers, Chicken	\$12.32
DOT Info: Non-Regulated	

Rheoscopic Fluid

Rheoscopic solution is a pearly-white, water-based solution that can be used to demonstrate concepts that are usually difficult to see such as: ocean currents, turbulence, and convection. Adding food coloring to color the solution will make it even easier and more exciting to simulate oceanic and atmospheric patterns.

DOT Info: Non-regulated

IS5060 1 Liter	\$20.54
IS5061	\$12.32
120mL concentrate	makes 7.5 liter

IS5062 \$79.16 Concentrate makes 62.5 liters

WARNING: This product can expose you to chemicals including Ethylene oxide, which is known to the State of California to cause cancer, reproductive harm. For more information go to www.P65Warnings.ca.gov.



Sucrose Solution Set



Red Cabbage Juice pH Indicator

Learn about acids, bases, and pH indicators with our red cabbage juice kit. A pH indicator is a chemical that behaves differently depending on the pH level it is exposed to. Red cabbage juice is a natural pH indicator that changes color from red to purple to yellow-green as the pH of a solution increases. Test out different substances to determine if they are acidic or basic, and see if you can make all six colors found on the color chart!





Artificial Urine

Use simulated urine to identify a variety of physiological conditions with urine test strips.

IS5070 - Urine, Artificial (Control) \$22.60 \$24.65 IS5071 - Urine, Artificial W/Vitamin C **IS5072** - Urine, Artifical W/Phosphates \$24.65 \$24.65 IS5073 - Urine, Artifical W/ Albumin IS5074 - Urine, Artifical W/ Ketone \$24.65 IS5075 - Urine, Artifical W Glucose \$24.65 IS5080 - Urine, Artifical Set Of 4 \$92.00 (5070, 5072, 5073, 5075)

DOT Info: Non-regulated

DOT Info: Non-regulated



Mannitol Salt Agar

A selective and differential medium designed for the growth of Staphylococcus aureus and Staphylococcus epidermidis bacteria. Use 11 grams with 100mL of deionized water.

Note: Media needs to be sterilized with autoclave prior to use.

Instant Snow Powder

Demonstrate the super absorbent properties of some polymers by making snow! As the polymer absorbs water it will expand to over 40 times its original volume, resulting in fluffy flakes of fake snow!

DOT Info: Non-regulated

IS5270 - 100g \$12.71

IS5271 - 400g \$46.26





IS5178 - 100g \$42.41 IS5179 - 500g \$105.88

Ph Up And Ph Down Set For Hydroponics

Innovating Science's pH Up and pH Down solutions are great for hydroponics! Use pH Down to lower the pH level when it is too high, or pH Up to raise the pH level when it is too low. A little solution goes a long way--after measuring your pH, add a small amount of pH Up or pH Down and then measure the pH again, repeat until the pH is at the proper level. Kit contains an instruction sheet and everything you need to test and adjust your pH level.

Aldon		www.aldon-chem.com	800-724-9877
IS5272	\$11.99		and the second s
DOT Info: Non-reo	Julated		The second secon
2	Pipettes		
1 pkg 2	pH Test strips Plastic cups		
250 mL	pH Down solution		
Kit Includes: 250 mL	pH Up solution		

Aldon Dry Erase Board Cleaner and Conditioner

Make dry erase surfaces look new again.

- Just spray and wipe
- Safe cleaner removes shadows, ghosting, and stains
- Can be used on Dry Erase and Magnetic Dry Erase Boards
- 8 oz. Spray Bottle

DOT: Non-regulated

IS5600

Laboratory Cleaner

This cleaner has been specially formulated for use on laboratory surfaces including stainless steel, stone/ composite bench tops and fume hood sashes. This cleaner is intended for removal of stains and residues. It is not intended as a disinfectant.

IS5601 - 4 oz Spray Bottle **\$4.78**

IS5602 - 32 oz Spray Bottle \$11.29

DOT Info: UN1760 Corrosive Liquids, 8, III, Ltd Qty



Aldon Screen Clean and Conditioner

\$5.29

Safe for use on all screens: phones, tablets, computer screens, televisions.

Integrated liquid glass will create the highest level of gloss on your screen while repelling and resisting dust, smudges and fingerprints.

DOT: Non-regulated

IS5605

\$14.18



Obliterase

Use Innovating Science's Obliterase to eliminate RNase, DNase, and DNA contamination.

- Use on glass, plastics, and stainless steel
- · Decontaminate beakers, flasks, pipettes, and tubes
- Residue free
- Phosphate free/biodegradable

DOT Info: Non-regulated

IS5603	32 oz	\$29.95
IS5603-CS/12	32 oz	\$199.00
IS5604	2 oz	\$6.67

Universal Indicator pH Color Charts

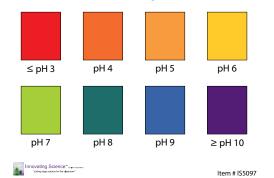
Universal indicator color charts are used to determine the pH of a solution. These handy charts display the colors and pH values an indicator would create for each pH unit from ≤ 3 to ≥ 10 . The charts are available as an overhead transparency, 8.5×11" sheet and smaller chart cards.

IS5097 - Laminated 8½" x11"	\$10.80
IS5098 - 81/2" x11" Overhead Transparency	\$10.80
IS5099 - pack of 30 chart cards	\$5.91

For use with: **IS30000** Universal Indicator Solution 500ml **IS30002** Universal Indicator Solution 25ml

WARNING:This product can expose you to chemicals including Phenolphthalein and Methanol/Methyl isobutyl ketone, which is known to the State of California to cause cancer, birth defects and reproductive harm. For more information go to www. P65Warnings.ca.gov.

Universal Indicator pH Color Chart





114

800-724-9877



Popular Chemical Reagents

Item #	Description	DOT Info	Price		
IS5801	Janus Green - 1% Alcohol Solution	Small Qty Exemption 49CFR173.4	\$4.09		
IS5802	Neutral Red - 1% Alcohol Solution	Small Qty Exemption 49CFR173.4	\$3.99		
IS5803	Bismarck Brown - 1% Alcohol Solution	Small Qty Exemption 49CFR173.4	\$3.99		
IS5804	Brilliant Cresyl Blue - 1% Alcohol Solution	Small Qty Exemption 49CFR173.4	\$3.99		
IS5805	Copper(II) Acetate - 3% Aqueous Solution	Non-regulated	\$3.99		
IS5806	Copper(II) Sulfate - 1% Aqueous Solution	Non-regulated	\$3.99		
IS5807	Methylene Blue - 1% Aqueous Solution	Non-regulated	\$3.99		
IS5808	Methylene Blue, Saturated 1%Alcohol Solution	Small Qty Exemption 49CFR173.4	\$3.99		
IS5809	Carbol Fuchsin Ziehl Neelsen Solution	Small Qty Exemption 49CFR173.4	\$4.35		
IS5810	Carbol Rose Bengal Solution	Small Qty Exemption 49CFR173.4	\$3.99		
IS5811	Methylene Blue (Loeffler's) Solution	Small Qty Exemption 49CFR173.4	\$3.99		
IS5812	Crystal Violet - 1% Alcohol Solution	Small Qty Exemption 49CFR173.4	\$3.99		
IS5813	Safranin O Solution	Non-regulated	\$3.99		
IS5814	Ethyl Alcohol Denatured, 95%	Small Qty Exemption 49CFR173.4	\$3.99		
IS5815	Gram's lodine Solution	Non-regulated	\$3.99		
IS5816	Congo Red - 1% Aqueous Solution	Non-regulated	\$3.99		
IS5817	HCI - 1% In Alcohol	Small Qty Exemption 49CFR173.4	\$3.99		
IS5818	Formalin-Nigrosin Solution	Non-regulated	\$3.99		
IS5819	Malachite Green - 1% Aqueous Solution	Non-regulated	\$3.99		
IS5820	Eosin Y - 1% Solution 30ml	Non-regulated	\$3.99		
IS5821	Eosin B - 1% Solution 30ml	Non-regulated	\$3.99		

Popular Chemical Reagents

Item #	Description	DOT Info	Proposition 65
IS10023	Acetone ACS 12oz	Limited Quantity	
IS10035	Agar, Bacteriological Powder L/G 100g	Non-Regulated	
IS10072	Ammonia Household 500ml	Non-Regulated	
IS11035	Benedict's Qualitative 100ml	Non-Regulated	
IS11053	Biuret Reagent 100ml	Limited Quantity	
IS11070/ IS11072	Bromothymol Blue Sol 0.04% Aq 500ml/100ml	Non-Regulated	
IS11073	Bromothymol Blue Sol 0.5% (Aq) 30ml	Non-Regulated	
IS11104	Buffer Solution Ph 7.00 Yellow 500ml	Non-Regulated	
IS12035	Calcium Chloride Anhyd L/G (4-20 Mesh) 500g	Non-Regulated	
IS12192	Copper (II) Sulfate 5-Hydrate L/G Pwd 500g	Non-Regulated	
IS12196	Copper (II) Sulfate 1.0M 500ml	Non-Regulated	
IS14034	Ethyl Acetate 30ml w/ Dropper	Limited Quantity	
IS14025	Ethyl Alcohol 95% ACS 8oz	Limited Quantity	WARNING: This product can expose you to chemicals including Methanol, which are known to the State of California to cause reproductive harm. For more information go to www.P65Warnings.ca.gov.
IS14018	Ethyl Alcohol 95% Denatured L/G 500ml	Limited Quantity	WARNING: This product can expose you to chemicals including Methanol, which are known to the State of California to cause reproductive harm. For more information go to www.P65Warnings.ca.gov.
IS14021	Ethyl Alcohol 95% Denatured R/G 500ml	Limited Quantity	WARNING: This product can expose you to chemicals including Methanol, which are known to the State of California to cause reproductive harm. For more information go to www.P65Warnings.ca.gov.
IS17013	Hydrochloric Acid 1.0M 500ml	Limited Quantity	
IS17019	Hydrochloric Acid 2.0M 500ml	Limited Quantity	
IS17020	Hydrochloric Acid Sol, 0.1M 1L	Non-Regulated	
IS17046	Hydrogen Peroxide 6% 500ml	Non-Regulated	
IS18019	Iodine Solution (Starch Test) L/G 100ml	Non-Regulated	
IS22062	Methyl Alcohol ACS Acetone Free 500ml	Limited Quantity	WARNING: This product can expose you to chemicals including Methanol, which are known to the State of California to cause reproductive harm. For more information go to www.P65Warnings.ca.gov.
IS22075	Methylene Blue L/G Pwd 25g	Non-Regulated	
IS25043	Phenol Red 0.02% 500ml	Non-Regulated	
IS25139/ IS25138	Potassium Permanganate Reagent 100g/500g	Limited Quantity	
IS28041	Silver Nitrate 0.1M 50ml	Non-Regulated	
IS28078	Sodium Carbonate Anhydrous R/G 500g	Non-Regulated	
IS28113	Sodium Hydroxide Pellet L/G 100g	Limited Quantity	
IS28128/ IS28130	Sodium Hydroxide 0.1M 1L/500ml	Non-Regulated	
IS28121	Sodium Hydroxide 1.0M 1L	Limited Quantity	
IS28256/ IS28255	Sulfuric Acid 1.0M 500ml/1L	Limited Quantity	WARNING: This product can expose you to chemicals including Strong inorganio acid mists containing sulfuric acid, which are known to the State of Californ to cause cancer. For more information go to www.P65Warnings.ca.gov.
IS35002	Zinc Metal Granular (20 Mesh) R/G 100g	Non-Regulated	

General Chemistry

Innovating Science®

Endothermic Reactions

Kit Includos:

When chemical reactions absorb heat the environment around the reaction becomes colder. Students will create an endothermic reaction and monitor the change in temperature as the reaction occurs. Students will then examine a commercial application that exploits endothermic reactions, the instant cold pack. The kit contains enough materials for 15 groups. Teacher's Manual and Student Study Guide copymasters are included.

Kit Includes: 8 pkg x 2 tabs 15 15	Sodium bicarbonate/citric acid tablets Plastic cups Stirring sticks	Teacher's Manual and Student Study Guide copymasters are included.
15 DOT Info: Small quantity exempt This package conform for domestic highway o	s to 49 CFR 173.4 or rail transport only	Kit Includes: 100g Calcium chloride 15 Plastic cups 15 Stirring sticks 15 Bags containing: 25g Iron powder 1g Sodium chloride 5g Calcium chloride 1 tbsp Vermiculite
chemicals including which is known to t	oduct can expose you to J Acetyl Salicylic Acid/Aspirin, he State of California to harm. For more information nings.ca.gov.	DOT Info: Non-Regulated
IS2500 \$46.50		IS2501 \$46.50

Exothermic Reactions

Allow students to see how chemistry can be used in a

beneficial manner. Most students are probably familiar

are guite useful in cold environments. In this activity,

of iron, in a process similar to rusting, can be used to

with heat packs, or hand warmers. Utilizing the release of

chemical energy, these self-contained exothermic reactions

students will first investigate the temperature change of an

exothermic reaction and then examine how the oxidation

create the little bags of heat used to keep hands and feet warm. The kit contains enough materials for 15 groups.

Chemiluminescence

Most physical and chemical reactions that involve the release of light energy also involve the release of heat energy. One unique and interesting form of light-emitting reaction is called chemiluminescence, or "cool light." Similar to the familiar glow of a firefly, students will create a chemiluminescent reaction in the classroom, and then observe the fascinating blue glow of the reaction, which lasts for over ten minutes. This is chemistry that never fails to amaze students. The kit contains enough materials for 15 groups. Teacher's Manual and Student Study Guide copymasters are included.

Kit Includes: 15 x 25mL 25mL 15 15 DOT Info: Non-Regulated	Luminol solution 6% hydrogen peroxide Plastic cups, 30mL Transfer pipettes, 1mL		Incovering Sciences Incovering Sciences Incove
IS2502		\$62.75	Etter Anne

Aldon

Paper Chromatography

Chromatography is the oldest documented technique to separate chemical substances. In this activity, students will perform paper chromatography on three individual dyes and a mixture of dyes to determine if all three dyes are in the mixture. Students will understand not only the components of a chromatography system but also why different substances move at different rates within the system. The kit contains enough materials for 15 groups. Teacher's Manual and Student Study Guide copymasters are included.



Thin Layer Chromatography

Commercial food colors may contain more than meets the eye. Students will perform thin layer chromatography, a highly effective separation procedure, on four different food colors to determine if there may be more in each color than visual appearance would lead them to believe. Each group will perform the chromatography procedure in one of three different solvents and compare their results to other student groups, allowing students to reach conclusions regarding the solubility of each food dye in different solvents. The kit contains enough materials for 15 groups. Teacher's Manual and Student Study Guide copymasters are included.

Kit Includes: Food coloring sample set containing: 1 0.5mL Yellow 0.5mL Red 0.5mL Blue 0.5mL Green 5 x 30mL Chromatography solvent #1 (Deionized water) 5 x 30mL Chromatography solvent #2 (50% Ethanol) UN1170 5 x 30mL Chromatography solvent #3 (95% Ethanol) UN1170 15 TLC (thin layer chromatography) sheets 4 Capillary pipettes WARNING: This product can expose DOT Info: you to chemicals including Methanol/ Methyl isobutyl ketone, which is known to the State of California to cause cancer, Small quantity exemption 173.4 This package conforms to 49 CFR 173.4 for domestic highway or rail transport only reproductive harm. For more information go to www.P65Warnings.ca.gov IS2504 \$98.00

Properties of Polymers

Polymers affect every aspect of our daily lives. From the plastics we use to the clothes we wear, polymer chemistry is everywhere. In this kit, students will perform three fun and fascinating activities involving polymers. In the first two activities, students will perform a crosslinking procedure on polymers to create polymer "worms" and the classic "slime." In the third activity, students will investigate the properties of a super-absorbent polymer, capable of absorbing hundreds of times its own weight in water, to create instant "polymer snow." The kit contains enough materials for 15 groups. Teacher's Manual and Student Study Guide copymasters are included.

Kit Includes:

4g	Sodium alginate (Alginic acid)
30mL	Calcium chloride concentrate
4 x 250mL	3% polyvinyl alcohol
200mL	4% sodium borate
35g	Sodium polyacrylate powder (snow)
45	Clear plastic cups
15	Graduated measuring cups, 30mL
15	Plastic pipettes, 1mL

DOT Info: Non-Regulated



\$64.76



800-724-9877



Acids, Bases, and the pH Scale

In this lab, you will gain an understanding of the basic differences between the properties of acids and bases, learn the role of hydrogen and hydroxide ions in acids and bases, and comprehend the nature of the pH scale with regards to acid and base strength. Students will examine the effects of acids and bases on several chemical pH indicators, determine the pH of several common household materials, and use the knowledge gained to determine the composition of four unknown clear solutions. Kit contains enough materials for 15 groups. Teacher's Manual and Student Study Guide copymasters are included.

25ml

25mL

25mL

25mL

25mL

25mL 25mL

15

2

Soap solution

Filtered water

Reaction trays

Vitamin C solution

Wide-range pH test strips, pkg/50

Unknown solution #4 (Dilute Acid)

Unknown solution #3 (Phenolphthalein)

Unknown solution #1 (Water) Unknown solution #2 (Dilute base)

Kit Includes:			
25mL	Deionized water		
25mL	Dilute acid (0.1M HCI)		
25mL	Dilute base (0.1M NaOH)		
25mL	0.5% litmus		
25mL	0.02% methyl red		
25mL	0.5% bromothymol blue		
25mL	1.0% phenolphthalein		
25mL	Vinegar		
25mL	Household ammonia		

DOT Info:

Small quantity exemption 173.4 This package conforms to 49 CFR 173.4 for domestic highway or rail transport only

WARNING: This product can expose you to chemicals including Phenolphthalein, which is known to the State of California to cause cancer. For more information or to www.P65Warnings.ca.gov.

IS2507

Chemical Identification of Biomolecules

Understand the importance of proteins, carbohydrates, and lipids in living organisms. Students will learn to identify a positive test result for proteins using biuret reagent, examine the reaction between Benedict's reagent and a simple sugar, use iodine/ potassium iodide to test for the presence of starch, and test for the presence of lipids using a fat-soluble dye. Kit contains enough materials for 15 groups. Teacher's Manual and Student Study Guide copymasters are included.



Chemical Element Observation Set

Explore the Periodic Table of Elements with the Chemical Element Observation Set. Students will study the similarities and differences between these elements. Each element is packaged in a break resistant plastic container and clearly labeled. A periodic table is included for review.

Kit Includes:

TAL III	ciuues.			
1.0g	Aluminum Metal	1.0g	Iron	
1.0g	Antimony Metal	1.0g	Lead	
0.5g	Barium Metal	1.0g	Magnesium Metal	
1.0g	Bismuth Metal	1.0g	Manganese	
1.0g	Cadmium Metal	1.0g	Nickel Metal	
1.0g	Calcium Metal	1.0g	Silicon Metal	
1.0g	Carbon (Charcoal)	1.0g	Silver	
1.0g	Chromium Metal	1.0g	Sulfur	
1.0g	Cobalt Metal	1.0g	Tin Metal	
1.0g	Copper Metal	1.0g	Tungsten Metal	
1.0g	Carbon (Graphite)	1.0g	Zinc Metal	
0.1g	Germanium			
	quantity exemption 173	4		
	ackage conforms to 49		34	
	mestic highway or rail tra			
	0 ,	•	2	
	WARNING: This product can including Cadmium and cad			
<u> </u>	hexavalent compounds, Col			
	lead compounds, Respirable			
	metallic, which are known to cause cancer, reproductive			
cause cancer, reproductive harm. For more information go to www.P65Warnings.ca.gov.				

IS2510

\$129.95



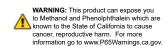
Introduction to Chemical Properties

\$95.00

How do scientists identify an unknown solution? What makes one solution different from another? The experiments in this kit will help students understand the impact different chemicals have when added to a solution. Students will test and identify unknown solutions based on a comparison of the chemical properties to a known sample. Kit contains enough materials for 10 groups. Teacher's manual and Student Study Guide copymasters are included.

Kit Includes: 10 x 3mL 0.1M Cupric Sulfate 10 x 3mL 1.0N Acetic Acid 10 x 3mL 0.1M Sodium Bicarbonate 10 x 3mL 0.1M Sodium Carbonate 10 x 1.5mL Unknown A 10 x 1.5mL Unknown C 10 Preprinted Acetate Sheets

Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173.4 for domestic highway or rail transport only



10 x 3mL 1.0N Ammonium Hydroxide 10 x 3mL 0.1N Sodium Hydroxide 10 x 3mL 0.1N Hydrochloric Acid 10 x 3mL Phenolphthalein Indicator 10 x 1.5mL Unknown B 10 x 1.5mL Unknown D



\$187.00 IS2511

Aldon

Introduction to Reaction Rates

Chemical reactions vary greatly. Some chemicals react extremely fast like an explosion, while others, like rusting metal, can react over a long period of time. Understanding factors that influence the rate of a reaction allows a scientist to speed up or slow down reactions as desired. This kit has been designed to introduce students to the concepts of reaction rates through a series of experiments demonstrating the impact of concentration, heat, surface area and catalysts. Kit contains enough materials for 10 groups. Teacher's manual and Student Study Guide copymasters are included.

Kit Includes:

500ml 0.01M Sodium Hydroxide 10 x 1mL Phenolphthalein Indicator 300mL Iodine Clock Reagent A 400mL Iodine Clock Reagent B 0.15M Sodium Thiosulfate 1 L 300mL 2.0M Hydrochloric Acid Copper Metal Powder 1g 20 Effervescent Tablets 20 Aspirin Tablets 80 Reaction Vessels 1 Pack Wooden Dowels



DOT Info:

UN1789, Hydrochloric acid, 8, III, Ltd Qty

WARNING: This product can expose you to chemicals including Phenolphthalein/Methanol and Acetly Salicylic Acid/Aspirin, which are known to the State of California to cause cancer, reproductive harm. For more information go to www.P65Warnings.ca.gov.

IS2512 \$144.95

Classification of Chemicals: Elements, Compounds and Mixtures

The three major classes of matter are elements, compounds and mixtures. Students will do three lab activities to help understand the differences between elements, compounds and mixtures. Kit contains enough materials for 15 groups. Teacher's manual and Student Study Guide copymasters are included.



Introduction to Chemical Equations

Chemical reactions take place all around us. Being able to identify reactions and communicate what is happening is vital to the scientific community. This kit will introduce two common types of chemical reactions: single replacement and double replacement. Students will perform a series of experiments highlighting the differences in these reactions and use what they learn to write balanced chemical reaction equations. Kit contains enough materials for 15 groups. Teacher's manual and Student Study Guide copymasters are included.

Kit	Includes:	

	KIL IIICIUU	65.		
	5 x 12mL	Cobalt Nitrate 0.1M	5 x 12mL	Cupric Nitrate 0.1M
	5 x 12mL	Ferric Nitrate 0.1M	5 x 12mL	Nickel Nitrate 0.1M
	5 x 12mL	Aluminum Nitrate 0.1M	5 x 60mL	Sodium Hydroxide 0.2M
5 x 30mL Hydrochloric Acid 2.0M		Hydrochloric Acid 2.0M	5 x 30mL	Silver Nitrate 0.1M
	6g	Zinc Metal	15	Copper Wire pieces
	1 pack	Wooden Dowels	32	Graduated Pipettes
	•			
	DOTIC			

DOT Info:

IS2514

UN1824, Sodium hydroxide solution, 8, III, Ltd Qty UN1789, Hydrochloric acid, 8, PGIII, Ltd Qty

WARNING: This product can expose you to chemicals including Nickel (soluble compounds), which are known to the State of California to cause cancer, reproductive harm. For more information go to www.P65Warnings.ca.gov.



Chemical Analysis Using Titrations

A major challenge for any scientist is identifying what and how much of something is in a solution. One of the most common measurement techniques used is titration. The activities in this kit will focus on how acid-base titrations work and how to identify the concentration of an unknown acid. Kit contains enough materials for 10 groups. Teacher's manual and Student Study Guide copymasters are included.

ill do three fferences ixtures. roups. Guide	Kit Includes: 10 x 10mL 10 x 2mL 10 x 1.5mL 10 x 1mL 10 10	0.01M Sodium Hydroxide 0.1N Hydrochloric Acid Phenolphthalein Indicator Universal Indicator Universal Indicator Color Charts Reaction Vessels
	 DOT Info: Malt quantity exemption 173.4 TAS PACKAGE CONFORMS TO 49 CFR 173.4 or domestic highway or rail transport on to Marking: This product can expose you Marking: This product can expose you State of California to cause information go to www.P65Warnings.ca.gor Marking: State of California to cause State State of California to cause State State of California to cause State of California to cause State of California to cause State of the state of the s	
uct can expose you I Crystal Violet and		



Element Groups

The periodic table is more than just a list of elements. The activities in this kit focus on developing a better understanding of why periodic trends occur within the elements. Students will perform two sets of experiments focusing on the alkaline earth metals and the halogens groups. Kit contains enough material for 8 groups. Teacher's Manual and Student Study Guide copymasters are included.

Kit Includes:	
2 x 20mL 0.5M Sulfuric Acid	8 x 3.5mL lodine Water
8 x 3.5mL Chlorine Water	8 x 15mL Mineral Oil
8 x 3mL 1.0M Potassium Bromide	8 x 3mL 1.0M Potassium Chloride
8 x 10mL 0.2M Potassium Iodate	8 x 10mL 1.0M Magnesium Nitrate
8 x 10mL 1.0M Calcium Nitrate	8 x 10mL 1.0M Strontium Nitrate
8 x 10mL 0.2M Barium Nitrate	8 x 10mL 1.0M Sulfuric Acid
8 x 10mL 1.0M Sodium Carbonate	8 x 10mL 0.25M Ammonium Oxalate
8 x 3mL 1.0M Potassium lodide	0.5g Potassium Bromate
1.72g Potassium Bromide	120 Graduated Plastic Pipette
8 Vials for Bromine Water	8 Labels for Bromine Water
DOT Info:	
Small quantity exemption 173.4	
This package conforms to 49 CFR 173.4 for de	omestic
highway or rail transport only	and a loss of the state

WARNING: This product can expose you to chemicals including Strong inorganic acid mists containing sulfuric acid and Potassium Bromate, which are known to the State of California to cause cancer. For more information go to www.P65Warnings.ca.gov

IS2516 \$169.00

Physical and Chemical Properties of Matter

The activities in this kit focus on developing a better understanding of what is matter and what the different types of properties are. Students perform a series of experiments focusing on the physical properties of density, smell, color, melting point, boiling point and solubility. Other experiments focus on chemical properties: students perform a series of reactions as they try and identify an unknown solution. Kit contains enough material for 10 groups. Teacher's Manual and Student Study Guide copymasters are included.



Introduction to Ionic Reactions

This kit is designed to introduce students to the concept of ionic reactions. Students will test a series of chemical combinations and observe reactions. The contents of this kit have been designed to minimize cross contamination and waste generation, while streamlining teacher preparation. This kit has all of the materials needed for 15 groups. Teachers Manual and Student Study Guide copymasters are included.

Kit Includes:
3 x 10mL 0.1 M Sodium Acetate
3 x 10mL 0.1 M Sodium Chloride
3 x 10mL 0.1 M Sodium Carbonate
3 x 10mL 0.1 M Sodium Ferrocyanide
3 x 10mL 0.1 M Sodium Hydroxide
3 x 10mL 0.1 M Sodium Iodide
3 x 10mL 0.1 M Sodium Oxalate
3 x 10mL 0.1 M Sodium Phosphate
3 x 10mL 0.1 M Sodium Silicate
3 x 10mL 0.1 M Sodium Sulfate
3 x 10mL 0.1 M Lead Nitrate
3 x 10mL 0.1M Silver Nitrate
3 x 10mL 0.1 M Cupric Nitrate
3 x 10mL 0.1 M Cadmium Nitrate
3 x 10mL 0.1 M Strontium Nitrate
3 x 10mL 0.1 M Cobalt Nitrate
3 x 10mL 0.1 M Calcium Nitrate
3 x 10mL 0.1 M Zinc Nitrate
3 x 10mL 0.1 M Aluminum Nitrate
3 x 10mL 0.1 M Chromium Nitrate
3 x 10mL 0.1 M Ferric Nitrate
3 x 10mL 0.1 M Barium Nitrate
1 box Toothpicks
15 Acetate Sheets (Preprinted)

DOT Info⁻ Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173.4 for domestic highway or rail transport only



General Chemistry

WARNING: This product can expose you to chemicals including Lead and lead compounds, Cadmium and cadmium compounds and Chromium/hexavalent compounds, which are known to the State of California to cause cancer reproductive harm. For more information go to www.P65Warnings.ca.gov.

IS2518

\$180.00

Molarity Lab Investigation

The experiments in this kit have been designed to introduce students to the concept of molarity. Students will first try to make a solution with a specific molarity to demonstrate the importance of good laboratory technique. In the second set of experiments students will perform a titration on a known acid solution before using the concepts they have learned to identify the concentrations of three unknown solutions. This kit has enough materials for 15 groups. Teacher's Manual and Student Study Guide copymasters included.

Kit Include 400mL 3x15mL 5x 1L 2 x 600g 400mL 400mL 400mL	es: 0.3 M Hydrochloric Acid Phenolphthalein Solution 0.1 M Sodium Hydroxide Sucrose Unknown Hydrochloric Acid #1 Unknown Hydrochloric Acid #2 Unknown Hydrochloric Acid #3	
	ydrochloric Acid 8, III, Ltd Qty opropanol solution, 3, II, Ltd Qty	
IS251	9 \$89.99	WARNING: This product can expose you to chemicals including Phenolphthalein, which is known to the State of California to cause cancer. For more information go to www.P65Warnings.ca.gov.



A Safer Flame Test: Identification of Metal Lab

The flame test is an analytical technique often used for the identification of certain elements, primarily metal ions. The color of the flame is observed and the spectra of light emitted from the flame is viewed through a spectroscope. Traditionally, this test poses safety concerns for students in a laboratory setting. This kit minimizes these concerns by utilizing small candles that are easier to manage in a lab setting. There is no loss in the clarity of the emission spectra from the candles, making it easy for students to observe and deduce what element is causing the colored flame. This kit contains enough material for 15 groups. Teacher's Manual and Student Study Guide copymasters are included.

Kit Includes: 5mL 10% Boric Acid in Methanol 500g Sand 2 pkg Color Flame Candles 2 Alcohol Burners 15 Diffraction Slides 1 Spectroscope 24 Small Trays

DOT Info: Small quantity exemption 173.4 this package conforms to 49 CFR 173.4 for domestic highway or rail transport only

IS2520

California to cause cancer, reproductive harm. For more information go to www.P65Warnings.ca.gov. \$106.00

chemicals including Methanol and Respirable crystalline silica, which are known to the State of

WARNING: This product can expose you to

Fractional Distillation

Crude oil is a complex mixture of many useful hydrocarbons such as gasoline, kerosene, butane, and motor oil. In order to be of use, these hydrocarbons must first be separated from each other. In this activity, students will complete a fractional distillation experiment to simulate the fractionation of crude oil. Topics such as boiling and condensation points, conservation of matter, and forces between molecules are covered as they factor into how a distillation is performed. This kit contains enough material for 15 groups. Teacher's Manual and Student Study Guide copymasters are included.



Chromatography of Amino Acids

Amino acids are the fundamental building blocks of all proteins. Often times scientists may need to identify the amino acid contained in a protein. One such method for separating and identifying individual amino acids is paper chromatography. In this activity, students will perform a paper chromatography experiment on three known amino acids, and then use their results to identify the components of an unknown amino acid mixture. This kit contains enough material for 15 groups. Teacher's Manual and Student Study Guide copymasters are included.

	ides: Ninhydrin Powder 95% Ethanol Solution Chromatography Solven Alanine Control Glycine Control Leucine Control Unknown Amino Acid Miz Capillary Tubes	
20	Chromatography Paper	Methyl isobutyl ketone, which is known to the State of California to cause cancer, reproductive harm. For more information
this packa	ntity exemption 173.4 ge conforms to 49 CFR 173.4 for d r rail transport only	go to www.P65Warnings.ca.gov. Iomestic

IS2522

\$52.00

Separation of a Mixture of Solids

Performing liquid-liquid extractions allows students to apply their knowledge of the properties of molecules. Students will learn how the properties of solubility can be utilized to separate two solids in a mixture. This kit provides for a qualitative test that will help students visualize that two different solids were collected, despite their similar appearance. Students will also explore the concept of the law of conservation of mass as well as how acid-base reactions can be utilized to manipulate molecules. This kit contains enough material for 15 groups. Teacher's Manual and Student Study Guide copymasters are included.

Kit Includes: 35g Benzoic Acid/Acetanilide 1:1 Mixture 35g Benzoic Acid/Acetanilide 2:1 Mixture Benzoic Acid/Acetanilide 3:1 Mixture 35g 2g Methyl Violet 400mL Ethyl Acetate 200mL Sodium Bicarbonate 250mL Hydrochloric Acid 1.0N 35 Pipettes DOT Info: UN1173 Ethyl acetate 3 II I td Otv UN1789 Hydrochoric acid, 8, III, Ltd Qty

IS2523

\$56.00

122

800-724-9877



Periodic Table -Nonmetals, Metals and Metalloids

During the nineteenth century, a Russian chemist Dmitri Mendeleev, began constructing a table of the elements. The work of Mendeleev allows us to understand that the periodic table of the elements is an organized classification of chemical elements based on certain properties of each element in relation to other elements. In this lab students will learn about three different categories of elements metals, nonmetals, and metalloids. They will examine the physical properties of several elements, test the chemical reactivity of each element and classify each of the tested elements as either metal, nonmetal, or metalloid. This kit contains enough material for 15 groups. Teacher's Manual and Student Study Guide copymasters are included.

25g

25g

105

15

45pcs

Kit Includes

IS2524

The moluae	
25g	Aluminum Metal Shot
25g	Silicon Metal Lumps
25g	Sulfur Lumps
45 pcs	Magnesium Metal 0.5"
4 x 30ml	Copper (II) Chloride
30	Pipettes

DOT Info[.] Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173.4 for domestic highway or rail transport only

\$70.17

Carbon Activated Pellets Zinc Shot Copper Metal Foil 0.5" 4 x 30ml Hvdrochloric Acid 1.0N Test Tubes Nails



Exploring the Solubility Rules: A Guided Inquiry

Take your students on an investigation to determine some of the basic rules of solubility. This experiment challenges students to identify patterns within the experiment and then apply that knowledge in a practical exercise. With this new twist to an old experiment, learners are exposed to a new experience while still attaining the same critical information.

Kit Includes: 20ml Ferrous Chloride 0 1M Solution

ZUML	Ferrous Chloride U. 1M Solution	
20mL	Potassium Iodide 0.1M/0.1N Solution	
20mL	Sodium Bromide 0.1M Solution	
20mL	Calcium Chloride 0.1M Solution	
20mL	Sodium Carbonate 0.1M/0.2N Solution	
20mL	Potassium Phosphate 0.1M Solution	
20mL	Calcium Acetate 0.1M Solution	
20mL	Sodium Silicate 0.1M Solution	
20mL	Aluminum Sulfate 0.1M Solution	
20mL	Silver Nitrate 0.1M Solution	
20mL	Zinc Sulfate 0.1M Solution	
20mL	Potassium Hydroxide 0.1N Solution	
20mL	Unknown 0.1M Solution	
210	Pipettes	
45	Trays 🚽 🕻 🚽	
	310	
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DOT Info:		
Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173.4		
for domestic highway or rail transport only		

IS2525

\$99.99

Equilibrium and Le Chatelier's Principle

Students will investigate equilibrium and Le Chatelier's principle through four different experiments. They will apply various stresses to different chemical systems at equilibrium, then use their understanding of Le Chatelier's principle to explain their results. Kit contains enough material for 15 groups. Teacher's Guide and Student Study Guide Copymasters are included.

Kit Includes:

DOT Info

Non-Regulated

25mL Bromothymol blue 50mL Sodium hydroxide, 0.1M 150mL Potassium thiocyanate solution 25mL Hydrochloric acid, 0.1M 25mL Ferric nitrate solution 15g Sodium phosphate crystals 100mL Ammonium hydroxide, 1M 50mL Methyl red

15g Potassium thiocyanate crystals 150mL Copper (II) sulfate solution 25mL Hydrochloric acid, 1M



IS2526 \$29.31



Introduce students to stoichiometry and mole ratios with the reaction of copper and silver nitrate! Students will watch as beautiful silver crystals are formed on the copper wire, then determine the moles of silver formed and moles of copper reacted to calculate the mole ratio. Using this data, they will write the balanced chemical equation for the reaction, and determine the limiting reactant and their percent yield. Kit contains enough materials for 15 groups of students. Teacher's Manual and Student Study Guide Copymasters are included.





The Ideal Gas Law

Explore the gas laws through the reaction of sodium bicarbonate and hydrochloric acid. Students will perform the reaction and use the ideal gas law to calculate the amount of carbon dioxide produced. They will then use stoichiometry to determine their percent yield. Kit contains enough materials for 15 groups of students. Teacher's Manual and Student Study Guide Copymasters are included.

Kit Includes: 1 x 40g Sodium bicarbonate 15 x 25mL Hydrochloric acid, 1M 1 pkg String 15 Balloons

DOT Info: Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173.4 for domestic highway or rail transport only

IS2528

\$49.61

CHOKING HAZARD Children under 8 years can choke or suffocate on uninflated or broken balloons. Adult supervision required. Keep uninflated balloons from children. Discard broken balloons at once.

Energy - Cool It! A Guided Inquiry

The first law of thermodynamics is a physical law that governs the conservation of energy and heat transfer within a system. Energy - Cool It! is a kit that allows the demonstration of this law on an intermediate level. The included experimental protocol enables students to plan and conduct an experiment on their own with basic guidance, and allows the assessment of student learning capabilities. This kit contains enough materials for 15 groups of students. Teacher's manual and Student Study Guide copymasters are included.

Kit Includes:

- 30 Styrofoam Cups
- 18 Thermometers
- 15 50 mL Plastic Beakers
- 300 Zinc Washers: 15 Sets of 2,6, and 12 washers
- 15 Plastic Forceps

DOT Info: Non regulated **IS2529**

\$245.00

16 48

4

Matter - All Mixed Up! A Guided Inquiry

Everything from elements to molecules, objects, plants, and animals, even you and I, are all made up of matter. Matter - All Mixed Up! allows students to explore the physical properties of matter through the separation of various mixtures. The included experimental protocol enables students to plan and conduct an experiment on their own with basic guidance, and allows the assessment of student learning capabilities. This kit contains all materials needed for 15 groups to perform the experiment on one of the four mixtures. Teacher's manual and Student Study Guide copymasters are included.

Kit Contains:

Mixture #1 - Ei	nough for 5 Samples
Mixture #2 - E	nough for 5 Samples
Mixture #3 - E	nough for 5 Samples
Mixture #4 - E	nough for 5 Samples
200mL	Mixture #2 Liquid
125mL	Mixture #4 Liquid A
125mL	Mixture #4 Liquid B
8	Metal Tweezers
4	Sieves

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DOT Info: Non regulated

IS2530	\$178.31
IS2530-REF	\$61.68

4 oz Jars with Lids
Transfer Pipettes
Magnets
Plastic Funnels
Pieces of Filter Paper
7 oz Plastic Cups
Disposable Petri Dishes
Magnifying Glasses







Unknown Compositions - Potassium Hydrogen Phthalate

Learn the importance of substance purity by determining the composition of a set of impure potassium hydrogen phthalate samples via one of the most common analytical methods: Acid-Base Titration. This kit contains materials for 25 groups. Teacher's Manual and Student Study Guide copymasters are included.

Kit Includes:	
10 x 25g	Numbered Impure Potassium Hydrogen Phthalate Samples
2 x 500mL	1 N Hydrochloric Acid Solution
2 x 30mL	Bromothymol Blue Indicator

DOT Info:

UN1823, Sodium Hydroxide, 8, II, Ltd Qty. UN1789, Hydrochloric Acid, 8, III, Ltd Qty.



IS2531 \$42.50

Unknown Compositions - Ferrous Ammonium Sulfate

In this experiment, students will perform Gravimetric analysis and Spectrophotometric analysis to determine the purity of ferrous ammonium sulfate samples of unknown composition. This kit contains materials needed for 25 groups. Teacher's Manual and Student Study Guide copymasters are included.

Kit Includes:

5 x 25g	Numbered Impure Ferrous Ammonium Sulfate Samples
5 x 25g	Lettered Impure Ferrous Ammonium Sulfate Samples
12.5g	Phenanthroline
3 x 50mL	Ammonia Solution

DOT Info: UN2811, Phenanthroline, 6.1, III, Ltd. Qty.



IS2532

\$164.00

Unknown Compositions - Soda Ash

In this experiment, students will compare two analytical methods, Gravimetry and Double Acid-Base Titration, by using these methods to determine the purity of sodium carbonate. This kit contains materials needed for 25 groups. Teacher's Manual and Student Study Guide copymasters are included.

Kit Includes:	
10 x 25g	Numbered Impure Sodium Carbonate Samples
75g	Calcium Chloride
2 x 500mL	3 M Hydrochloric Acid Solution
2 x 250mL	2 N Sodium Hydroxide Standard Solution
2 x 30mL	Bromothymol Blue Indicator
2 x 30mL	Methyl Orange Indicator
	, ,

DOT Info: UN1789, Hydrochloric Acid, 8, III, Ltd. Qty. UN1824, Sodium Hydroxide, 8, II, Ltd. Qty.



IS2533



ACS - American Chemical Society Lab Activities These activities are developed by the American Chemical Society through grants from the National Science Foundation and National Institutes of Health.

American Chemical Society: Chemistry – Investigating Your World Lab Activity

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This series of activities, developed by the American Chemical Society through grants from the National Science Foundation and National Institutes of Health, is comprised of four lessons that investigate chemical changes. Using the instruction manual obtained from the American Chemical Society's website, and the materials included in the kit, students will take an imaginary trip around the globe to meet scientists and get a sense of the wide variety of ways that chemistry is used to benefit mankind. Activities include the production of a gas, chemistry and color change, formation of a precipitate, and temperature change of a chemical reactions. Contains enough materials for eight groups.

100g	Citric Acid
100g	Sodium Carbonate
250g	Sodium Bicarbonate
250g	Calcium Chloride
8x25mL	Universal Indicator
30mL	Dish Detergent
30mL	Bromothymol Blue
16	3oz cups
32	2oz cups
9	Medicine Cups
18	Scoops
	Toothpicks

Student Thermometer Vials Spot Plates Pipettes Test Tube w/Rubber Stopper Balloon Drinking Straw Universal Indicator Chart Self Inflating Reaction Bag Foot Warmer, 2 pack Spoons



DOT Info: Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173.4 for domestic highway or rail transport only

IS2550

\$152.00

CHOKING HAZARD Children under 8 years can choke or sufficate on uninflated or broken balloons. Adult supervision required. Keep uninflated balloons from children. Discard broken balloons at once.

WARNING: This product can expose you to chemicals including Phenolphthalein/Methanol/ Methyl isobutyl ketone and Phthalates, which are known to the State of California to cause cancer, reproductive harm. For more information go to www.P65Warnings.ca.gov.

American Chemical Society: Chemistry's Colorful Clue Lab Activity

Through the lessons in this lab students will learn how human activity may affect natural water sources. Using an indicator, students will determine how acids and bases impact the pH of water. Upon completion of this activity, students will then create carbon dioxide gas through a chemical reaction and determine if the carbon dioxide could have affected the pH of a local stream.

Kit Includes: 30g Citric Acid 9 x 25mL Bromothymol Blue Indicator 16 Portion Cups, 2oz 18 Clear 3oz Plastic Cups 9 Small Plastic Spoons 1 pkg Toothpicks 1 Plastic Spoon 5 pieces Paper

DOT Info: Non-regulated

30g Baking Soda 9 Small Metric Measuring Cups 25 Clear 10oz Plastic Cups 16 Clear 9oz Plastic Wide Cups 8 Pipettes 32 Cotton Swabs 1 Straw

\$58.58



www.aldon-chem.com

IS2551



General Chemistry

Innovating Science®



ACS - American Chemical Society Lab Activities

These activities are developed by the American Chemical Society through grants from the National Science Foundation and National Institutes of Health.

American Chemical Society: Glow It Up Lab Activity

This lesson explores chemical reactions that release energy in the form of light, known as chemiluminescence. Teachers begin with a demonstration that uses a luminol mixture and oxidizer to create a chemical reaction that glows. Students will then conduct an experiment to see how the amount of oxidizer affects the brightness and duration of the light. Students will also add a fluorescent highlighter dye to the reaction to see how this affects the brightness and duration of the glow. Additionally, they will learn about the invention of the glow stick and relate their experience to how a glow stick works.

Kit Includes:

9 x 5g Luminol Copper Blend 8 Polypropylene Beaker 1 Clear 10oz Plastic Cup 17 Pipettes 8 pieces Construction Paper 40 pieces White Paper Squares, 1" x 1

9 x 5g Sodium Percarbonate 18 Small Plastic Scoops 25 Clear 3oz Plastic Cups 2 Yellow Highlighters 5 Small Glow Sticks

DOT Info: Small quantity exemption 173.4 THIS PACKAGE CONFORMS TO 49 CFR 173.4 for domestic highway or rail transport only

IS2552



American Chemical Society: CO₂ To the Rescue Lab Activity

This lesson begins with a design challenge: to invent a small device that could rescue a cell phone that accidentally falls into water. The teacher starts off by showing students a balloon that inflates with carbon dioxide gas as chemicals inside the balloon react with one another. Students conduct a pair of chemical reactions to determine which of two acids react with baking soda to produce the most carbon dioxide gas. Once students determine the best acid to use, they compare the amount of gas produced with different amounts of baking soda. Finally, trying to use the smallest volumes possible, students discover how much of each reactant is needed to fully inflate a small zip-closing plastic bag to see if it can get a model clay cell phone to float.

Kit Includes:

175g Baking Soda
8 x 5g Calcium Phosphate
8 Small Metric Measuring Cups
8 Clear 10oz Plastic Cup
16 Clear 9oz Plastic Wide Cups
8 Short Pipettes
16 Tubes
1 Plastic Spoon
1 Clear Self-inflating Balloon

DOT Info: Non-regulated



CHOKING HAZARD Children under 8 years can choke or suffocate on uninflated or broken balloons. Adult supervision required. Keep uninflated balloons from children. Discard broken balloons at once.

IS2553

8 x 25g Citric Acid 25mL Liquid Dish Detergent 16 Portion Cups, 2oz 16 Clear 3oz Plastic Cups 9 Long Pipettes 24 Small Scoops 40 Reclosable Bags 5 Self-inflating Balloons 160g Clay



\$111.21

Aldon

Nanotechnology: Ferrofluids

Understand the definition of nanotechnology as it applies in current usage. Learn of one nanomaterial, ferrofluid, and some of its applications in modern technology. Create magnetite nanoparticles through a precipitation reaction, while employing a surfactant to create a colloidal suspension of magnetite nanoparticles (ferrofluid). In the end you will examine the response of ferrofluid upon exposure to a magnetic field. Kit contains enough materials for 15 groups. Teacher's Manual and Student Study Guide copymasters are included.

Kit Includes:

THE INCIUM	53.
8g	Ferrous Chloride NA 1759
22g	Ferric Chloride UN1759
20mL	10N Hydrochloric Acid UN1789
2x25mL	Ammonium Hydroxide UN2672
20mL	Tetramethylammonium Hydroxide UN1835
15	Pipettes
15	Petri Dishes

DOT Info:

Small quantity exemption 173.4 This package conforms to 49 CFR173.4 for domestic highway or rail transport only



IS2800

\$63.74

Ferrofluid

Ferrofluid is a colloidal liquid made of particles less than 10 nanometers in diameter. When exposed to a magnetic field, the nanoparticles form regular patterns of peaks and valleys. You can demonstrate magnetic fields for your students in a very visual way.

Kit Includes: 50mL Ferrofluid

DOT Info: Non-regulated

IS2801

\$37.01

WARNING: This is not a toy, this product is intended for scientific use by ages 13 and up.

from synthetic elements. Name, symbol, atomic number, weight, mass, electron configuration are included.

Periodic Table

IS2900 Periodic Table, Laminated	\$11.29
IS2910 Periodic Table, Paper, Set/25	\$74.54
Measures 11" by 17"	

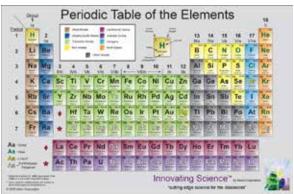
Colored printing helps your students distinguish between

metals, non-metals, and noble gas families as well as allowing

them to differentiate types of metals, solids, liquids and gases

IS2940 Periodic Table Poster 21" x 34"	\$13.00
IS2941 Periodic Table Poster 45" x 35"	\$84.00
IS2942 Periodic Table Poster, vinyl 4' x 8'	\$204.57

DOT - Non-regulated



Periodic Table with Reference Charts

Our great updated periodic table in full color as pictured above, with the addition of handy reference charts on the back including periodic trends, physical constants, conversion factors, and more! Measures 8.5" x 11", printed on 100 lb glossy paper, and is 3-hole punched for easy use in notebooks and binders.

IS2943 Periodic Table with Reference Charts	\$3.03
IS2944 Periodic Table with Reference Charts pk/1	0 \$25.64



800-724-9877



General Chemistry

\$11.05

Innovating Science®

Periodic Table Pen

The most convenient reference guide available - the Periodic Table Pen. This pen holds a double sided Periodic Table. The back side is a handy miniature reference chart. A great reference right at your fingertips!



IS2920 - Periodic Table Pen\$4.64IS2920-CS/25 - Periodic Table Pen\$130.97DOT - Non-regulated\$130.97

Acid Base Strength Chart

Easily determine the relative strength of different acids and bases on this laminated 11x17 inch chart.

	Name of Acid	Acid		Base	Кa	
	Hydriodic	HI	\longrightarrow	H+ + I	>> 1	
Г	Hydrobromic	HBr	\longrightarrow	H+ + Br ⁻	>>1	
GB	Hydrochloric	HCI	\longrightarrow	H+ + CI=	>>1	
orrong Acids	Sulfuric	H ₂ SO ₄	\longrightarrow	H+ + HSO4	> 1	
DI C	Nitric	HNO ₃	\longrightarrow	H+ + NO3	> 1	
	Hydronium Ion	H ₃ O ⁺	⇒	$H^+ + H_2O$	5.50 X 10 ¹	
	Oxalic	H ₂ C ₂ O ₄		H+ + HC ₂ O ₄ -	5.90 X 10 ⁻²	
	Sulfurous	H ₂ SO ₃	\rightleftharpoons	H+ + HSO3	1.41 X 10 ⁻²	
	Hydrogen Sulfate Ion	HSO₄ [−]	<u> </u>	H+ + SO.2-	1.20 X 10 ⁻²	
	Phosphoric	H ₃ PO ₄		H ⁺ + H ₂ PO ₄ ⁻	7.52 X 10 ⁻³	
	Nitrous	HNO ₂		H ⁺ + NO ₂ ⁻	5.62 X 10 ⁻⁴	
	Hydrofluoric	HF		H ⁺ + F ⁻	3.53 X 10 ⁻⁴	
weak Acids	Formic	HCO ₂ H	<u> </u>	H ⁺ + HCO ₂ ⁻	1.78 X 10 ⁻⁴	
eak #	Benzoic	C ₆ H ₅ CO ₂ H	—	H ⁺ + C ₆ H ₅ CO ₂ ⁻	6.46 X 10 ⁻⁵	
Ň	Acetic	CH ₃ CO ₂ H	⇒	H ⁺ + CH ₃ CO ₂ ⁻	1.76 X 10 ⁻⁵	
	Carbonic	H ₂ CO ₃	⇒	H+ + HCO3	4.30 X 10 ⁻⁷	
	Hydrogen Sulfite Ion	HSO3	<u> </u>	H+ + SO32-	1.02 X 10 ⁼⁷	
Weakest Acids	Dihydrogen Phosphate Ion	H ₂ PO ₄	<u> </u>	H+ + HPO42-	6.23 X 10 ⁻⁸	
	Boric	H ₃ BO ₃	\rightleftharpoons	$H^{+} + H_{2}BO_{3}^{-}$	5.79 X 10 ⁻¹⁰	
	Ammonium Ion	NH_4^+	\Longrightarrow	H ⁺ + NH ₃	5.64 X 10 ⁻¹⁰	
	Hydrogen Carbonate Ion	HCO3	<u> </u>	H+ + CO32-	5.61 X 10 ⁻¹¹	
veake	Monohydrogen Phosphate Ion	HPO42-	\Longrightarrow	H+ + PO ₄ ³⁻	2.20 X 10 ⁻¹³	
	Water	H ₂ O		H+ + OH-	1.00 X 10 ⁻¹⁴	
A	Phosphate Ion	H ₂ O	=	н⁺ + он ⁻ Science [®]		

DOT - Non-regulated

IS2985

Aldon

\$11.05

Principles of Stoichiometry

Ever wonder what would happen if you mixed this chemical with that one? This kit has been designed to allow students to perform experiments to see what types of reactions occur when various solutions are mixed together. Four different experiments focus on the concepts of decomposition reactions, single replacement reactions, double replacement reactions and balanced reactions. The materials in this kit are packaged to minimize the need for teacher preparation providing 5 unique sets of materials for students working in groups of 6. The dropper topped bottles allow students to perform the array experiments without the risk for cross contamination of chemicals. Teacher's Manual and Student Study Guide copymasters are included.

GHS Poster

quickly and easily.

DOT - Non-regulated

Identify new Global Harmonization Symbols

GHS PICTOGRAMS

IS2980 Laminated 11" x 17"

Kit Includes:

50mL	Cupric Sulfate 1.0 M
15mL	Nickel Nitrate 0.1 M
15mL	Cobalt Nitrate 0.1 M
15mL	Silver Nitrate 0.1 M
15mL	Iron (III) Nitrate
15mL	Sodium Sulfate 0.1 M
15mL	Sodium Chloride 0.1 M
75mL	Sodium Iodide 0.1 M
15mL	Sodium Chromate 0.1 M
15mL	Sodium Hydroxide 0.1 M
5x12mL	Hydrochloric Acid 1.0 M
5x12mL	Sulfuric Acid 1.0 M
5x12mL	Sodium Oxalate 0.1 M
5pkg	Zinc Metal Strips
5pkg	Magnesium Metal Strips
15	Dual Well Combo Plates
5	Forceps
5 sets	Cotton String

15mL	Cupric Nitrate 0.1 M
75mL	Lead Nitrate 0.1 M
15mL	Nitric Acid 0.1 M
15mL	Aluminum Nitrate 0.1 M
15mL	Sodium Carbonate 0.1 M
15mL	Ammonium Hydroxide 0.1 M
15mL	Ethylenediamine TetraAcetic Acid 0.1 M
15mL	Sodium Thiocyanate 0.1 M
15mL	Sodium Dichromate 0.1 M
150mL	Deionized Water
65mL	Nitric Acid 1.0 M
5x12mL	Calcium Nitrate 0.1 M
5pkg	Aluminum Metal Strips
5pkg	Copper Metal Strips
15	Preprinted Acetate Grid Sheet
1box	Wooden Dowels
24	Carbon Electrodes

DOT Info: UN2031, Nitric acid, 8, II UPS Hazard charge applies

IS8060

WARNING: This product can expose you to chemicals including Lead and lead compounds, Chromium' hexavalent compounds, Nickel (soluble compounds) and Strong inorganic acid mists containing sulfuric acid which are known to the State of California to cause cancer, reproductive harm. For more information go to www.P65Warnings.ca.gov.

\$375.00

800-724-9877

Engineer and Explore Your Own Enteric Coated Drugs

In this lab, students will learn the basic structures and pathway of the digestive system and understand the different functions of the stomach and the small intestine in regards to digestion. They will investigate how the properties of different enteric coatings react in different sections of the digestive system. Next they will explore the purpose of an enteric coating and make a simulated "coating" and engineer a coating most suitable for certain pharmaceutical needs. This lab was written by a Biomedical Engineer and has enough materials for 15 groups. Includes Teacher's manual and student guide.



Biomaterials to Make Your Own Contact Lenses

Engineers play an integral role in the process of finding a "perfect" material or ratio of materials to maximize the desired properties of an invention and decrease the amount of negative effects of other properties. For example, when developing the perfect material for use in contact lenses, an engineer will have to test many different materials before finding the right one. A hydrogel is a polymer similar to plastic that has favorable optical properties and favorable flexibility/strength but like paper and glasses/ceramics is hydrophilic, which is necessary for a contact lens to function properly. Testing a hydrogel, which is favorable for all three of the main properties needed for a contact lens to function properly, is the main job of a biomedical engineer developing contact lenses today. They make a hydrogel and find the perfect "ratio" of polymer to water. The composition of the hydrogel needs to be hydrophilic enough to maintain a wet environment in the eye but not too hydrophilic so that the contact begins to swell and change size and shape. Students will engineer a lens using 2 different materials. They will determine the correct material and the concentration that has similar properties as a contact lens, while discovering the correct optical properties, tensile strength and hydrophobicity. Kit contains enough materials for 15 groups. Teacher's Manual and Student Study Guide copymasters are included.



Innovating Science[®]

Engineering a Drug Delivery System

Biomedical engineers design devices, equipment, and processes to improve medical treatment methods and the quality of human health and life. They apply engineering principles to biology and medicine to create products such as prosthetic limbs and artificial organs, among other things. One major focus of biomedical engineers is the design of drug delivery systems, which can control the transport and release of medicine in the body to more effectively treat diseases and illnesses. In this experiment, students will gain an understanding of drug delivery and its importance by testing simulated delivery systems in multiple environments that mimic different parts of the body. The delivery systems will be analyzed and compared, allowing students to engineer an optimal solution and learn what properties must be considered when designing new drug delivery systems. This lab has enough materials for 15 groups. Teacher's manual and Student Study Guide copy masters included.

Kit Includes:

- Simulated Drug 5g
- Hydrochloric Acid 1.0M 1L
- 1L Sodium Hydroxide 1.0M
- 45 Hydrogel Delivery Systems
- 45 **Plastic Delivery Systems**
- 45 Cups, 5oz

DOT Info: UN1789, Hydrochloric acid, 8, III, Ltd Qty UN1824, Sodium hydroxide solution, 8, II, Ltd Qty

IS3403

\$67.85



Engineer and Explore Materials for Prosthetics

In this lab, students will explore the materials that biomedical engineers use when creating prosthetic devices. They will first investigate the physical properties of various metals, ceramics, and polymers to understand how they differ, and determine which materials would be most suitable for use in a hip replacement prosthetic. Students will then engineer their own prosthetic arm that meets certain design requirements and mimics the functionality of a real hand. Kit contains enough materials for 15 groups. Teacher's Manual and Student Study Guide copy masters are included.





STEM - Science, Technology, Engineering, and Mathematics For use with Neulog[®] sensors

Green Fuel Cell Kit - IS 6500

Monitor the voltage generated by a biological fuel cell, using the yeast Saccharomyces cerevisiae. A simple sugar will be used as a food source for the yeast and methylene blue will be employed as an electron mediator. The cell will be constructed without the mediator and a baseline voltage will be established as some electrochemical activity will be occurring. Once the cell has stabilized, the mediator will be added and the effect on voltage will be monitored and recorded using a Neulog[®] Voltage sensor. There are enough materials for 15 set ups.

WARNING: This product can expose you to chemicals including Phthalates, which is known to the State of California to cause reproductive harm. For more information go to www.P65Warnings.ca.gov.



Photosynthesis and Respiration Chamber - IS6501 \$39.00



The photosynthesis/respiration demonstration chamber may be used to monitor oxygen levels, carbon dioxide levels, or both in an enclosed environment. The chamber has been designed to be used with the Neulog[™] Oxygen logger sensor and/or the Neulog[®] Carbon dioxide logger sensor as well as Neulog[™] software for data acquisition. As an option, a light sensor may also be used when running photosynthesis experiments to correlate light intensity to changes in carbon dioxide and oxygen levels in the chamber. The lid of the chamber provides openings for both the oxygen sensor and the carbon dioxide sensor. If both sensors will not be used for a particular experiment, place the proper rubber stopper in the opening that will not contain a sensor during the experimental run. You can set up an experiment and record light and dark cycles over a weekend or period of time without even being in the classroom as the sensors will capture all of the information for your student to study.

\$80.25

WARNING: This product can expose you to chemicals including Phthalates, which is known to the State of California to cause reproductive harm. For more information go to www.P65Warnings.ca.gov.

Effectiveness of Sunscreen - IS6502 \$39.00

Electromagnetic radiation in the UV range of the spectrum cannot be perceived by the human eye. Using the Neulog[®] UVB sensor specifically designed to measure UVB radiation, you will compare the amount of UVB radiation reaching the sensor. The sensor will be protected by both SPF15 and SPF30 sunscreens to determine if the sunscreens actually provide UVB protection and if so, how much protection. There are enough materials for 15 set ups.

Optimized Fermentation and Yeast - IS6503 \$43.50

Determine the optimum food source for yeast by monitoring growth rate and respiration. Requires pressure sensor.

Physical Characteristics of Gases - IS6509 \$131.00

The gas laws (Boyle's, Charles's and Gay-Lussac's) are simple mathematical relationships between volume, temperature and pressure. By varying one factor you can measure the effect on the others. Requires temperature and pressure sensors.

MARNING: This product can expose you to chemicals including Phthalates, which is known to the State of California to cause reproductive harm. For more information go to www.P65Warnings.ca.gov.





DOT Info: Non-regulated

800-724-9877



Biotechnology

Electrophoresis and DNA Fragment Length Determination

In this activity students will perform agarose electrophoresis on three DNA samples that have already been treated with restriction enzymes. Students will not only learn about the process of DNA electrophoresis but learn the techniques associated with the process, such as setting up an agarose gel, loading DNA samples in the agarose gel, and staining the gel in order to visualize the DNA bands. They will also learn how to determine the sizes of unknown DNA fragments after examining their results. The kit includes specially-treated DNA samples that do not require refrigeration or freezing, prepared agarose that may simply be melted in a hot water bath or microwave, TBE electrophoresis buffer, and DNA stain. There is enough DNA to run 10 gels.

Kit Contains: 200mL 500mL 60mL

0.8% Agarose 5X Tris Borate ETA Buffer, 20X DNA Stain LAMBDA DNA ECORI DIGEST With Loading Dye LAMBDA DNA HINDIII DIGEST With Loading Dye LAMBDA DNA ECORI/HIND III DIGEST With Loading Dye

DOT Info: Non-regulated

IS3300

\$118.00

DNA Fingerprinting Electrophoresis Lab Activity

DNA fingerprinting is one of the strongest forms of evidence in forensic investigations. In this activity students will perform agarose gel electrophoresis on four DNA samples to determine if DNA from suspects matches that of DNA found at a crime scene. Students will also learn about restriction enzymes and the importance of these enzymes in the DNA fingerprinting process. The kit contains enough materials for 8 groups and includes a Teacher's Guide and Student Study copymasters.

Kit Includes: 200mL Prepared Agarose 0.8% Solution 60mL DNA Stain 20X 10µg Victim DNA 10µg Suspect #2 DNA

DOT Info: Non-regulated

IS3301

\$120.00

500mL Tris-Borate-EDTA Buffer 5X

Crime Scene DNA Suspect #1 DNA

10µg

10µg



DNA Paternity Testing Electrophoresis Lab Activity

While over 99% of DNA is the same among all humans, the remaining part is unique to every individual (with the exception of identical twins). These differences are hereditary, with parts of the unique sequence coming from each parent. By using DNA fingerprinting, or DNA profiling, a child's paternal relationship can be tested by comparing the child's DNA profile to the profile of the mother and possible father to determine if an individual is the father or not. In this activity students will use electrophoresis to separate DNA samples of a child, mother, and two potential fathers to determine if either father is a possible paternal match. The kit contains enough materials for 8 groups of students and includes a Teacher's Guide and Student Study Guide copymasters.

Kit Includes: 200mL Prepared Agarose 0.8% Solution 60mL DNA Stain 20X 10µg Child's DNA Sample 10µg Possible Father #2 DNA Sample

DOT Info: Non-regulated

IS3303

500mL Tris Borate EDTA Buffer 5X 10µg Mother's DNA Sample 10µg Possible Father #1 DNA Sample





Aldon

	Prepared Agarose	
Agarose dye marker set can be used with a 2% agarose gel to get students accustomed to the electrophoresis process without using expensive DNA.	Simply melt the prepared agarose in a water bath or Pour into a gel casting tray, allow the molten agarose run your gel.	
fou can run up to 8 gels.	IS5201 \$14.99 0.8% Agarose 200mL - Resolves DNA fragments 0	.6-35kb long
mL Bromophenol Blue mL Xylene Cyanol mL Orange G	IS5202 \$14.99 1.0% Agarose 200mL - Resolves DNA fragments 0	.5-20kb long
mL Crystal Violet mL Malachite Green mL Dye Mixture	IS5203 \$17.99 2.0% Agarose 200mL - Resolves DNA fragments 0	, i i i i i i i i i i i i i i i i i i i
OT Info: Non-Regulated		. Fold long
WARNING: This product can expose you to chemicals including Crystal Violet, which is known to the State of California to cause cancer. For more information go to www.P65Warnings.ca.gov.		
S5007 \$12.00	DOT Info: Non-Regulated	
	Powdered Agarose	
For Biotechnology equipment and Histology Reagents	Prepare your own agarose gels at any concentration addition of TBE running buffer. Can be stored at room for years.	
please visit our website!	IS5204 - Agarose 5 gram bottle (Low EEO)	\$20.06
	IS5205 - Agarose 25 gram bottle (Low EEO)	\$67.85

Agarose Gel Reagents

 IS5206 - TBE Buffer 5XConcentrate 500mL. 5X running buffer concentrate makes 2.5L of 1X working concentration \$14.50

 IS5207 - Tris-EDTA (TE) Solution 10X concentrate 25mL. Use for diluting DNA samples
 \$5.65

 IS5208 - Loading Dye 10x 5mL. Dye that is used for tracking DNA during agarose electrophoresis.
 \$7.18

 IS5209 - DNA20x Stain 60mL bottle. Stain and Destain in 40 Minutes. This highly sensitive stain will stain a gel in about 20 minutes and destain in 20minutes. Once the gels are stained they can be stored for months without fading. Stain up to 20 gels with 60mL.

 IS5250 - Agarose Gel Electrophoresis Reagent Pack. All the necessary reagents to prepare and run agarose gels in one convenient package. Contains: TBE buffer, 5x concentrate 500mL; Agarose, powdered, 5g; DNA Stain 20x concentrate, 60mL; Loading Dye, 10x concentrate, 10mL

DOT Info: Non-Regulated



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Formation Native Copper Mineral Deposit

Buffering Activity/Common Products Preparation Of Effective Buffers Complete Set/16 New AP Chem

Presumptive Gun Shot Residue Presumptive Blood Test Kit

Crystal Growing Kit

Document Analysis

IS9011 Crime Scene Investigation Lab

Hair Analysis

Green Chemistry & Purification

Thermochromism Determination Of Water Hardness

Absorption Spectroscopy Demo

Exothermic Reactions

Synthetic Rubber

Formation Of Silicate

Polyurethane Foam

The Silicate Garden

Water To Wine

War Of The Acids

Electrolysis of Tin

Electric Art

Activity Series

Acids & Bases

Reaction Kinetics

Qualitative Analysis

Freezing Point Depression

Vapor Pressure Electrochemical Cells Thin Layer Chromatography

Synthesis Of Aspirin

Acid Base Indicators

Electrochemical Series Liquid Chromatography

Principles Of Stoichiometry

Separation Of Molecules What's In That Bottle?

Beer Lambert Law

Mole Ratio of Reactants

Lechatelier's Principle Formation Of Esters

Formation Of Tin Wool

Solid Fuel

Dehydration Of Sucrose

Ampholytes

IS7006

IS7018

IS7023

IS7030

IS7037

IS7039

IS7042

IS7045

IS7049

IS8001

IS8005

IS8008

IS8012

IS8016

IS8019

IS8023

IS8027

IS8030

IS8102

IS8106

IS8109

IS9002

IS9012 Carbon Fingerprinting IS9012-REF Fingerprint Powder Refill IS9013 Chemical Detection Of Fingerprints 96 37 96 97 37 IS9014 Blood Spatter: Bloodstain Analysis IS9015 Soil & Mineral Analysis IS9015-refill Soil & Mineral Analysis IS9016 Drug And Poison Analysis 37 97 37 38 97 98 IS9016-REF Drug And Poison Analysis Refill IS9017 Document Analysis: Comprehensive IS9017-REF Document Analysis: Refill IS9018 Forensic Analysis Of Fabric And Fibers 38 98 38 98 98 38 38 99 39 IS9019 Who Stole The Gold? Forensic Analysis 99 IS9020 Forensics Mastery IS9020-REF Forensic Mastery Refill 99 98 39 39 39 IS9021 Forensic Dental Analysis 100 IS9021-refill Forensic Dental Analysis Refill IS9022 Forensic Toxicology 100 40 40 100 40 IS9023 Forensic Drug Testing:Sim Immunoassay IS9024 Forensic Case Study: Small Town Big Problem 101 40 101 41 IS9024-REF Small Town Big Problem refill 101 41 IS9025 Forensic Case Study: Murder at Eagle Nest Harbo IS9025-REF Forensic Case Study: Murder at Eagle Nest 102 102 41 IS9026 Forensic Analysis of Narcotics 102 Case of the Contaminated Lake Forensic Urine Drug Test Kit 42 IS9027 6 The Oxidation-Reduction Flag Oscillating Reaction- The Redox Rainbow Oxidation Of Glycerin 42 42 IS9050 IS9075, IS9077 Simulated Blood Spatter 97 IS9350 Properties Of Toothpaste IS9351 Properties Of Soaps And Detergents IS9352 Properties Of Antacids 43 103 43 43 104 189353 Properties Of Shampoo 189353-REF Properties Of Shampoo Refill 189354 Properties Of Aspirin 44 104 44 104 105 44 45 IS9355 Science In The Kitchen 105 Production Of Biodiesel Electrochemical Remediation Wastewater Detergents And The Environment 45 189500 51 51 45 IS9501 46 IS9502 51 46 IS9503 Synthesis Of Acetyl- Salicylic Acid 52 Det. Composition Of Unknown Mix. Alternative Iodine Clock Reaction Synthesis Of Silver Nanoparticles 47 189504 52 Fluorescent Slime Using Polyvinyl Alcohol Instant Light Powder Kit Snow Polymer Demonstration 47 47 IS9505 52 IS9506 53 48 IS9507 Hydrogen Fuel Cell Demonstration 53 48 159508 Green Fuel Cell Water Treatment And Filtration 53 54 48 IS9700 49 IS9701 Nitrates, Phosphates, Eutrophication 54 49 159703 Acid Rain, Weathering, And Erosion Acidity And Plant Growth 54 55 55 55 IS9704 50 50 Oil Spill Clean Up IS9705 5 IS9707 Effect Of Salinity On Seawater Ocean Pollution Dissolved Oxygen Concentraton Water Hardness 159708 7 6 Firefly Flask: Oscillating Chemiluminescence Thermochemistry & Hess Law . 56 29 IS9751 56 29 IS9752 Water's Alkalinity 56 29 159753 Nitrite And Nitrate Test Ammonia Concentration In Water 56 57 30 30 IS9754 IS9755 Determiniation Of Salinity 57 30 189756 Ocean Acidification 8 57 30 IS9757 Determination of Dissolved CO₂ Co Plant Tissue Macronutrients ed CO, Conce 8,58 31 31 IS9758 8,58 IS9759 Carbon Cycle in Soil 9.58 Estuary Monitoring Water Test Kit Soil Macronutrients Urban Water Testing Kit 31 159760 9 59 31 32 IS9761 IS9762 10.60 32 IS9763 Total Water Investigation Kit 10.60 32 IS9799 Environmental Testing Kit IS2514-SGL Introduction To Chemical Equations 59 82 32 33 IS2518-SGL Introduction To Ionic Reactions IS2520-SGL A Safer Flame Test: Identification Of Metal IS2521-SGL Fractional Distillation IS2522-SGL Chromatography Of Amino Acids 82 33 82 33 83 33 34 83 IS2523-SGL Separation Of A Mixture Of Solids 83 IS3002-SGL DNA/Chromosome Staining IS3002-SGL DNA/Chromosome Staining IS3002-SGL DNA Extraction IS3003-SGL Diffusion and Cell Size 34 84 84 Grignard Synthesis Of Benzoic Acid Complete Set Of All 22 AP Sets Determination Of Properties Of Buffer Sol Empirical Formula Of MgO 34 84 IS3009-SGL Enzymes And The Process Of Digestion IS3010-SGL Mendelian Genetics: Genes And Probability IS30114-SGL Microbiology - Bacterial Growth & Staining 35 85 35 85 Determination Of The Molar Vol. Of A Gas 35 35 IS3016-SGL Deluxe Owl Pellet Dissection 85 IS3102-SGL Deluxe Ow Pellet Dissection IS3102-SGL Genetics Of Blood Types IS3704-SGL Diffusion And Osmosis IS8013-SGL Oxidation-Reduction Reactions 36 86 36 86 36 86 36 IS8030-SGL Electrochemical Series 87 IS8701-SGL Formation Native Copper Mineral Deposit IS8702-SGL Soil Analysis Earth Science 129 87 23 23 87 IS9006-SGL Unknown Substances 88 23 IS9007-SGL Hair Analysis 88 24 IS9350-SGL Properties Of Toothpaste IS9351-SGL Properties Of Soaps And Detergents 88 89 24 24 IS9354-SGL Properties Of Aspirin 89 IS9355-SGL Science In The Kitchen IS9500-SGL Production Of Biodiese 25 89 25 90 25 IS9700-SGL Water Treatment And Filtration 90 26 IS9701-SGL Nitrates, Phosphates, Eutrophication 90 IS10023-IS35002 Popular Reagents 116 IS30000-IS30002 Universal Indicator Solution 114 MD0100-MD0309 Prepared Microbiology Media 75 26 26 27 XD1360-XD1398 Antibiotic Disks 28 28 28 91

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